

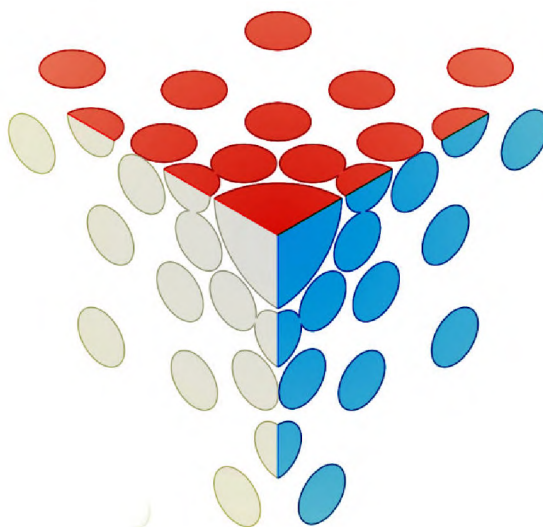
The joint event of

The Eleventh Young Researchers' Conference
Materials Science and Engineering

and

The First European Early Stage Researchers' Conference on
Hydrogen Storage

Belgrade, December 3rd - 5th, 2012



Under the auspices of

Materials Research Society of Serbia
COST Action MP1103
University of Belgrade
Serbian Academy of Sciences and Arts
Italian Embassy

PROGRAM AND THE BOOK OF ABSTRACTS

**MATERIALS RESEARCH SOCIETY of SERBIA
INSTITUTE of TECHNICAL SCIENCES of SASA
VINČA INSTITUTE of NUCLEAR SCIENCES, UNIVERSITY of BELGRADE
HYDROGEN STORAGE INITIATIVE SERBIA**

PROGRAM AND THE BOOK OF ABSTRACTS

**JOINT EVENT OF THE 11TH YOUNG RESEARCHERS' CONFERENCE: MATERIALS
SCIENCE AND ENGINEERING**

AND

**THE 1ST EUROPEAN EARLY STAGE RESEARCHERS' CONFERENCE ON HYDROGEN
STORAGE**

**Edited by:
Jasmina Grbović Novaković
Nenad Ignjatović**

Joint event of the 11th Young Researchers' Conference: Materials Science and Engineering and the 1st European Early Stage Researchers' Conference on Hydrogen Storage

Book Title: *Joint Event of the 11th Young Researchers' Conference: Materials Science and Engineering and the 1st European Early Stage Researches' Conference on Hydrogen Storage*
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**Dr. Jasmina Grbović Novaković
Prof. Dr. Nenad Ignjatović**

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Dear Colleagues,

Welcome to the Joint meeting of **THE ELEVENTH YOUNG RESEARCHERS' CONFERENCE - MATERIALS SCIENCE AND ENGINEERING 2012 & THE FIRST EUROPEAN EARLY STAGE RESEARCHERS' CONFERENCE ON HYDROGEN STORAGE** and the city of Belgrade!

Young Researchers' meetings are held annually late in December since 2002 and they are organized by the Materials Research Society of Serbia. Originally conceived as seminars, since 2007 these meetings were transformed into conferences. The previous ten meetings featured presentations based on the research of various young scientists from Serbia, Bosnia and Herzegovina, Montenegro, Slovenia, Brazil, Germany, United States of America, China, Poland, Belgium, Spain, Romania, United Kingdom, Austria, Italy, Hungary, Russia, Canada, etc. At the Conference, young researchers, students of doctoral, master and undergraduate studies, are given the opportunity to make an overview of their research into materials science and engineering through oral and poster presentations.

As for the scientific content of the conference, we have given full priority to research topics that are currently considered as being on the frontier of the field. Nanotechnology and Advanced Materials, Synthesis and Engineering of Biomaterials, Application of Biomaterials, Theoretical Modeling of Materials and Advanced Methods for Synthesis and Processing present only some of those exciting topics that will be given the central stage and most attention during this meeting. The conference is free of charge and the participants are invited to submit their papers to the journals Tehnika – Novi Materijali, Processing and Application of Ceramics and Energija.

Early Stage Researchers' Conference of Hydrogen Storage is held in Belgrade for the first time under the auspices of the COST ACTION MP 1103. The aim of the meeting is to gather the young researchers from all over Europe and the World dealing with hydrogen energy, to discuss on the important issues regarding hydrogen storage and production. The papers from this conference will be published in International Journal of Hydrogen Energy Special Issue and Energija in order to disseminate the knowledge and to improve the visibility of our COST Action MP1103.

Prof. Dr. Nenad Ignjatović
Dr. Jasmina Grbović Novaković

MISSION AND GOAL OF COST

COST is an intergovernmental framework for European **Co**operation in Science and Technology, allowing the coordination of nationally-funded research on a European level. As a precursor of advanced multidisciplinary research, COST plays a very important role in building a European Research Area (ERA). It anticipates and complements the activities of the EU Framework Programmes, constituting a "bridge" towards the scientific communities of emerging countries increasing the mobility of researchers across Europe and fostering the establishment of scientific excellence. Through its inclusiveness COST supports integration of research communities, leverages national research investments and addresses issues of global relevance.

One of COST's main characteristics is its flexibility, allowing for an easy implementation and light management of the research initiatives. Activities are launched following a "bottom-up" approach, meaning that the initiative of launching a COST Action comes from the European researchers themselves. The member countries participate on a "à la carte" principle, in that only countries interested in the Action participate.

COST is governed by the COST Member States; 36 COST countries (35 member states and one cooperating state) govern COST. In addition, non-COST country institutions are welcome to participate in the COST programme. COST has also signed reciprocal agreements with Australia, New Zealand, South Africa and Argentina. The organization of COST reflects its intergovernmental nature.

Researchers who wish to launch a new COST Action have to submit a short proposal to the continuous COST Open Call for Proposals, which is then evaluated by the Domain Committee. If the proposal is accepted, the applicants are invited to submit a full proposal with a complete Technical Annex (the scientific and managerial description of an Action). The full proposal is then assessed through a peer review process. Recently, 287 COST Actions are running. Serbia participates in 145 COST Actions.

Prof. Biljana Stojanović, CNC COST Serbia

*Joint event of the 11th Young Researchers' Conference: Materials Science and Engineering and
the 1st European Early Stage Researchers' Conference on Hydrogen Storage*

PROGRAM

of

**The 11TH YOUNG RESEARCHERS' CONFERENCE:
MATERIALS SCIENCE AND ENGINEERING**

Conference program 11YRC

Monday, 3rd December 2012.

8:30- 9:30

Registration

9:30-10:00

Opening ceremony

MAIN HALL

Prof. Dr. Nenad Ignjatović, President of the Scientific and Organizing Committee of 11YRC

Prof. Dr. Ivanka Popović, Vice-Rector, University of Belgrade

Prof. Dr. Dragan Uskoković, President of the Materials Research Society of Serbia

Dr. Paolo Battinelli Scientific Attache', Italian Embassy in Belgrade

Prof. Dr. Biljana Stojanović, CNC-COST National Coordinator

Prof. Dr. Amelia Montone, Chair of COST ACTION MP1103

Dr. Jasmina Grbović Novaković, President of the Scientific and Organizing Committee of 1ERS

10:00-10:15

Coffee break

SESSION I

Plenary

MAIN HALL

Chair: A. Montone, N. Ignjatović

10:15-11:00

**MECANOCHEMISTRY UNDER HYDROGEN GAS:
POTENTIALITIES FOR THE SYNTHESIS OF EFFICIENT
HYDROGEN STORAGE MATERIALS**

Fermin Cuevas, Junxian Zhang, Michel Latroche

Institut de Chimie et des Matériaux Paris-Est, CNRS-UPEC, UMR7182, 2-8
rue Henri Dunant, 94320, Thiais Cedex, France

11:00-11:45

**KEY CONCEPTS IN THE DESIGN OF FUNCTIONAL
NANOPARTICLES**

Vuk Uskoković

Therapeutic Micro and Nanotechnology Laboratory, Department of
Bioengineering and Therapeutic Sciences, University of California San
Francisco, 1700 4th Street, San Francisco, CA 94158-2330, USA

11:45-12:30

**WHAT WE LEARN FROM THE HISTORY TO MASTER THE
FUTURE: THE ROLE OF HYDROGEN**

Andreas Züttel

EMPA, Materials Science & Technology, Dübendorf, Switzerland

12:30-14:00

Lunch break

Chair: Bojana Obradović, Magdalena Stevanović

14:00-14:15

**EFFECTS OF FULLERENOL NANO PARTICLES C₆₀(OH)₂₄ ON
FILAMENTOUS FUNGUS *ASPERGILLUS NIGER* TIEGH**

**Nikola Unković¹, Miloš Stupar¹, Milica Ljaljević Grbić¹, Jelena
Vukojević¹, Marina Seke², Danica Jović², Aleksandar
Djordjević²**

¹University of Belgrade, Faculty of Biology, Institute of Botany and
Botanical Garden "Jevremovac", Takovska 43, 11000 Belgrade, ²Faculty
of Science, Department of Chemistry, Biochemistry and Environment,
Trg Dositeja Obradovića 3, 21000 Novi Sad, University of Novi Sad

14:15-14:30

**PHASE AND MICROSTRUCTURAL EVOLUTION DURING
SINTERING OF Zr-DOPED HYDROXYAPATITE**

**Miodrag J. Lukić¹, Smilja Marković¹, Srečo Davor Škapin²,
Dragan Uskoković¹**

¹Institute of Technical Sciences of SASA, Belgrade, Serbia, ²Institute Jozef
Stefan, Ljubljana, Slovenia

14:30-14:45

**THE BIOACTIVITY INVESTIGATION OF ELECTRODEPOSITED
SILVER/HYDROXYAPATITE/LIGNIN COATINGS IN
SIMULATED BODY FLUID**

**Sanja Eraković¹, Ana Janković¹, Ivana Matić², Zorica Juranić²,
Maja Vukašinović-Sekulić¹, Tatjana Stevanović³, Vesna
Mišković-Stanković¹**

¹Faculty of Technology and Metallurgy, University of Belgrade, Belgrade,
Serbia; ²Institute of Oncology and Radiology of Serbia, University of
Belgrade, Belgrade, Serbia; ³Département des Sciences du Bois et de la
Forêt, Université Laval, Québec, Canada

14:45-15:00

**PRODUCTION AND ANTIMICROBIAL ACTIVITY OF FIBERS
BASED ON ALGINATE WITH INCORPORATED SILVER
NANOPARTICLES**

**Srđan Vidović, Milan Stevanović, Maja Vukašinović-Sekulić,
Bojana Obradović**

Faculty of Technology and Metallurgy, University of Belgrade,
Karnegijeva 4, Belgrade, Serbia

15:00-15:15

EVALUATION OF NOVEL Ag/ALGINATE MICROBEADS FOR POTENTIAL BIOMEDICAL APPLICATIONS

Jasmina Stojkovska, Željka Jovanović, Danijela Kostić, Maja Vukašinović-Sekulić, Vesna Mišković-Stanković, Bojana Obradović

Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4, 11120 Belgrade, Serbia,

15:15-15:30

COPPER – ALGINATE MICROBEADS: BIOACTIVE, ANTIMICROBIAL BIOMATERIALS FOR POTENTIAL BIOMEDICAL APPLICATIONS

Ivana Madžovska, Maja Vukašinović – Sekulić, Bojana Obradović

Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4, 11000 Belgrade, Serbia

15:30-15:45

HEMOLYTIC ACTIVITY OF BIOACTIVE NANOCOMPOSITES

Z. Ajduković¹, N. Ignjatović², **N. Petrović**¹, S. Najman³, J. Rajković⁴, D. Kenić Marinković¹, V. Krstić³, D. Uskoković²

¹University of Niš, Faculty of Medicine, Department of Prosthodontics, Clinic of Stomatology, Niš, Serbia, ² Institute of Technical Sciences of SASA, Belgrade, Serbia, ³ University of Niš, Faculty of Medicine, Institute of Biomedical Research, Niš, Serbia, ⁴ University of Niš, Department of Biology and Ecology, Faculty of Science, Niš, Serbia

15:45-16:00

Coffee break

SESSION II

Biomaterials Science

HALL 16

Chair: Nebojša Nikolić, Igor Pašti

16:00-16:15

CHRONOAMPEROMETRIC DETERMINATION OF HYDROGEN-PEROXIDE BY MULTIWALL CARBON NANOTUBES MODIFIED CARBON BASED ELECTRODES

Jasmina Zbiljić¹, Olga Vajdle¹, Valéria Guzsány¹, Ákos Kukovecz², Zoltán Kónya², Božo Dalmacija¹, Kurt Kalcher³

¹University of Novi Sad, Faculty of Sciences, Department of Chemistry, Biochemistry and Environmental Protection, Trg D. Obradovića 3, 21000 Novi Sad, R. Serbia, ²University of Szeged, Department of Applied and Environmental Chemistry, Rerrich 1, 6720 Szeged, Hungary, ³Karl-Franzens University, Universitätsplatz 3, 8010 Graz, Austria

- 16:15-16:30** **ELECTROCHEMICAL SYNTHESIS AND CHARACTERIZATION OF SILVER NANOPARTICLES IN POLY(VINYL ALCOHOL) SOLUTION**
Rade Surudžić, Željka Jovanović, Vesna Mišković-Stanković
Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4, Belgrade, Serbia
- 16:30-16:45** **VOLTAMMETRIC DETERMINATION OF DOXORUBICIN BY RENEWABLE SILVER-AMALGAM FILM ELECTRODE**
Bojan Tasić, Jasmina Zbiljić, Olga Vajdle, Danica Jović, Valéria Guzsvány, Aleksandar Đorđević
University of Novi Sad, Faculty of Sciences, Department of Chemistry, Biochemistry and Environmental Protection, Trg D. Obradovića 3, 21000 Novi Sad, Serbia
- 16:45-17:00** **CORROSION AND ION RELEASE BEHAVIOR OF NITRIC ACID PASSIVATED STAINLESS STEEL**
Dušan Lj. Petković, Goran M. Radenković
Faculty of Mechanical Engineering, University of Niš, A. Medvedeva 14 Niš, Serbia
- 17:00-17:15** **VOLTAMMETRIC DETERMINATION OF FENOXANIL FUNGICIDE BY RENEWABLE SILVER-AMALGAM FILM ELECTRODE**
Olga Vajdle¹, Mariola Brycht², Jasmina Zbiljić¹, Valéria Guzsvány¹, Sławomira Skrzypek²
¹University of Novi Sad, Faculty of Sciences, Department of Chemistry, Biochemistry and Environmental Protection, Trg D. Obradovića 3, 21000 Novi Sad, Serbia, ²University of Łódź, Department of Instrumental Analysis, Pomorska 163, 90-236, Łódź, Poland
- 17:15-17:30** **CHARACTERIZATION OF ORGANIC INHIBITOR FOR CORROSION OF MILD STEEL IN 3 WT. % NaCl SOLUTION SATURATED WITH CO₂**
Ivana Jevremović¹, Marc Singer², Srđan Nešić², Vesna Mišković-Stanković¹
¹Faculty of Technology and Metallurgy, University of Belgrade, Belgrade, Serbia; ²Institute for Corrosion and Multiphase Technology, Ohio University, Athens, OH, USA

17:30-17:45

THE INFLUENCE OF ADDITIVES ON THE MORPHOLOGICAL AND CORROSION PROPERTIES OF ELECTRODEPOSITED Zn-Mn ALLOYS

M. Bučko¹, U. Lačnjevac², J. Rogan¹, B. Jokić¹, J.B. Bajat¹

¹Faculty of Technology and Metallurgy, University of Belgrade, P.O. Box 3503, 11120 Belgrade, Serbia, ²Institute for Multidisciplinary Research, University of Belgrade, 11030 Belgrade, P.O. Box 33, Serbia

20-22h

Conference dinner

HOTEL MOSKVA

Tuesday, 4th December 2012.

8:30-9:00

Registration

SESSION IV Synthesis, Characterisation and Properties of Materials Part I

HALL 16

Chair: Smilja Marković, Vladimir Srdić

9:00-9:20

IDENTIFICATION OF PROMISING CHEMICAL SYSTEMS FOR THE SYNTHESIS OF NEW MATERIALS STRUCTURE TYPES: AN AB INITIO MINIMIZATION DATA MINING APPROACH

D. Zagorac, J.C. Schön, M. Jansen

Max Planck Institute for Solid State Research, Stuttgart, Germany

9:20-9:35

KINETICS OF CRYSTALLIZATION OF Fe_{89.8}Ni_{1.5}Si_{5.2}B₃C_{0.5} AMORPHOUS ALLOY

Milica M. Vasić¹, Vladimir A. Blagojević¹, Dušan M. Minić², Dragica M. Minić¹

¹Faculty of Physical Chemistry, University of Belgrade, Serbia

²Military Technical Institute, Belgrade, Serbia

9:35-9:50

CHARACTERIZATION OF PULSED LASER DEPOSITED CDS ON TiO₂ NANOTUBES ARRAYS

A. Bjelajac¹, V. Djokić¹, R. Petrović¹, G. Socol², I. Mihailescu², I. Florea³, O. Ersen³, Dj. Janačković¹

¹Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4, 11000 Belgrade, Serbia, ²National Institute for Lasers, Plasma, and Radiation Physics, Bucharest-Magurele, Romania, ³Institut de Physique et Chimie des Materiaux de Strasbourg, CNRS - Universite de Strasbourg (UDS) UMR 7504, 23, rue du Loess, BP 43, 67037 Strasbourg cedex 02, France

9:50-10:05

INFLUENCE OF PARTICLE SIZE AND MORPHOLOGY OF ZNO POWDERS ON THEIR OPTICAL PROPERTIES

A. Stanković¹, Z. Stojanović¹, Lj. Veselinović¹, N. Abazović², S.D. Škapin³, S. Marković¹, D. Uskoković¹

¹Institute of Technical Sciences of SASA, Belgrade, Serbia, ²Vinča Institute of Nuclear Science, Belgrade, Serbia, ³Jožef Štefan Institute, Ljubljana, Slovenia,

10:05-10:20

COMPACTION PRESSURE'S INFLUENCE ON DENSITY AND ELECTRICAL PROPERTIES OF SINTERED CORDIERITE-BASED CERAMICS

N. Obradović¹, N. Đorđević², A. Peleš¹, M. Mitrić³, V. Pavlović¹

¹Institute of Technical Sciences of SASA, Knez Mihajlova 35/IV, 11000 Belgrade, Serbia, ²Institute for Technology of Nuclear and Other Mineral Raw Materials, Bulevar Franse d'Eperea 86, 11000 Belgrade, Serbia, ³Vinča institute of Nuclear Sciences, University of Belgrade, Mike Alasa 12-14, 11000 Belgrade, Serbia

10:20-10:35

DEPOSITION OF THE DLC STRUCTURES IN THE LOW-PRESSURE OXY-ACETYLENE FLAT FLAME

Bojana Radojković¹, Marija Krmar¹, Miroljub Vilotijević², Bojan Gligorijević¹, Ana Alil¹

¹Institute Goša, Milana Rakića 35, Belgrade, Serbia, ²University of Belgrade, Vinča Institute of Nuclear Sciences "", 11001 Belgrade, Serbia

10:35-10:50

CHEMICAL STABILITY OF $\text{Fe}_x(\text{Sb}_2\text{S}_3)_{0.75}(\text{SbI}_3)_{0.25}1-x$ GLASSES

N. Ćelić, M. D. Vučkovac, I. O. Gut, M. Avramov

University of Novi Sad, Faculty of Sciences, Department of Physics, Trg D. Obradovića 4, Novi Sad, Serbia

10:50-11:10

Coffee break

SESSION V

Theoretical Modelling of Materials

HALL 16

Chair: Željka Nikitović, Nebojša Mitrović

- 11:10-11:25** **WELDING PARAMETERS PREDICTION BY THE ANALYTICAL METHODS AND USE OF NUMERICAL SIMULATION SOFTWARE FOR GS-36MN5 STEEL**
S. Budimir, B. Jegdić, **M. Prokolab**, M. Prvulović, Z. Milutinović
Institut Goša, Milana Rakića 35, Beograd, Srbija
- 11:25-11:40** **CALCULATION OF TEMPERATURE DISTRIBUTION OF THE POWDER PARTICLES IN THE SPRAY DEPOSITION PROCESS**
Milorad Kočić, **Mirjana Prvulović**, Jasmina Obradović, Marko Ristić, Milan Prokolab
Institute Goša, Milana Rakića 35, 11000 Belgrade, Serbia
- 11:40-11:55** **APPLICATION OF FINITE ELEMENTS METHOD IN ANALYSIS OF DYNAMIC BEHAVIOR OF COMPOSITE LAMINATES**
Dragan Čukanović¹, Aleksandar Radaković², Miroslav Živković³
¹Faculty of Technical Sciences, Kosovska Mitrovica, ²State University in Novi Pazar, ³Faculty of Engineering, Kragujevac, Serbia
- 11:55-12:15** **PROPAGATION OF BULK WAVES IN A TRANSVERSAL ISOTROPIC MEDIUM**
Radaković Aleksandar¹, Čukanović Dragan², Dragan Milosavljević³, Gordana Bogdanović³, Ljiljana Veljović³
¹State University in Novi Pazar, ²Faculty of Technical Sciences, Kosovska Mitrovica, ³Faculty of Engineering, Kragujevac, Serbia
- 12:15-12:30** **ON THE TRANSITION FROM DELOCALIZED TO LOCALIZED VIBRON STATES IN MACROMOLECULAR CHAINS**
M. Nešić, S. Galović, D. Čevizović
Vinča Institute of Nuclear Sciences, University of Belgrade, Serbia
- 12:30-12:45** **SUBSTITUTION EFFECTS OF SUMANENE BENZYLIC POSITIONS WITH BORON AND NITROGEN ATOMS**
Stevan Armaković¹, Sanja J. Armaković², Igor J. Šetrajčić¹, Jovan P. Šetrajčić^{1,*}
¹University of Novi Sad, Faculty of Sciences, Department of Physics, Trg Dositeja Obradovića 4, 21000, Novi Sad, Vojvodina, Serbia, ²University of

Novi Sad, Faculty of Sciences, Department of Chemistry, Biochemistry and Environmental Protection, Trg Dositeja Obradovića 3, 21000, Novi Sad, Vojvodina, Serbia, *Academy of Sciences and Arts of Republic of Srpska, Bana Lazarevića 1, 78000 Banja Luka, Republic of Srpska, B&H

12:45-13:00

THERMAL PROPERTIES OF CARBON NANOTUBES

Z. P. Popović, M. Damjanović, I. Milošević

NanoLab, Faculty of Physics, University of Belgrade, Serbia

13:00- 13:15

NEWS FROM JEOL - ELECTRON MICROSCOPES AND

SPECTROMETERS – Slavko Žižek, univ.dipl.inž.el.,Scan d.o.o
HALL 16

13:15-14:00

Lunch break

SESSION VI Synthesis, Characterisation and Properties of Materials Part II

HALL 16

Chair: Dragana Jugović, Vladimir Srdić

14:00-14:20

RAMAN SCATTERING STUDY OF IRON-CHALCOGENIDE SUPERCONDUCTORS

N. Lazarević¹, S. Bogdanović¹, Hechang Lei², C. Petrović², Z. V. Popović¹

¹Center for Solid State Physics and New Materials, Institute of Physics Belgrade, University of Belgrade, Pregrevica 118, 11080 Belgrade, Serbia, ²Condensed Matter Physics and Materials Science Department, Brookhaven National Laboratory, Upton, New York 11973-5000, USA

14:20-14:35

EFFECTS OF OPTICAL BEAM MULTIPLE REFLECTIONS TO PHOTOTHERMAL RESPONSE - CORRECTION FACTOR

Slobodan Todosijević¹, Zlatan Šoškić¹, Nebojša Bogojević¹, Dalibor Čevizović², Slobodanka Galović^{1,2,3}

¹Faculty of Mech. Eng. Kraljevo, University of Kragujevac, Dositejeva 19, Kraljevo, Serbia, ²Vinča Institute of Nuclear Sciences, University of Belgrade, p.o.Box 522, Belgrade, Serbia, ³Joint Institute for Nuclear Research, Bogoliubov Laboratory of Theoretical Physics, Dubna, Russia

- 14:35-14:50** **SiC MOSFET – MODEL OF CURRENT-VOLTAGE CHARACTERISTICS AND POSSIBILITIES FOR IT'S IMPLEMENTATION IN BIOMEDICAL ELECTRO EQUIPMENT**
Zorana Z. Golubović¹, Petar M. Lukić¹, Milan T. Milovanović², Vladan M. Lukić¹, Rajko M. Šašić³
¹Faculty of Mechanical Engineering, University of Belgrade, Kraljice Marije 16, 11120 Belgrade, Serbia, ²Military Medical Academy, Crnotravska 17, 11000 Belgrade, Serbia, ³Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4, 11120 Belgrade
- 14:50-15:05** **STUDY OF THE SHUNGITE AND COANP STRUCTURES NONLINEAR PROPERTIES**
Svetlana Likhomanova, Natalia V. Kamanina
Vavilov State Optical Institute, 12, Birzhevaya Line, St.-Petersburg, 199034, Russia
- 15:05-15:20** **STUDY OF THE CARBON NANOTUBES INFLUENCE ON THE REFLECTION SPECTRUM AND THE MICROHARDNESS OF KBr, BaF₂**
Pavel V. Kuzhakov, Natalia V. Kamanina
Vavilov State Optical Institute, 12, Birzhevaya Line, St.-Petersburg, 199034, Russia
- 15:20-15:35** **TG-DTG STUDY OF BULK As₂S₃ CHALCOGENIDE GLASSES DOPED WITH BISMUTH**
M. R. Dimitrievska, M. Šeljegović, Lj. Dačanin, K. Čajko
University of Novi Sad, Department of Physics, Trg Dositeja Obradovića 4, Novi Sad, Serbia
- 15:35-15:50** **SYNTHESIS AND CHARACTERIZATION OF NiFe₂O₄/BaTiO₃ THIN FILMS**
Branimir Bajac^{1,*}, Jelena Vukmirović¹, Ivan Tokić¹, Stevan M. Ognjanović¹, Akos Kukovecz², Vladimir V. Srdić¹
¹Department of Materials Engineering, Faculty of Technology, University of Novi Sad, Novi Sad, Serbia, ²Department of Applied and Environmental Chemistry, University of Szeged, Szeged, Hungary
- 15:50-16:05** **PHASE DIAGRAM OF SYSTEM ZnGeAs₂-CdGeAs₂**
I.V. Fedorchenko¹, A.N. Aronov¹, A.H. Yusupov², L. Kilanski³, V. Domuchowski³
¹Kurnakov Institute of General and Inorganic Chemistry Russian Academy of Sciences, 119991, Moscow, Russia, Leninskiy pr.31,

²National University of Science and Technology "MISIS", 119049, Moscow, Russia, Leninskiy pr.4, ³Institute of physics, Polish Academy of Sciences, 02-668, Warsaw, Poland, al. Lotnikow 32/46

16:15 -18:00 POSTER SESSION and coffee break
SASA Gallery of Science and Technology, Đure Jakšića 2
Chair: Bojana Obradović, Željka Nikitović, Smilja Marković

Wednesday, 5th December 2012.

9:00-9:30 Registration

SESSION VII	Polymeric Materials	HALL 16
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Chair: Gordana Ćirić-Marjanović, Edin Suljovrujić

9:30-9:45 THE INFLUENCE OF THE STRUCTURE OF THE POLYBUTADIENE BLOCK TO THE MECHANICAL PROPERTIES OF THE POLYSTYRENE-*b*-POLYBUTADIENE-*b*-POLY(METHYL METHACRYLATE) COPOLYMER

Dragutin Nedeljković, Aleksandar Stajčić, Aleksandar Grujić, Jasmina Stevanović, Jasna Stajić-Trosić
University of Belgrade, Institute of Chemistry, Technology and Metallurgy, Njegoševa 12, Belgrade, Serbia

9:45-10:00 INFLUENCE OF GOLD IONS IMPLANTATION AND OZONE TREATMENT ON SURFACE CHARACTERISTICS OF HIGH DENSITY POLYETHYLENE

Danilo Kisić¹, Miloš Nenadović¹, Borivoj Adnađević², Zlatko Rakočević¹

¹INS Vinča, Laboratory of Atomic Physics, University of Belgrade, Mike Alasa 12-14, 11001 Belgrade, Serbia, ²Faculty of Physical Chemistry, University of Belgrade, Studentski trg 12-16, 11158 Belgrade, Serbia

10:00-10:15 SURFACE DIELECTRIC PROPERTIES AND FOURIER TRANSFORM INFRARED (FTIR) SPECTROSCOPY SPECTRA OF LOW-DENSITY POLYETHYLENE (LDPE) ETCHED BY TRICHLOROACETIC ACID AT DIFFERENT TEMPERATURES

K. Simonović¹, F. Marinković¹, J. Dojčilović¹, D. Dudić²

¹Faculty of Physics, University of Belgrade, Studentski trg 12-16, 11001 Belgrade, Serbia, ²Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001 Belgrade, Serbia

10:15-10:30

NOVEL ASYMMETRIC POLYETHERSULFONE MEMBRANES FOR ULTRAFILTRATION APPLICATION

Aleksandra Nešić¹, Sava Veličković², Filip Radovanović³, Aleksandra Nastasović³

¹Vinča Institute for Nuclear Sciences, University of Belgrade, Mike Petrovića Alasa 12-14, Belgrade, Serbia, ²Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4, Belgrade, Serbia, ³Institute for Chemistry, Technology and Metallurgy, University of Belgrade, Studentski trg 16, Belgrade, Serbia

10:30-10:45

SURFACE MODIFICATION OF PECTIN SPHERES CROSS-LINKED IN LEAD(IV) AND COPPER(II) SOLUTIONS

Sanja Šešlija¹, Jasmina Stevanović², Tatjana Volkov-Husović³, Sava Veličković³

¹Innovation Centre of the Faculty of Technology and Metallurgy, University of Belgrade, Belgrade, Serbia, ²Institute of Chemistry, Technology and Metallurgy, Department of Electrochemistry, Belgrade, Serbia, ³University of Belgrade, Faculty of Technology and Metallurgy, Belgrade, Serbia

10:45-11:00

Coffee break

SESSION VII

Polymeric Materials

HALL 16

Chair: Gordana Ćirić-Marjanović, Edin Suljovrujić

11:00-11:15

FERROMAGNETIC POLYANILINE/TiO₂ NANOCOMPOSITES

Marija Radoičić¹, Zoran Šaponjić¹, Gordana Ćirić-Marjanović², Zorica Konstantinović³, Jovan Nedeljković¹

¹Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia, ²Faculty of Physical Chemistry, University of Belgrade, Belgrade 11158, Serbia, ³Institut de Ciència de Materials de Barcelona, CSIC, Bellaterra 08193, Spain

11:15-11:30

COMPARISON OF DIELECTRIC PROPERTIES OF POLYVINYLDENE FLUORIDE + Cu COMPOSITES MEASURED BY SINUSOIDAL AND TRIANGLE SIGNALS

I. Petronijević¹, K. Simonović¹, R. Dojčilović², V. Đoković², D. Dudić²

¹Faculty of Physics, University of Belgrade, Studentski trg 12-16, 11001 Belgrade, Serbia, ²Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001 Belgrade, Serbia

11:30-11:45

POLY(METHACRYLIC ACID)/ZEOLITE HYDROGEL COMPOSITES AS ADSORBENTS FOR CATIONIC DYE REMOVAL

Vesna V. Panić¹, Željka Madžarević², Tatjana Volkov-Husović², Sava J. Veličković²

¹Innovation Center of Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4, Belgrade, Serbia, ²Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4, Belgrade, Serbia

11:45-12:00

SYNTHESIS OF POLYSTYRENE-G-STARCH COPOLYMERS USING NEW INITIATOR/CO-INITIATOR SYSTEM

Vladimir Nikolić¹, Veličković Sava², Popović Aleksandar³

¹Innovation Center, Faculty of Chemistry, University of Belgrade, Studentski trg 12-16, 11000 Belgrade, Serbia, ²Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4, 11000 Belgrade, Serbia, ³Faculty of Chemistry, University of Belgrade, Studentski trg 12-16, 11000 Belgrade, Serbia

12:00-12:15

A SURVEY OF GRP PIPES DEFECTS AND DAMAGE DUE TO FABRICATION PROCESSES

Abdelrhman Aljadeed¹, Ahamed Younaes¹, Ramadan A. Al-Madani¹, M. Jarnaz²

¹Al-Jabel Algharbi University, Engineering Faculty, Gharian, Libya,

²Academy of Graduate Studies, Tripoli - Libya

12:15-12:30

Awards and closing ceremony

POSTER SESSION 11YRC SASA Gallery of Science and Technology (Đure Jakšića 2)

NOTE: Please bring the posters to the conference registration desk on 03.12.2012

Y1. INVESTIGATION OF THE STRUCTURE OF NANOSIZED Na_nCl_n ($n = 8, 16, 24, 32$) CLUSTERS USING GLOBAL OPTIMIZATION

J. Zagorac^{1,2}, J. C. Schön¹, M. Jansen¹

¹Max Planck Institute for Solid State Research, Stuttgart, Germany, ²Materials Science Laboratory, Vinča Institute of Nuclear Sciences, Serbia

Y2. ELECTRICAL PROPERTIES OF MULTIDOPED CERIA

M. Stojmenović¹, B. Matović¹, M. Žunić²

¹Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia, ²Institute for Multidisciplinary Research, University of Belgrade, Belgrade, Serbia

Y3. DOPED WITH Cu AND Mn ZINC OXIDE CATALYSTS FOR OXIDATION OF CO IN NOXIOUS EMISSIONS

Katya Milenova¹, Penko Nikolov²

¹Institute of Catalysis, Bulgarian Academy of Sciences, Akad. G. Bontchev str. bl 11, 1113, Sofia, Bulgaria, ²University of Chemical Technology and Metallurgy, 8 Kliment Ohridski, 1756, Sofia, Bulgaria

Y4. MINIMAL REQUIRED WALL THICKNESS COMPARATIVE ANALYSIS OF A STABLE TANK MADE OF STAINLESS STEEL WITH AND WITHOUT STRAIN HARDENING

Katarina Jovičić¹, Iva Najvirt², Ana Alil¹, Milica Pilipović³

¹Institut Goša d.o.o., Milana Rakića 35, 11000 Belgrade, Serbia, ²Nordvik d.o.o., Pariske Komune 22, 11000 Belgrade, Serbia, ³Grundfos Srbija d.o.o., Omladinskih Brigada 90b, 11000 Belgrade, Serbia

Y5. REDUCING THE PARTICLE SIZE OF Bi_2O_3 AND Fe_2O_3 FOR THE SYNTHESIS OF BiFeO_3

Maria Čebela, Marija Prekajski, Mia Omerašević, Branko Matović

Materials Science Laboratory, Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

Y6. COMPARATIVE ANALYSIS OF CESIUM SORPTION BEHAVIOR OF THERMALLY, MECHANO-CHEMICALLY MODIFIED AND RAW DIATOMITE

Mia O. Omerašević¹, Miljana M. Mirković¹, Uroš D. Jovanović², Anja Došen¹, Ljiljana M. Kljajević¹, Maria Ž. Čebela¹, Snežana S. Nenadović¹

¹Laboratory for Material Science, Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia, ²Laboratory of Chemical Dynamics and Permanent Education, Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

Y7. ELECTRON SPIN RESONANCE STUDY OF GAMMA-IRRADIATED PLLA

D. Miličević, D. Milivojević, E. Suljovrujić

Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

Y8. ADSORPTION OF ARSENIC(III) FROM AQUEOUS SOLUTIONS ON CARBON CRYOGEL

Tamara Minović, Biljana Babić, Milovan Stoilković, Vesna Maksimović, Jelena Gulicovski, Jelena Pantić, Branko Matović

Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

Y9. THE EFFECT OF CHEMICAL COMPOSITION ON SWELLING AND MECHANICAL PROPERTIES OF P(HEMA/IA/OEGA) HYDROGELS

M. Mičić, E. Suljovrujić

Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

Y10. NUMERICAL SIMULATION OF ROPEWAY OF AIRLIFT VEHICLES: CALCULATION METHODS

Miranda Kullolli, **Alfred Hasanaj**

Polytechnic University of Tirana, Albania

Y11. LOCAL AND ELECTRONIC STRUCTURE AROUND MANGANESE IN MULTI-COMPONENT SEMICONDUCTORS

M. Medić, I. Radisavljević, N. Novaković, B. Kuzmanović, B. Paskaš Mamula, N. Ivanović

Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001 Belgrade, Serbia

Y12. SURFACE DIELECTRIC RELAXATIONS OF ISOTACTIC POLYPROPYLENE (iPP) AFTER ILLUMINATION AT DIFFERENT TEMPERATURES

V. Čubrović¹, I. Petronijević¹, J. Dojčilović¹, D. Dudić²

¹Faculty of Physics, University of Belgrade, Studentski trg 12-16, 11001 Belgrade, Serbia

²Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001 Belgrade, Serbia

Y13. SURFACE DIELECTRIC PROPERTIES OF LOW-DENSITY POLYETHYLENE (LDPE) IN VARIABLE VISIBLE LIGHT CONDITIONS

F. Marinković¹, V. Čubrović¹, J. Dojčilović¹, D. Dudić²

¹Faculty of Physics, University of Belgrade, Studentski trg 12-16, 11001 Belgrade, Serbia, ²Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001 Belgrade, Serbia

Y14. NEW FACILE SYNTHESIS ROUTE FOR OBTAINING PHASE PURE LiFePO_4/C COMPOSITE

M. Milović¹, D. Jugović¹, M. Mitrić², B. Jokić³, D. Uskoković¹

¹Institute of Technical Sciences of SASA, Belgrade, Serbia, ²Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia, ³Faculty of Technology and Metallurgy, University of Belgrade, Belgrade, Serbia

Y15. BIOACTIVE COMPOSITE MATERIALS IN REGENERATION OF THE RESORBED BONE OF ALVEOLAR RIDGES

Z. Ajduković¹, N. Ignjatović², **J. Rajković³**, S. Najman⁴, D. Mihailović⁵, N. Petrović¹, D. Kenić Marinković¹, D. Uskoković²

¹Department of Prosthodontics, Clinic of Stomatology, Faculty of Medicine University of Niš, Serbia, ²Institute of Technical Sciences of SASA, Belgrade, Serbia, ³Department of Biology and Ecology, Faculty of Science and Mathematics, University of Niš, Serbia, ⁴Institute for Biomedical Research, Faculty of Medicine, University of Niš, Serbia, ⁵Institute of Pathology, Faculty of Medicine, University of Niš, Serbia

Y16. EFFECTS OF DIFFERENT CRYOPROTECTANTS ON MORPHOLOGY OF LYOPHILIZED POLY(E-CAPROLACTONE) MICRO- AND NANOSPHERES

Petar Stupar¹, Magdalena Stevanović², Nenad Filipović², Vladimir Pavlović^{2,3}, Jana Nunić⁴, Sandra Cundrič⁵, Metka Filipič⁴, Dragan Uskoković²

¹Faculty of Physical Chemistry, University of Belgrade, Belgrade, Serbia, ²Institute of Technical Sciences of, Belgrade, Serbia, ³Faculty of Agriculture, University of Belgrade, Belgrade, Serbia, ⁴Department of Genetic Toxicology and Cancer Biology, National Institute of Biology, Ljubljana, Slovenia, ⁵Biotechnical Faculty, University of Ljubljana, Ljubljana, Slovenia

Y17. SYNTHESIS AND CHARACTERIZATION OF SELENIUM NANOPARTICLES IN THE PRESENCE OF BOVINE SERUM ALBUMIN OR POLY (L-GLUTAMIC ACID) FOR BIOMEDICAL APPLICATION

Nenad Filipović¹, Magdalena Stevanović¹, Srečo D. Škapin², Ines Bračko², Dragan P. Uskoković¹

¹Institute of Technical Sciences of SASA, Knez Mihailova 35/IV, 11000 Belgrade, Serbia, ²Advanced Materials Department, Jožef Štefan Institute, Jamova 39, 1000 Ljubljana, Slovenia

Y18. THE EFFECT OF AGING OF MILLED CLAY ON THE REMOVAL EFFICIENCIES OF HEAVY METAL IONS

Anđelka Đukić¹, Ksenija Kumrić², Tatjana Trtić-Petrović², Jasmina Grbović Novaković¹, Sanja Milošević¹, Igor Milanović¹, Ljiljana Matović¹

¹Laboratory of Materials Sciences, Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001 Belgrade, Serbia, ²Laboratory of Physics, Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001 Belgrade, Serbia

Y19. DRUG DELIVERY PARADIGM OF SILVER IONS RELEASED FROM AG/PVA HYDROGEL NANOCOMPOSITES OBTAINED BY RADIATION CHEMISTRY NANOREACTOR METHOD

Jelena Krstić, Jelena Spasojević, Aleksandra Radosavljević, Zorica Kačarević-Popović

Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

Y20. SWELLING STUDIES OF THERMO- AND pH-SENSITIVE Ag-POLY(NiPAAm/IA) HYDROGEL NANOCOMPOSITES SYNTHESIZED BY GAMMA IRRADIATION

Jelena Spasojević¹, Jelena Krstić¹, Aleksandra Radosavljević¹, Melina Kalagasidis-Krušić², Zorica Kačarević-Popović¹

¹Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia, ²Faculty of Technology and Metallurgy, University of Belgrade, Belgrade, Serbia

Y21. KINETIC STUDY OF PALLADIUM SORPTION BY MACROPOROUS COPOLYMER

Bojana Ekmešić¹, Danijela Maksin², Jelena Marković², Aleksandra Nastasović¹, Antonije Onjia²

¹University of Belgrade, Institute for Chemistry, Technology and Metallurgy, Njegoševa 12, Belgrade, Serbia, ²University of Belgrade, Vinča Institute of Nuclear Sciences, Mike Petrovića Alasa 12-14, Vinča, Serbia

Y22. EXAMINATION OF THE INFLUENCE OF NANOMATERIALS CALCIUM PHOSPHATE/POLY-(DL-LACTIDE-CO-GLYCOLIDE) AND COBALT-EXCHANGED HYDROXYAPATITE ON THE VIABILITY OF SAOS-2 CELLS

Stevo Najman¹, **Sanja Stojanović¹**, Žarko Mitić², Zorica Ajduković³, Nenad Ignjatović⁴, Dragan Uskoković⁴

¹University of Niš, Faculty of Medicine, Institute of Biology and Human Genetics, Serbia, ²University of Niš, Faculty of Medicine, Department of Pharmacy, Serbia, ³University of Niš, Faculty of Medicine, Clinic of Stomatology, Department of Prosthodontics, Niš, Serbia, ⁴Institute of Technical Sciences of SASA, Belgrade, Serbia

*Joint event of the 11th Young Researchers' Conference: Materials Science and Engineering and
the 1st European Early Stage Researchers' Conference on Hydrogen Storage*

PROGRAM

of

**The 1st EUROPEAN EARLY STAGE RESEARCHERS' CONFERENCE
on HYDROGEN STORAGE
COST MP1103**

25

Conference program 1EESRC

Monday, 3rd December 2012.

8:30-9:30

Registration

9:30-10:00

Opening ceremony

MAIN HALL

Prof. Dr. Nenad Ignjatović, President of the Scientific and Organizing Committee of 11YRC

Prof. Dr. Ivanka Popović, Vice-Rector, University of Belgrade

Prof. Dr. Dragan Uskoković, President of the Materials Research Society of Serbia

Dr. Paolo Battinelli Scientific Attache', Italian Embassy in Belgrade

Prof. Dr. Biljana Stojanović, CNC-COST National Coordinator

Prof. Dr. Amelia Montone, Chair of COST ACTION MP1103

Dr. Jasmina Grbović Novaković, President of the Scientific and Organizing Committee of 1EESRC

10:00-10:15

Coffee break

SESSION I

Plenary

MAIN HALL

Chair: A. Montone, N. Ignjatović

10:15-11:00

**MECANOCHEMISTRY UNDER HYDROGEN GAS:
POTENTIALITIES FOR THE SYNTHESIS OF EFFICIENT
HYDROGEN STORAGE MATERIALS**

Fermin Cuevas, Junxian Zhang, Michel Latroche

Institut de Chimie et des Matériaux Paris-Est, CNRS-UPEC, UMR7182, 2-8 rue Henri Dunant, 94320, Thiais Cedex, France

11:00-11:45

**KEY CONCEPTS IN THE DESIGN OF FUNCTIONAL
NANOPARTICLES**

Vuk Uskoković

Therapeutic Micro and Nanotechnology Laboratory, Department of Bioengineering and Therapeutic Sciences, University of California San Francisco, 1700 4th Street, San Francisco, CA 94158-2330, USA

11:45-12:30

**WHAT WE LEARN FROM THE HISTORY TO MASTER THE
FUTURE: THE ROLE OF HYDROGEN**

Andreas Züttel

EMPA, Materials Science & Technology, Dübendorf, Switzerland

12:30-14:00

Lunch break

Chair: E. Callini, S. Milošević, M. Lelis

14:00-14:25

**POLYMORPHISM AND STABILITY OF MAGNESIUM
BOROHYDRIDE**

V. Ban¹, R. Černý², B. Richter³, T.R. Jensen³, C.J. Webb⁴, Y. Filinchuk¹

¹Institute of Condensed Matter and Nanosciences, Université Catholique de Louvain, Place L. Pasteur 1, 1348 Louvain-la-Neuve, Belgium; ²Laboratory of Crystallography, Université de Genève, 24 quai Ernest Ansermet, 1211 Genève, Switzerland; ³Center for Materials Crystallography, Interdisciplinary Nanoscience Center & Dept. of Chemistry, Aarhus University, Langelandsgade 140, 8000 Aarhus C, Denmark, ⁴Queensland Micro- and Nanotechnology Centre, Griffith University, Nathan 4111, Brisbane, Australia

14:25-14:50

**ACHIEVING HIGH CAPACITY REVERSIBLE HYDROGEN
STORAGE: NaBH₄@Ni CORE-SHELL NANOPARTICLES**

Meganne Christian, Kondo-François Aguey-Zinsou
Merlin Group, ARC Centre for Functional Nanomaterials, School of Chemical Engineering, The University of New South Wales, Sydney NSW 2052, Australia

14:50-15:15

**HYDRAZINE BORANE AND DERIVATIVES AS CHEMICAL
HYDROGEN STORAGE MATERIALS**

Romain Moury, Umit B. Demirci, Philippe Miele
IEM (Institut Européen des Membranes), UMR 5635 (CNRS-ENSCM-UM2), Université Montpellier 2, Place E. Bataillon, F- 34095, Montpellier, France

15:15-15:30

**PHYSICAL AND CHEMICAL PROPERTIES OF VOLATILE
COMPLEX HYDRIDES**

E. Callini¹, A. Borgschulte¹, I. Lindemann², A.J. Ramirez-Cuesta³, M. Chong⁴, C.M. Jensen⁴, O. Gutfleisch^{2,5}, A. Züttel¹
¹Empa, Materials Sciences and Technology, CH-8600 Dübendorf, Switzerland; ²IFW Dresden, Helmholtzstrasse 20, D-01069 Dresden, Germany; ³ISIS Facility, Rutherford Appleton Laboratory, Chilton, Didcot, Oxon OX11 0QX, United Kingdom; ⁴Univ Hawaii Manoa, Dept Chem, Honolulu, HI 96822 USA, ⁵Technische Universität Darmstadt, Petersenstrasse 23, 64287 Darmstadt, Germany

15:30-15:45

TUNING THE HYDROGEN STORAGE OF MOFS BY INTRODUCING METALS ON THE LINKER SIDE

Petra Ágota Szilágyi, Jana-Juan Alcañiz, Jorge Gascon, Hans Geerlings, Bernard Dam
Delft University of Technology, Netherlands

15:45-16:00

Coffee break

SESSION III

Various Problems in Hydrogen Application

HALL 8

Chair: Ž. Rašković-Lovre, A. Bliersbach, D. Mirabile Gattia

16:00-16:25

THE EFFECTS OF SI AND EXPANDED PTFE SUBSTRATES ON FORMATION AND HYDROGENATION OF Mg AND Mg-Ti FILMS

M. Lelis^{1,2}, D. Milcius¹, R. Zostautiene³

¹Lithuanian Energy Institute, Breslaujos st.3, LT-44403 Kaunas, Lithuania;

²Vytautas Magnus University, Vileikos st. 8, LT-44404 Kaunas, Lithuania;

³Kaunas University of Technology, K.Donelaičio st.73, LT-44029 Kaunas, Lithuania

16:25-16:50

METAL HYDRIDE FILMS FOR OPTO-ELECTRONIC APPLICATIONS

Trygve Mongstad

Institute for Energy Technology, Norway

16:50-17:15

LASER SURFACE MODIFICATION OF MATERIALS

Sumsun Naher

Faculty of Engineering and Computing, Dublin City University, Dublin 9, Ireland

17:15-17:40

MODIFICATIONS IN MICROSTRUCTURE OF MgH₂-BASED COMPOSITE UPON HYDROGEN CYCLING

A. Montone, A. Aurora, **D. Mirabile Gattia**, M. Vittori Antisari

ENEA – R.C. Casaccia – Technical Unit of Materials - Via Anguillarese, 301 00123 Rome, Italy

20-22h

Conference dinner

HOTEL MOSKVA

Tuesday, 4th December 2012.

8:30-9:00

Registration

SESSION IV

Hydrogen in Alloys

HALL 8

Chair: M. Ponthieu, M. Sahlberg, S. Kurko

9:00-9:25

SYNTHESIS AND HYDROGENATION OF $\text{Mg}_6\text{Pd}_{1-x}\text{TM}_x$ (TM = Ag, Cu) NOVEL SYSTEMS

M. Ponthieu^{1,2}, J. F. Fernandez¹, F. Cuevas², J.R. Ares¹, P. Adeva³, C. Sanchez¹

¹MIRE, Dpto. Física de Materiales, Facultad de Ciencias, Universidad Autónoma de Madrid, 28049, Madrid, Spain; ²ICMPE/CNRS-UPEC UMR 7182, 2-8 rue Henri Dunant, 94320 Thiais Cedex, France; ³Department of Physical Metallurgy, CENIM-CSIC, Av. De Gregorio del Amo 8, 28040 Madrid, Spain

9:25-9:50

STRUCTURE AND HYDROGEN STORAGE PROPERTIES OF THE HEXAGONAL LAVES PHASE $\text{Sc}(\text{Al}_{1-x}\text{Ni}_x)_2$

Martin Sahlberg

Department of Chemistry – Ångström Laboratory, Uppsala University, Box 538, SE-751 21 Uppsala, Sweden

9:50-10:15

MICROSTRUCTURE EVOLUTION IN THE CHIPS OF Mg-10 at.% Ni ALLOY DURING HYDROGENATION

Larisa Popilevsky¹, Vladimir Skripnyuk¹, Yuri Estrin², A.K. Dahle³, Eugen Rabkin¹

¹Department of Materials Science and Engineering, Technion-Israel, Institute of Technology, Haifa, Israel; ²Centre for Advanced Hybrid Materials, Monash University, Clayton, Victoria 3800, Australia; ³Materials Engineering, The University of Queensland, Frak White Bldg. #43, Brisbane, QLD 4072, Australia

10:15-10:40

COMBINATORIAL SYNTHESIS AND HIGH THROUGHPUT EXAMINATION OF HYDROGEN STORAGE ALLOY LIBRARIES

M. Polanski, I. Kuncze, J. Bystrzycki

Department of Advanced Materials and Technology, Military University of Technology, 2 Kaliskiego Str., 00-908 Warsaw, Poland

10:40-10:55

**STRUCTURAL CHARACTERIZATION AND ELECTROCHEMICAL
HYDROGEN STORAGE PROPERTIES OF Ti₂Ni AND (Ti_{1-x}Zr_x)₂Ni (x=0, 0.05, 0.1) ALLOYS PREPARED BY MECHANICAL
ALLOYING**

Xianda Li¹, Omar Elkedim¹, M. Nowak², M. Jurczyk²

¹FEMTO-ST, Fc-Lab, Université de Technologie de Belfort-Montbéliard,
Site de Sévenans, 90010 Belfort cedex, France; ²Institute of Materials
Science and Engineering, Poznan University of Technology, Skłodowska-
Curie 5 Sq., 60-965 Poznan, Poland

10:55-11:10

Coffee break

SESSION V Theoretical Investigation of Materials for Hydrogen Storage HALL 8

Chair: P. B. Jensen, R. Vujasin, N. Novaković

11:10-11:35

HYDROGEN STORAGE PROPERTIES OF SUMANENE

Stevan Armaković¹, Sanja J. Armaković², Jovan P. Šetrajčić^{1,*}

¹University of Novi Sad, Faculty of Sciences, Department of Physics, Trg
Dositeja Obradovića 4, 21000, Novi Sad, Vojvodina, Serbia; ²University of
Novi Sad, Faculty of Sciences, Department of Chemistry, Biochemistry
and Environmental Protection, Trg Dositeja Obradovića 3, 21000, Novi
Sad, Vojvodina, Serbia; *Academy of Sciences and Arts of Republic of
Srpska, Bana Lazarevića 1, 78000 Banja Luka, Republic of Srpska, B&H

11:35-12:00

**INVESTIGATION OF SURFACE AND NEAR-SURFACE EFFECTS
ON HYDROGEN DESORPTION PROPERTIES OF MgH₂**

Sandra Kurko, Sanja Milošević, Igor Milanović, Radojka
Vujasin, Ljiljana Matović, Jasmina Grbović Novaković, Nikola
Novaković

Vinča Institute of Nuclear Sciences, University of Belgrade, POB 522,
11000 Belgrade, Serbia

12:00-12:25

**COMPUTATIONAL STUDY OF H STORAGE IN Pd AND Pt
NANOPARTICLES SUPPORTED ON MgO**

Sergey M. Kozlov¹, Hristiyan A. Aleksandrov^{1,2}, Konstantin
M. Neyman^{1,3}

¹Departament de Química Física and Institut de Química Teòrica i
Computacional (IQTCUB), Universitat de Barcelona, C/Marti Franques 1,
08028 Barcelona, Spain; ²Faculty of Chemistry, University of Sofia, 1126
Sofia, Bulgaria; ³Institució Catalana de Recerca i Estudis Avançats
(ICREA), 08010 Barcelona, Spain

- 12:25-12:40** **COMPUTATIONAL SCREENING OF MIXED METAL HALIDE AMMINES**
Peter Bjerre Jensen¹, Steen Lysgaard¹, Ulrich Quaade², Tejs Vegge¹
¹DTU Energy Conversion, Technical University of Denmark, Fysikvej 307, DK-2800 Kgs. Lyngby ²AmmineX A/S, Gladsaxevej 363, DK-2860 Søborg
- 12:40-12:55** **STRUCTURAL AND HYDROGEN SORPTION PROPERTIES OF SmNi_{5x}Ga SYSTEM – AN EXPERIMENTAL AND THEORETICAL STUDY**
Nikola Biliškov, Goran I. Miletić, Ivka Štefanić, Antun Drašner
Laboratory of Solid State and Complex Compounds Chemistry, Division of Materials Chemistry, Ruđer Bošković Institute, Bijenička c. 54, HR-10000 Zagreb, Croatia
- 13:00-13:15** **NEWS FROM JEOL - ELECTRON MICROSCOPES AND SPECTROMETERS – Slavko Žižek, univ.dipl.inž.el., Scan d.o.o HALL 16**

12:55-14:00 Lunch break

SESSION VI	Synthesis and Characterisation	HALL 8
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Chair: M. Polanski, H. Fahlquist, Lj. Matović

- 14:00-14:25** **INDUCTION PLASMA SYNTHESIS OF Mg₂Ni NANOPARTICLES FOR HYDROGEN STORAGE**
Burak Aktekin¹, Gülhan Çakmak², Tayfur Öztürk¹
¹Middle East Technical University, Metallurgical and Materials Eng. Dept., Ankara, Turkey; ²Bülent Ecevit University, Metallurgical Eng. Dept., Zonguldak, Turkey
- 14:25-14:40** **POLYANIONIC GALLIUM-HYDROGEN CLUSTER IONS AS POSSIBLE INTERMEDIATES IN DEHYDROGENATION REACTIONS**
Henrik Fahlquist
Stockholm University, 106 91 Stockholm, Sweden

14:40-15:05 **EFFECT OF A $\text{CaF}_{2-x}\text{H}_x$ SOLID SOLUTION FORMATION IN THE MIXED SYSTEM $\text{CaH}_2+\text{CaF}_2+\text{MgB}_2$**

C. Pistidda¹, F. Karimi¹, S. Garroni², C. Milanese³, L. Rude⁴, J. Skibsted⁴, T. R. Jensen⁴, P. Nolis⁵, C. Horstmann¹, C. Gundlach⁶, M. D. Baró⁷, T. Klassen¹, M. Dornheim¹

¹Institute of Materials Research, Materials Technology, Helmholtz-Zentrum Geesthacht, Max-Planck-Straße 1, D-21502 Geesthacht, Germany; ²Dipartimento di Chimica, Università di Sassari and INSTM, Via Vienna 2, I-07100 Sassari, Italy; ³Pavia H2 Lab, C.S.G.I. & Dipartimento di Chimica, Sezione di Chimica Fisica, Università di Pavia, Viale Taramelli 16, I-27100 Pavia, Italy; ⁴Center for Materials Crystallography, iNANO, and Department of Chemistry, Aarhus University, Langelandsgade 140, DK-8000. Denmark; ⁵Servei de Ressonància Magnètica Nuclear (SeRMN), Universitat Autònoma de Barcelona, 08193 Bellaterra, Spain; ⁶MAX-lab, Lund University, S-22100 Lund, Sweden; ⁷Departament de Física, Universitat Autònoma de Barcelona, 08193 Bellaterra, Spain

15:05-15:30 **DESIGN, SIMULATION AND OPTIMISATION OF AN ANODE SUPPORTED SOFC**

Hirad Azmin, Zahir Dehouche

School of Engineering and Design, Brunel University, Middlesex, UK

15:30-18:30

SG MEETING

HALL 8

Chair: B. Paskaš Mamula, I. Milanović, Á. Kis-Tóth

16:15 -18:00

POSTER SESSION and coffee break

SASA Gallery of Science and Technology, Đure Jakšića 2

Wednesday, 5th December 2012.

9:00-9:30

Registration

SESSION VII Carbon Based Materials for Hydrogen Storage

HALL 8

Chair: A. Ampoumogli, P. Javadian, A. Nale

9:30-9:55

HYDROGEN STORAGE IN POROUS MEDIA AT ROOM TEMPERATURE

Liga Grinberga, Janis Kleperis

Institute of Solid State Physics, University of Latvia, 8 Kengaraga Street, Riga, LV-1063, Latvia

9:55-10:20

ENHANCING THE HYDROGEN CARBON INTERACTION

Andreas Bliersbach

Empa, Swiss Federal Laboratories for Materials Science and Technology, Hydrogen and Energy, Überlandstrasse 129, CH-8600 Dübendorf, Switzerland

10:20-10:45

HYDROGEN STORAGE PROPERTIES OF DIFFERENT METAL-DOPED CARBON-BASED COMPOSITES

A. Ampoumogli¹, D. Giasafaki¹, A. Bourlinos²,
G. Charalambopoulou¹, A. Stubos¹, D. Gournis³,
Th. Steriotis¹

¹National Center for Scientific Research "Demokritos", Athens, Greece;

²Department of Physics, University of Ioannina, Greece; ³Department of Material Science Engineering, University of Ioannina, Greece

10:45-11:00

Coffee break

SESSION VIII

Carbon Based Materials for Hydrogen Storage

HALL 8

Chair: A. Ampoumogli, P. Javadian, A. Nale

11:00-11:25

PRODUCTION OF SINGLE – WALL CARBON NANOTUBES BY KRF EXCIMER LASER ABLATION

Petru-Marian Bota, Iulian Boerasu, Dorel Dorobantu,
Dionezie Bojin, Marius Enachescu

Centre of Surface Science and Nanotechnology – University Politehnica of Bucharest, Splaiul Independentei 313, Bucharest, Romania

- 11:25-11:50** **CONFINEMENT OF LIBH₄ IN NANOPOROUS CARBON MATERIALS: SIZE AND INTERFACE EFFECTS ON HYDROGEN MOBILITY AND RELEASE**
A. Nale, P.E. de Jongh
Inorganic Chemistry and Catalysis, Debye Institute for Nanomaterials Science, Utrecht University, TB Utrecht, The Netherlands
- 11:50-12:15** **SCANDIUM FUNCTIONALIZED CARBON AEROGELS FOR HYDROGEN STORAGE**
Payam Javadian¹, Marek Polanski², Thomas Plocinski², Dariusz Zasada², Jerzy Bystrzycki², Flemming Besenbacher³, Torben R. Jensen¹
¹Center for Energy Materials, Interdisciplinary Nanoscience Center (iNANO), and Department of Chemistry, Aarhus University, DK-8000 Aarhus, Denmark; ²Faculty of Advanced Technology and Chemistry, Military University of Technology, 2 Kaliskiego Str., 00-908 Warsaw, Poland; ³Interdisciplinary Nanoscience Center (iNANO) and Department of Physics and Astronomy, Aarhus University, DK-8000 Aarhus C, Denmark
- 12:15-12:30** **DEVELOPMENT OF NOVEL 3D TEM SAMPLE STRUCTURING METHODOLOGIES AND ADVANCED FIB AND TEM TOMOGRAPHY APPLICATIONS FOR NANOPOROUS MATERIALS**
Meltem Sezen
Sabanci University Nanotechnology Research and Application Center, Orhanli, Tuzla, 34956 Istanbul, Turkey
- 12:30** **Awards and closing ceremony**

POSTER SESSION 1EESRC SASA Gallery of Science and Technology (Đure Jakšića 2)

NOTE: Please bring the posters to the conference registration desk on 03.12.2012

E1. HYDROGEN SORPTION PROPERTIES OF $\text{MgH}_2/\text{NaBH}_4$ COMPOSITES

Sandra Kurko¹, Annalisa Aurora², Daniele Mirabile Gattia², Vittoria Contini²,
Amelia Montone², Željka Rašković-Lovre¹, Jasmina Grbović Novaković¹

¹Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11000 Belgrade, Serbia,

²ENEA, UTTMAT Research Centre of Casaccia, Via Anguillarese 301, 00123 Rome, Italy

E2. COMBINED SOLID STATE AND HIGH PRESSURE HYDROGEN STORAGE

Elisabeth Grube, Torben R. Jensen

Center for Energy Materials, Interdisciplinary Nanoscience Center, Department of Chemistry,
Aarhus University, Langelandsgade 140, DK-8000 Aarhus, Denmark

**E3. STRUCTURE AND HYDROGEN STORAGE PROPERTIES OF HIGH ENTROPY
TiVCRNIFE AND TiVCRNiFeZr ALLOYS SYNTHESIZED BY USING LASER ENGINEERED
NET SHAPING (LENS)**

I. Kunce, M. Polanski, J. Bystrzycki

Department of Advanced Materials and Technology, Military University of Technology, 2 Kaliskiego
Str., 00-908 Warsaw, Poland

**E4. CHARACTERIZATION OF MULTILAYERED HYDROGEN STORAGE MATERIALS BY
TRANSMISSION ELECTRON MICROSCOPY**

Tyché Perkisas¹, Sara Bals¹, Lennard Mooij², Bernard Dam²

¹Electron Microscopy for Materials Research (EMAT), University of Antwerp, Groenenborgerlaan
171, 2020 Antwerp, Belgium, ²Department of Chemical Engineering, Delft University of
Technology, Julianalaan 136, 2628 BL, Delft, The Netherlands

**E5. ELECTRONIC STRUCTURE AND FORMATION ENERGIES OF HfV_2H_x LAVES
PHASE**

Jana Radaković, Jelena Belošević-Čavor, Vasil Koteski, Katarina Ćirić

Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11000 Belgrade, Serbia

**E6. STRUCTURAL AND ELECTRONIC FACTORS GOVERNING HYDROGEN STORAGE
PROPERTIES OF TMNi COMPOUNDS (TM=Ti, Zr, Hf)**

Katarina Ćirić, Jana Radaković, Vasil Koteski, Dragica Stojić

Laboratory for Nuclear and Plasma Physics, Vinča Institute of Nuclear Sciences, University of
Belgrade, P.O. Box 522, 11000 Belgrade, Serbia

E7. HYDROGEN STORAGE, MICROSTRUCTURE AND MECHANICAL PROPERTIES OF $\text{Mg}_{65}\text{Ni}_{20}\text{Cu}_5\text{Y}_{10}$ METALLIC GLASS DEFORMED BY HIGH-PRESSURE TORSION

Á. Révész¹, Á. Kis-Tóth^{1,*}, L.K. Varga², E. Schafler³, T. Spassov⁴

¹Department of Materials Physics, Eötvös University, Budapest, H-1518, P.O.B. 32, Budapest, Hungary, ²Research Institute for Solid state Physics and Optics, Hungarian Academy of Sciences, H-1525 Budapest, P.O.B. 49, Hungary, ³Physics of Nanostructured Materials, Faculty of Physics, University of Vienna, A-1090 Vienna, Austria, ⁴Department of Chemistry, University of Sofia "St. Kl. Ohridski", 1 J. Bourchier str., 1164 Sofia, Bulgaria

E8. EFFECTS OF PARTICLE SIZE AND TYPE OF CONDUCTIVE ADDITIVE ON THE ELECTRODE PERFORMANCES OF GAS ATOMIZED AB_5 -TYPE HYDROGEN STORAGE ALLOY

L. Scherbakova, M. Spodaryk, Yu Solonin, A. Samelyuk

Institute for Problems of Materials Science (NASU), Krzhizhanovsky str., 3, Kyiv, Ukraine

E9. HYDROGEN DESORPTION PROPERTIES OF $\text{MgH}_2/\text{LiAlH}_4$ COMPOSITES

Igor Milanović¹, Sanja Milošević¹, Ljiljana Matović¹, Radojka Vujasin¹, Nikola Novaković¹, Riccardo Checchetto², Jasmina Grbović Novaković¹

¹Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11000 Belgrade, Serbia,

²Faculty of Physics, University of Trento, 38123 Povo, Italy

E10. EVALUATION OF THE HYDROGEN EMBRITTLEMENT OF METALLIC MEMBRANES BY THE DISK TEST METHOD

A. Metallari, I. Markja

Polytechnic University of Tirana, Albania

E11. ELECTRONIC STRUCTURE AND CHARGE TOPOLOGY STUDY OF ALKALI HYDRIDES

Bojana Paskaš Mamula, Mirjana Medić, Bojana Kuzmanović, Ivana Radisavljević, Nenad Ivanović, Nikola Novaković

Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001 Belgrade, Serbia

E12. REACTIVELY DEPOSITED THIN FILMS OF MAGNESIUM NICKEL HYDRIDE

Željka Rašković-Lovre¹, Trygve Mongstad², Simon Lindberg³, Stefano Deledda², Smagul Karazhanov²

¹Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 552, Belgrade, Serbia,

²Institute for Energy Technology, Box 40, NO-2027 Kjeller, Norway, ³Chalmers University of Technology, SE-412 96 Gothenburg, Sweden

E13. INVESTIGATION OF NUCLEATION PROCESS IN MgH_2 THIN FILMS

Željka Rašković-Lovre¹, Sandra Kurko¹, Nenad Ivanović¹, Jasmina Grbović Novaković¹, Jose Francisco Fernández², Jose Ramon Ares Fernández², Carlos Sánchez², Nikola Novaković¹

¹Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11000 Belgrade, Serbia,

²Departamento de Física de Materiales C-IV, Facultad de Ciencias, Universidad Autónoma de Madrid, Cantoblanco, 28049 Madrid, Spain

E14. PREPARATION AND CHARACTERIZATION OF MgH_2 -BASED COMPOSITE PELLETS

D. Mirabile Gattia¹, A. Montone¹, L. Pasquini²

¹ENEA – R.C. Casaccia – Technical Unit of Materials - Via Anguillarese, 301 00123 Rome, Italy,

²Department of Physics and CNISM, University of Bologna, V. Berti-Pichat 6/2, 40127 Bologna, Italy

E15. HYDROGEN-INDUCED DEFECTS IN THIN FILMS CHARACTERIZED BY VARIABLE ENERGY SLOW POSITRON ANNIHILATION SPECTROSCOPY

O. Melikhova¹, J. Cizek¹, W. Anwand², G. Brauer²

¹Charles University in Prague, Faculty of Mathematics and Physics, V Holesovickach 2, CZ-180 00 Praha 8, Czech Republic, ²Institut für Strahlenphysik, Helmholtz-Zentrum Dresden-Rossendorf, PO Box 510 119, D-01314 Dresden, Germany

E16. MAGNESIUM NANOPARTICLES AND NANODOTS: SYNTHESIS, STRUCTURE AND HYDROGEN SORPTION

Mattia Sacchi¹, Alan Molinari^{1,2}, Horst Hahn², Christiaan Boelsma³, Bernard Dam³, **Luca Pasquini¹**

¹Department of Physics and Astronomy, University of Bologna, v. Berti-Pichat 6/2, 40127 Bologna, Italy, ²Institute of Nanotechnology, Karlsruhe Institute of Technology, 76344 Eggenstein-Leopoldshafen, Germany, ³Department of Chemical Engineering, Faculty of Applied Sciences, Delft University of Technology, Julianalaan 136, 2628BL Delft, The Netherlands

E17. VANADIUM OXIDE AS HYDROGEN TECHNOLOGY MATERIAL

Dragica M. Minić¹, Dejan G. Minić², **Vladimir A. Blagojević^{1,3}**

¹University of Belgrade, Faculty of Physical Chemistry, Belgrade, Serbia, ²Kontrola LLC, Austin, TX, USA, ³Vida Holdings Corp. Ltd, Toronto, Ontario, Canada

E18. POSSIBLE PATHS OF HYDROGEN DIFFUSION IN TiO_2 – ROLE OF THE SURFACE

Radojka Vujasin, Igor Milanović, Sandra Kurko, Bojana Paskaš Mamula, Nikola Novaković

Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001 Belgrade, Serbia

E19. DESORPTION PROPERTIES OF MgH_2 DESTABILIZED WITH NaNH_2 CATALYST

Sanja Milošević¹, Igor Milanović¹, Bojana Paskaš Mamula¹, Anđelka Đukić¹, Ljiljana Matović¹, Jasmina Grbović Novaković¹, Luca Pasquini²

¹Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11000 Belgrade, Serbia,

²Dipartimento di Fisica, Università di Bologna and CNISM, v.le Berti-Pichat 6/2, I-40127 Bologna, Italy

E20. MICROSTRUCTURE OF ULTRAFINE-GRAINED WE43+12wt% Zn MAGNESIUM ALLOY AFTER EQUAL CHANNEL ANGULAR PRESSING

Jan Duchoň, Aleš Jäger, Viera Gärtnerová

Institute of Physics ASCR, v. v. i., Na Slovance 1999/2, 182 21 Praha 8, Czech Republic

E21. SEVERE PLASTIC DEFORMATION OF Mg-Mn ALLOY AT ROOM TEMPERATURE

Erik Švec, Jan Duchoň, Viera Gärtnerová, Aleš Jäger

Institute of Physics of the AS CR, v.v.i., Na Slovance 2, 182 21 Prague 8, Czech Republic

E22. ENHANCING THE HYDROGEN GRAPHENE INTERACTION BY METAL DECORATION

Andreas Bliersbach

Empa, Swiss Federal Laboratories for Materials Science and Technology, Hydrogen and Energy, Überlandstrasse 129, CH-8600 Dübendorf, Switzerland

E23. THE POTENTIAL APPLICATION OF HYDROGEN IN DEVELOPING COUNTRIES

Rabah Arkam

Euro T-Systems, 05, rue Chafai Ahmed, Tizi Ouzou Algeria

*Joint event of the 11th Young Researchers' Conference: Materials Science and Engineering and
the 1st European Early Stage Researchers' Conference on Hydrogen Storage*

THE BOOK OF ABSTRACTS

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P 1

MECANOCHEMISTRY UNDER HYDROGEN GAS: POTENTIALITIES FOR THE SYNTHESIS OF EFFICIENT HYDROGEN STORAGE MATERIALS

Fermin Cuevas, Junxian Zhang, Michel Latroche

Institut de Chimie et des Matériaux Paris-Est, CNRS-UPEC, UMR7182, 2-8 rue
Henri Dunant, 94320, Thiais Cedex, France

Mechanochemistry of hydrides by mechanical milling of metal powders under hydrogen gas is a powerful technique for the synthesis of metallic and complex hydrides (J. Huot *et al.* Prog. Mater. Sci.,58 (2013) 30). We review the historical advances on this technique and the current state-of-the-art. The synthesis of binary metal hydrides, solid solution BCC hydrides, hydrides from non-miscible metals, Mg- and alanate-complex hydrides is surveyed. Some particular phenomena such as ball-milling induced amorphisation, formation of metastable phases, multi-step reactions and incomplete hydride formation are reported. The obtained hydrides are typically nanocrystalline materials leading to fast kinetic reactions for hydrogen storage applications.

E-mail of corresponding author: cuevas@icmpe.cnrs.fr
Phone +33 1 4978 1225

P 2

KEY CONCEPTS IN THE DESIGN OF FUNCTIONAL NANOPARTICLES

Vuk Uskoković

Therapeutic Micro and Nanotechnology Laboratory, Department of
Bioengineering and Therapeutic Sciences, University of California San Francisco,
1700 4th Street, San Francisco, CA 94158-2330, USA

Nanomaterials are defined as solid physical structures whose critical sizes do not exceed 100 nm. In this presentation, the origins of the field of nanoscience will be discussed, alongside some of the essential concepts applicable in the design of functional nanoparticles. Results obtained over years during the lecturer's research in various materials science fields - ranging from the synthesis of magnetic nanoparticles in reverse micelles, to nanostructured manganites for biomedical applications, to the chemistry of precipitation of cholesterol, to biomimicry of nanohierarchically organized hard tissues, to controlled delivery of proteins and small therapeutic molecules using silicon nanowires and calcium phosphate nanoparticles - will be mentioned and placed in a broader context defined by the role, the significance, the prospects and the tools of the field of nanoscience and nanotechnologies.

E-mail of corresponding author: vuk21@yahoo.com

P 3

WHAT WE LEARN FROM THE HISTORY TO MASTER THE FUTURE: THE ROLE OF HYDROGEN

Andreas Züttel

EMPA Materials Science & Technology, Dübendorf, Switzerland

The worldwide energy demand increases just as rapidly as the average temperature of the atmosphere. The reserves of fossil fuels worldwide are limited and the combustion of the carbon fuels leads to a severe increase of the CO₂ concentration in the atmosphere. The latter is responsible for the climate change. The future of the industrialized world, i.e. the economy as well as the society, is determined by the ability to change from fossil fuels as energy carriers to renewable energy. The main difference between the fossil period and the future is that renewable energy occurs as energy fluxes and, therefore, an energy carrier has to be synthesized.

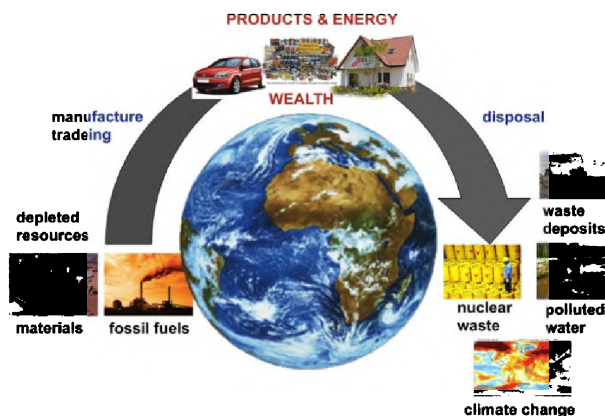


Fig. 1 Schematic representation of the economic system of the 20th century.

Hydrogen as an energy carrier can be produced and used in a purely technical and closed cycle. The storage of hydrogen in metals and complex hydrides as stable compounds offers a great volumetric storage density, however the gravimetric storage density is limited to less than 20mass% in the materials. In order to replace fossil fuels without scarification on energy density synthetic fuels based on hydrogen e.g. NH₃ or C₈H₁₈ have to be developed. The latter also represents a effective CO₂ sink for the atmosphere if more is produced than used.

E-mail of corresponding author: Andreas.Zuettel@empa.ch

TM 1

EFFECTS OF FULLERENOL NANO PARTICLES C₆₀(OH)₂₄ ON FILAMENTOUS FUNGUS *Aspergillus niger* Tiegh

Nikola Unković¹, Miloš Stupar¹, Milica Ljaljević Grbić¹, Jelena
Vukojević¹, Marina Seke², Danica Jović², Aleksandar Djordjević²

¹University of Belgrade, Faculty of Biology, Institute of Botany and Botanical
Garden "Jevremovac", Takovska 43, 11000 Belgrade,

²Faculty of Science, Department of Chemistry, Biochemistry and Environment, Trg
Dositeja Obradovića 3, 21000 Novi Sad, University of Novi Sad

The aim of this research is to investigate the effect of polyhydroxylated fullerene derivative, fullerenol C₆₀(OH)₂₄ nano particles (FNP), on mycelial growth and spore germination of model organism, filamentous fungus, *Aspergillus niger*. The effect of FNP was tested using microdilution method and spore germination assay. Here, we reported inhibitory effect of FNP on spore germination relative to the FNP concentration, as well as influence on morphophysiological characteristics of conidiogenous apparatus of *A. niger*. The most significant result is the depigmentation of the melanized conidial apparatus, since the melanin is the virulence factor in human pathogenic fungi.

E-mail of corresponding author: jmilica@bio.bg.ac.rs
Phone +381 64 460 7076

TM 2

PHASE AND MICROSTRUCTURAL EVOLUTION DURING SINTERING OF Zr-DOPED HYDROXYAPATITE

**Miodrag J. Lukić¹, Smilja Marković¹, Srečo Davor Škapin², Dragan
Uskoković¹**

¹Institute of Technical Sciences of SASA, Belgrade, Serbia

²Institute Jozef Stefan, Ljubljana, Slovenia

Doping of hydroxyapatite crystal structure with various ions is generally performed to improve biological, electrical, optical and mechanical properties of initial system. In this study, hydroxyapatite nanopowders doped with different amounts of zirconium (0, 1, 5, 10 at. %) have been sintered at different temperatures. Phase and microstructural evolution were examined by X-ray diffraction studies and field emission scanning electron microscopy.

E-mail of corresponding author: miodrag.lukic@itn.sanu.ac.rs

Phone +381 11 2651067

TM 3

THE BIOACTIVITY INVESTIGATION OF ELECTRODEPOSITED SILVER/HYDROXYAPATITE/LIGNIN COATINGS IN SIMULATED BODY FLUID

**Sanja Eraković¹, Ana Janković¹, Ivana Matić², Zorica Juranić², Maja
Vukašinović-Sekulić¹, Tatjana Stevanović³, Vesna Mišković-Stanković¹**

¹Faculty of Technology and Metallurgy, University of Belgrade, Belgrade, Serbia;

²Institute of Oncology and Radiology of Serbia, University of Belgrade, Belgrade,
Serbia;

³Département des sciences du bois et de la forêt, Université Laval, Québec,
Canada

The newest research on hydroxyapatite/lignin (HAP/Lig) scaffolds that assist bone repair refer us to employ organosolv lignin as the most suitable component for HAP composite that would provide better interconnected porous structure. The development of HAP composites containing Ag is necessary to improve antimicrobial activity of HAP composites. The Ag/HAP/Lig coatings were obtained by cathaphoretic deposition on Ti and characterized by various techniques in order to investigate the surface composition, structure and electrochemical characteristics as a function of soaking time in simulated body fluid (SBF). The coatings showed antibacterial properties, low toxicity, high bioactivity and corrosion stability in SBF.

E-mail of corresponding author: serakovic@tmf.bg.ac.rs

Phone +381 11 3303 686

TM 4

PRODUCTION AND ANTIMICROBIAL ACTIVITY OF FIBERS BASED ON ALGINATE WITH INCORPORATED SILVER NANOPARTICLES

**Srđan Vidović, Milan Stevanović, Maja Vukašinović-Sekulić, Bojana
Obradović**

Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4,
Belgrade, Serbia

The aim of our study was to investigate possibilities for production of hydrogels in the form of fibers based on alginate and poly(vinyl alcohol) (PVA) with incorporated silver nanoparticles (AgNPs) for potential use as antimicrobial wound dressings. Wet-spinning process was developed for production of nanocomposite Ag/alginate microfibers ($\sim 300 \mu\text{m}$ in diameter) as well as Ag/alginate/PVA fibers ($\sim 1\text{mm}$ in diameter) with incorporated electrochemically synthesized AgNPs at the concentrations of 1 and 2.6 mM, respectively. Antimicrobial activity of the obtained fibers in wet and dried forms, was investigated against *E. coli* and *St. aureus* revealing bacteriostatic and bactericidal effects

E-mail of corresponding author: svidovic@tmf.bg.ac.rs

Phone +381 64 339 84 09

TM 5

EVALUATION OF NOVEL Ag/ALGINATE MICROBEADS FOR POTENTIAL BIOMEDICAL APPLICATIONS

Jasmina Stojkovska, Željka Jovanović, Danijela Kostić, Maja
Vukašinović-Sekulić, Vesna Mišković-Stanković, Bojana Obradović

Faculty of Technology and Metallurgy, University of Belgrade
Karnegijeva 4, 11120 Belgrade, Serbia

In this study, we have produced and characterized alginate microbeads with incorporated silver nanoparticles aimed for biomedical applications, specifically as potential antimicrobial wound dressings and soft tissue implants. Therefore, Ag/alginate microbeads were studied regarding the release of silver nanoparticles and/or ions, antibacterial activity against *Staphylococcus aureus* and cytotoxicity in monolayers of bovine calf chondrocytes. Furthermore, a short-term cultivation study of microbeads with immobilized chondrocytes mixed with nanocomposite microbeads under biomimetic bioreactor conditions was performed mimicking potential applications as engineered tissue implants. Overall results have shown potentials of Ag/alginate microbeads for biomedical applications, possibly reducing the need for antibiotics.

E-mail of corresponding author: jstojkovska@tmf.bg.ac.rs
Phone +381 63306090

TM 6

COPPER – ALGINATE MICROBEADS: BIOACTIVE, ANTIMICROBIAL BIOMATERIALS FOR POTENTIAL BIOMEDICAL APPLICATIONS

Ivana Madžovska, Maja Vukašinović – Sekulić, Bojana Obradović

Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4,
11000 Belgrade, Serbia

Bioactivity and antimicrobial properties of alginate hydrogels can be enhanced by using Cu^{2+} ions as cross-linking agents. In this study, Cu-alginate microbeads were produced by electrostatic extrusion of Na-alginate solution using CuSO_4 as a gelling agent at different concentrations. The increase in CuSO_4 concentration was shown to increase the cation content and mechanical properties, while decreasing the size of the obtained microbeads. Cu-alginate microbeads also exhibited antibacterial activity against *Staphylococcus aureus* and *Escherichia coli*. The obtained results imply that the presented production procedure could be optimized so to control the Cu^{2+} ion release aiming to induce desired biological responses.

E-mail of corresponding author: madzovska@tmf.bg.ac.rs
Phone +381693362550

TM 7

HEMOLYTIC ACTIVITY OF BIOACTIVE NANOCOMPOSITES

Z. Ajduković¹, N. Ignjatović², **N. Petrović¹**, S. Najman³, J. Rajković⁴, D.
Kenić Marinković¹, V. Krstić³, D. Uskoković²

¹University of Niš, Faculty of Medicine, Department of Prosthodontics, Clinic of
Stomatology, Niš, Serbia

²Institute of Technical Sciences of SASA, Belgrade, Serbia

³University of Niš, Faculty of Medicine, Institute of Biomedical Research, Niš,
Serbia

⁴University of Niš, Department of Biology and Ecology, Faculty of Science, Niš,
Serbia

Huge range of tested biomaterials in recent decades has emerged as an ideal scaffold for cell growth, but few have demonstrated clinical efficacy. Among them, synthetic hydroxyapatite (HAp, $\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$) is the most promising because of its biocompatibility, bioactivity, and osteoconductivity. Biocompatibility represents the primary concern for any material to be used as a substitute for natural tissue. Hydroxyapatite particles interact with numerous cellular systems in vivo, and some of these interactions may lead to cell damage and to stimulate platelet activation, coagulation and thrombus formation. The aim of this work was to examine the hemocompatibility of nano-calcium hydroxyapatite substituted with 5% and 12% cobalt (Ca/Co-HAp) and hydroxyapatite/poly-lactid-coglicolid (HAp/PLGA) in relation to pure HAp by testing their hemolytic activities. The results show the discrepancy in hemolytic activity of implanted materials. The degree of crystallinity of samples had a more dominant influence on hemolysis than the percentage of substituted cobalt. Hemolysis ratios of the nano-calcium hydroxyapatite substituted with cobalt samples were below 3%, indicating good blood compatibility and that they are promising for medical application.

E-mail of corresponding author: knele987@gmail.com

Phone +381 64 3223231

TM 8

CHRONOAMPEROMETRIC DETERMINATION OF HYDROGEN- PEROXIDE BY MULTIWALL CARBON NANOTUBES MODIFIED CARBON BASED ELECTRODES

**Jasmina Zbiljić¹, Olga Vajdle¹, Valéria Guzsány¹, Ákos Kukovecz²,
Zoltán Kónya², Božo Dalmacija¹, Kurt Kalcher³**

¹University of Novi Sad, Faculty of Sciences, Department of Chemistry,
Biochemistry and Environmental Protection, Trg D. Obradovića 3, 21000 Novi Sad,
R. Serbia

²University of Szeged, Department of Applied and Environmental Chemistry,
Rerrich 1, 6720 Szeged, Hungary

³Karl-Franzens University, Universitätsplatz 3, 8010 Graz, Austria

Nowadays, multiwall carbon nanotubes (MWCNT) are very popular material for designing of amperometric sensors and biosensors. The aim of this work was to investigate the applicability of multiwall carbon nanotubes modified carbon paste electrodes (MWCNT-CPE) and screen printed carbon electrodes (MWCNT-SPCE) for the chronoamperometric monitoring of hydrogen-peroxide concentration in model systems. As first, optimization experiments were performed to prepare carbon paste electrodes (based on graphite powder and paraffin oil) bulk doped with unmodified (MWCNT-CPE) and chemically oxidized multi-walled carbon nanotubes (C-MWCNT-CPE). The influence of the dopant nature (unmodified and oxidized forms) and content (5% and 10% w/w) were investigated. Based on the results, 5% w/w of unmodified MWCNT in MWCNT-CPE showed better analytical performances in terms of hydrogen-peroxide determination than the unmodified CPE, and the C-MWCNT-CPE, because of lower noise and higher sensitivity and reproducibility of the developed analytical method. In our preliminary work, MWCNT-SPCE was also tested for chronoamperometric determination of hydrogen-peroxide.

E-mail of corresponding author: jasmina.zbiljic@dh.uns.ac.rs
Phone +381642805729

TM 9

ELECTROCHEMICAL SYNTHESIS AND CHARACTERIZATION OF SILVER NANOPARTICLES IN POLY(VINYL ALCOHOL) SOLUTION

Rade Surudžić, Željka Jovanović, Vesna Mišković-Stanković

Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4,
Belgrade

In this work, the optimal conditions for the electrochemical synthesis of silver nanoparticles in the solution of poly(vinyl alcohol), PVA, were investigated, by varying the values of the applied current density, ranging from 5 to 50 mA cm⁻². The galvanostatic synthesis is performed using aqueous solutions of 10 wt. % PVA, containing KNO₃ and AgNO₃. The presence of silver nanoparticles was confirmed by UV-vis spectroscopy. TEM analysis shows that the obtained silver nanoparticles were spherically shaped. FTIR spectroscopy indicates the interactions between silver nanoparticles and PVA molecules.

E-mail of corresponding author: rade.surudzic@gmail.com

Phone +381 11 3303 815

TM 10

VOLTAMMETRIC DETERMINATION OF DOXORUBICIN BY RENEWABLE SILVER-AMALGAM FILM ELECTRODE

**Bojan Tasić, Jasmina Zbiljić, Olga Vajdle, Danica Jović, Valéria
Guzsvány, Aleksandar Đorđević**

University of Novi Sad, Faculty of Sciences, Department of Chemistry,
Biochemistry and Environmental Protection, Trg D. Obradovića 3, 21000 Novi Sad,
Serbia

Voltammetric characterization and determination of Doxorubicin (DOX) was performed by direct cathodic square-wave voltammetry (SWV) and by a highly-sensitive adsorptive stripping square-wave voltammetry (SWAdSV) using renewable silver-amalgam film electrode (Hg(Ag)FE) in aqueous Britton-Robinson buffer solution as supporting electrolyte. In the investigated pH range between 2.0 and 8.0, the reduction peak of DOX was obtained in the interval of approximately -0.3 to -0.8 V. At pH 6.0 Hg(Ag)FE showed best analytical characteristics in respect of the reduction peak shape and intensity of DOX. The developed SWAdSV method was used to determine of target analyte in model solutions in the concentration range of 4,99-59,64 ng ml⁻¹.

E-mail of corresponding author: bjntasic@yahoo.com
Phone +381691709835

TM 11

CORROSION AND ION RELEASE BEHAVIOR OF NITRIC ACID PASSIVATED STAINLESS STEEL

Dušan Lj. Petković, Goran M. Radenković

Faculty of Mechanical Engineering, University of Niš, A. Medvedeva 14, Niš

The aim of this work was to identify the effects of stainless steel surface passivation on corrosion and release ion behavior. Cylindrical samples were wet ground, polished, rinsed with water and washed with ethanol in the ultrasonic cleaner. The samples were divided into 3 treatment groups: non-treated, treated with the nitric acid (65 %) 15 min and treated with the acid 30 minutes. Corrosion resistance of samples was determined by performing the potentiodynamic tests in Hank's solution at 37°C. Amount of released Ni, Cr and Fe ions in mg/dm³ was appointed right after treatment and cleaning. Obtained results indicate the passivation of the stainless steel could improve its corrosion resistance and decrease amount of metal ion released from the surfaces.

E-mail of corresponding author: dulep@masfak.ni.ac.rs

Phone +381 18 500 624

TM 12

**VOLTAMMETRIC DETERMINATION OF FENOXANIL FUNGICIDE
BY RENEWABLE SILVER-AMALGAM FILM ELECTRODE**

**Olga Vajdle¹, Mariola Brycht², Jasmina Zbiljić¹, Valéria Guzsány¹,
Sławomira Skrzypek²**

¹University of Novi Sad, Faculty of Sciences, Department of Chemistry,
Biochemistry and Environmental Protection, Trg D. Obradovića 3, 21000 Novi Sad,
Serbia

²University of Łódź, Department of Instrumental Analysis, Pomorska 163,
90-236 Łódź, Poland

Environmental friendly direct cathodic square-wave voltammetric (SWV) and adsorptive stripping square-wave voltammetric (SWAdSV) methods using a renewable silver-amalgam film electrode (Hg(Ag)FE) were developed for the characterization and determination of Fenoxanil (Fen, *N*-(1-cyano-1,2-dimethylpropyl)-2-(2,4-dichlorophenoxy)propanamide) fungicide in aqueous Britton-Robinson buffer solution as supporting electrolyte. In the investigated pH range between 2.0 and 9.0, two reduction peaks were observed for Fen, one was at -0.4 V and another one at -1.5 V. For determination of Fen the second peak was selected. The well shaped and most intensive reduction signal was obtained at pH 6.0, so this pH was chosen for analytical purposes. Under the optimal conditions, there was a linear relationship between the reduction peak height and concentration of Fen over the concentration range of 2.72-15.46 $\mu\text{g mL}^{-1}$ using SWV method and 0.71-2.29 $\mu\text{g mL}^{-1}$ using SWAdSV one. Finally, the developed SWV and SWAdSV methods were tested to determine Fen in selected real samples.

E-mail of corresponding author: olgavajdle@yahoo.com
Phone +381 64 299 47 27

TM 13

**CHARACTERIZATION OF ORGANIC INHIBITOR FOR CORROSION OF
MILD STEEL IN 3 wt. % NaCl SOLUTION SATURATED WITH CO₂**

**Ivana Jevremović¹, Marc Singer², Srđan Nešić², Vesna Mišković-
Stanković¹**

¹Faculty of Technology and Metallurgy, University of Belgrade, Belgrade, Serbia;

²Institute for Corrosion and Multiphase Technology, Ohio University, Athens, OH,
USA

In this study talloil diethylenetriamine imidazoline (TOFA/DETA imidazoline) was evaluated as corrosion inhibitor for mild steel in 3 wt. % aqueous NaCl solution at 20°C and 70°C using the following techniques: the electrochemical impedance spectroscopy, linear polarization resistance and weight loss measurements. The adsorption process was followed by quartz crystal microbalance and the morphology of the steel surface was investigated by scanning electron microscopy and atomic force microscopy. All measurements applied indicated that TOFA/DETA imidazoline significantly decreased the corrosion rate when 70 ppm (vol.) of corrosion inhibitor was added due to its ability to adsorb on the metal surface.

E-mail of corresponding author: ijevremovic@tmf.bg.ac.rs

Phone +381 11 3303715

TM 14

**THE INFLUENCE OF ADDITIVES ON THE MORPHOLOGICAL AND
CORROSION PROPERTIES OF ELECTRODEPOSITED Zn-Mn ALLOYS**

M. Bučko¹, U. Lačnjevac², J. Rogan¹, B. Jokić¹, J. B. Bajat¹

¹Faculty of Technology and Metallurgy, University of Belgrade, P.O. Box 3503,
11120 Belgrade, Serbia,

²Institute for Multidisciplinary Research, University of Belgrade, 11030 Belgrade,
P.O. Box 33, Serbia

The electrodeposition of Zn–Mn alloys in chloride electrolytes, as sacrificial coatings on steel, was investigated. It was found that depending on the plating parameters up to 36 at.% Mn could be obtained in Zn-Mn alloy coatings, however, at the cost of very low current efficiencies and with incorporation of a great amount of oxides/hydroxides. In order to prevent formation of oxides the addition of inhibitor for hydrogen evolution and surfactant was examined. Their effects upon the composition, morphology, microstructure and corrosion stability of the deposit were investigated using electrochemical methods, scanning electron microscopy and energy dispersive X-ray spectrometry.

E-mail of corresponding author: mihaelbucko@yahoo.com

Phone +381 064 186 4896

TM 15

IDENTIFICATION OF PROMISING CHEMICAL SYSTEMS FOR THE SYNTHESIS OF NEW MATERIALS STRUCTURE TYPES: AN *AB INITIO* MINIMIZATION DATA MINING APPROACH

D. Zagorac, J.C. Schön, M. Jansen

Max Planck Institute for Solid State Research, Stuttgart, Germany

In this research we performed data exploring for binary compounds with elements from groups V, IV - VI, and III - VII, with the goal to identify chemical systems where the recently proposed "5-5" structure type might be experimentally accessible. Among others, TlF, SnO, SnS, SnSe, GeS, GeSe, PbO, PbS, ZnO and ZnS, were chosen for the study. In the next step, for each of these systems a local optimization on *ab initio* level with the LDA functional was performed for the 5-5 structure type plus other experimentally observed and theoretically proposed structure types, for comparison. Afterwards, the results were combined with earlier theoretical work involving the 5-5 structure in the earth alkaline metal oxides and the alkali metal halides. As a result, we suggest the GeSe and the ZnO systems as the most suitable ones for synthesizing the 5-5 structure type.

E-mail of corresponding author: d.zagorac@fkf.mpg.de

Phone +49 (0)711 689 1367

TM 16

KINETICS OF CRYSTALLIZATION OF $\text{Fe}_{89.8}\text{Ni}_{1.5}\text{Si}_{5.2}\text{B}_3\text{C}_{0.5}$ AMORPHOUS ALLOY

**Milica M. Vasić¹, Vladimir A. Blagojević¹, Dušan M. Minić²,
Dragica M. Minić¹**

¹Faculty of Physical Chemistry, University of Belgrade, Serbia

²Military Technical Institute, Belgrade, Serbia

Kinetics of structural transformation of $\text{Fe}_{89.8}\text{Ni}_{1.5}\text{Si}_{5.2}\text{B}_3\text{C}_{0.5}$ amorphous alloy was examined using DSC measurements at different heating rates. Distinct peaks of glass transition and crystallization were observed. In order to understand crystallization process better, complex crystallization peak was deconvoluted into two steps, which correspond to the formation of $\alpha\text{-Fe}(\text{Si})$ and Fe_2B crystalline phases. Kinetic triplets for both crystallization steps were determined, showing relatively high values of activation energies as a consequence of simultaneous movement of large number of atoms. Determination of the correct form of the conversion function revealed that Šesták-Berggren model best describes both crystallization steps.

E-mail of corresponding author: vladab64@gmail.com, dminic@ffh.bg.ac.rs

TM 17

CHARACTERIZATION OF PULSED LASER DEPOSITED CdS ON TiO₂ NANOTUBES ARRAYS

**A. Bjelajac¹, V.Djokić¹, R. Petrović¹, G. Socol², I. Mihailescu², I. Florea³,
O. Ersen³, Dj. Janačković¹**

¹Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4,
11000 Belgrade, Serbia

²National Institute for Lasers, Plasma, and Radiation Physics, Bucharest-Magurele,
Romania

³Institut de Physique et Chimie des Materiaux de Strasbourg, CNRS - Universite de
Strasbourg (UDS) UMR 7504, 23, rue du Loess, BP 43, 67037 Strasbourg cedex 02,
France

Using field emission scanning electron microscopy and high resolution transmission electron microscopy with electron energy loss spectroscopy and energy filtered mode of transmission electron microscopy, the characterization of acquired samples was performed and the chemical mapping of the elements was obtained. The results showed that the samples consisted of highly oriented titania nanotube arrays with cadmium sulfide deposited inside and between nanotubes, which had a ~100 nm inner diameter. The deposition was performed via pulse laser deposition technique. The deposit thicknesses were ~ 5 nm and ~ 10 nm, increasing with the number of laser pulses, 250 and 500, respectively.

E-mail of corresponding author: abjelajac@tmf.bg.ac.rs

Phone +381-11-3303741

TM 18

INFLUENCE OF PARTICLE SIZE AND MORPHOLOGY OF ZnO POWDERS ON THEIR OPTICAL PROPERTIES

**A. Stanković¹, Z. Stojanović¹, Lj. Veselinović¹, N. Abazović²,
S.D. Škapin³, S. Marković¹, D. Uskoković¹**

¹Institute of Technical Sciences of SASA, Belgrade, Serbia

²Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

³Jožef Štefan Institute, Ljubljana, Slovenia

In this work we present details of different synthesis procedures for preparation of ZnO powders and a possibility to create various particles morphology and average size; from micrometer rods to nanometer spherical particles. We studied two synthesis methods, low temperature hydrothermal processing (HT) and ultrasound processing (US). The morphology and average particle size of the ZnO particles were modified by adjusting the mole ratio of $[\text{Zn}^{2+}]/[\text{OH}^-]$ in the reactants solution. Agglomeration of the particles was prevented using a polyvinyl pyrrolidone (PVP) as a capping agent. The synthesized powders were characterized by X-ray powder diffraction (XRD), scanning electron microscopy (FE SEM) and UV-Vis diffusion reflection spectroscopy (UV-Vis DRS). Based on the results of the scanning electron microscopy we proposed a possible growth mechanism of the ZnO particles depending on the mole ratio of $[\text{Zn}^{2+}]/[\text{OH}^-]$. Also, we considered the influence of particle size and morphology on the optical properties of the prepared powders. The observed results showed that the size and morphology of ZnO particles greatly affect the absorption of visible light (%) and the shift of the energy gap (eV).

E-mail of corresponding author: ana.stankovic@itn.sanu.ac.rs

Phone +381-11-2636-994

TM 19

COMPACTION PRESSURE'S INFLUENCE ON DENSITY AND ELECTRICAL PROPERTIES OF SINTERED CORDIERITE-BASED CERAMICS

N. Obradović^{1*}, N. Đorđević², A. Peleš¹, M. Mitrić³, V. Pavlović¹

¹Institute of Technical Sciences of SASA, Knez Mihajlova 35/IV, 11000 Belgrade, Serbia

²Institute for Technology of Nuclear and Other Mineral Raw Materials, Bulevar Franse d'Eperea 86, 11000 Belgrade, Serbia

³Vinča institute of Nuclear Sciences, University of Belgrade, Mike Alasa 12-14, 11000 Belgrade, Serbia

Mechanical activation is widely used for sample preparation before the sintering process. However, the influence of pressure, which is used for compaction, has not been completely investigated. Cordierite, $2\text{MgO} \cdot 2\text{Al}_2\text{O}_3 \cdot 5\text{SiO}_2$, is a very attractive high-temperature ceramic material, due to its outstanding electrical characteristics, such as the low temperature expansion coefficient, low dielectric constant and good mechanical properties. Based on our previous investigation, the mechanical activation of the starting mixtures with 5.00 mass% TiO_2 was performed in a high energy ball mill during 10 minutes. Compaction pressure varied from 0.5 to 6 t/cm². Differential thermal analysis (DTA) was used in order to determine characteristic temperatures within the system during heating. Sintering process was performed at 1350 °C for 4 h in air atmosphere. The phase composition of activated and sintered samples was analyzed by the X-ray diffraction method. Scanning electron microscopy was performed to analyze both compacted and sintered sample microstructure. This paper deals with the influence of compaction pressure on the densities of green bodies as well as on the sintered samples, along with electrical properties

E-mail of corresponding author: adriana.peles@itn.sanu.ac.rs

Phone +381-11-2027-203

TM 20

DEPOSITION OF THE DLC STRUCTURES IN THE LOW-PRESSURE OXY- ACETYLENE FLAT FLAME

**Bojana Radojković¹, Marija Krmar¹, Miroljub Vilotijević²,
Bojan Gligorijević¹, Ana Alil¹**

¹Institute Goša, Milana Rakića 35, Belgrade, Serbia

²University of Belgrade, Vinča Institute of Nuclear Sciences, 11001 Belgrade,
Serbia

The morphology and phase composition of the diamond-like carbon (DLC) structures as a function of the deposition process parameters are investigated. The DLC was deposited in the low-pressure oxy-acetylene flat flame onto molybdenum substrates. It was done with the designed flame burner by spreading the flame over the area of 40 mm in diameter with the overall acetylene and oxygen flow up to 4 l/min and by changing the substrate distance and the O₂/C₂H₂ ratio. The morphology and the phase composition of the DLC coatings significantly depend on the deposition process parameters, especially on the O₂/C₂H₂ ratio.

E-mail of corresponding author: bojana.radojkovic@institutgosa.rs

TM 21

CHEMICAL STABILITY OF $\text{Fe}_x(\text{Sb}_2\text{S}_3)_{0.75}(\text{SbI}_3)_{0.25}{}_{1-x}$ GLASSES

N. Ćelić, M. D. Vučkovac, I. O. Gut, M. Avramov

University of Novi Sad, Faculty of Sciences, Department of Physics, Trg D.
Obradovića 4, Novi Sad, Serbia

This paper presents the investigation of chemical stability of $\text{Fe}_x(\text{Sb}_2\text{S}_3)_{0.75}(\text{SbI}_3)_{0.25}{}_{1-x}$ chalcogenide glasses. The content of Fe vary from 0.01 to 5 at. %. The dissolution rates of investigated samples in potassium hydroxide solution of different concentrations at room temperature were calculated by measuring the mass loss. The samples showed relatively low dissolution rate, eg. large chemical stability but still lower than for the other chalcogenide glass systems. The dissolution rates increased with the increase of Fe concentration in the glass matrix, so the Fe atoms could be considered as catalysts for chemical dissolution.

E-mail of corresponding author: maja.vuckovac@df.uns.ac.rs
Phone +381642114422

TM 22

WELDING PARAMETERS PREDICTION BY THE ANALYTICAL METHODS AND USE OF NUMERICAL SIMULATION SOFTWARE FOR GS-36Mn5 STEEL

S. Budimir, B. Jegdić, M. Prokolab, M. Prvulović, Z. Milutinović

Institut Goša, Milana Rakića 35, Beograd, Srbija

The use of cast steel GS 36Mn5 in the production of welded construction requires the use of special measures during welding process or surfacing, such as preheating, heat treatment after welding, the use of buffer layers and etc. Analytical assessment of the material weldability was performed with different methods and result comparison with the results of software application "Weld simulator" for MMA process. The results show good agreement in the assessment of welding parameters by applied procedures, and suggest what measures must be taken to obtain high quality of welded joints.

E-mail of corresponding author: mirjanaprulovic@gmail.com

TM 23

CALCULATION OF TEMPERATURE DISTRIBUTION OF THE POWDER PARTICLES IN THE SPRAY DEPOSITION PROCES

**Milorad Kočić, Mirjana Prvulović, Jasmina Obradović, Marko Ristić,
Milan Prokolab**

Institute Goša, Milana Rakica 35, 11000 Belgrade, Serbia

A numerical simulation of spray deposition process for aluminum alloy onto rotating substrate is developed. For the modeling of ones was applied heat-flow analysis. The mathematical model was solved numerically. Temperature distribution for different particle size depending on the distance between the nozzle and the substrate is shown. Obtained results indicate the presence of smaller temperature gradients for the larger diameter particles.

E-mail of corresponding author: mirjanaprulovic@gmail.com

TM 24

APPLICATION OF FINITE ELEMENTS METHOD IN ANALYSIS OF DYNAMIC BEHAVIOR OF COMPOSITE LAMINATES

Dragan Čukanović¹, Aleksandar Radaković², Miroslav Živković³

¹Faculty of Technical Sciences, Kosovska Mitrovica,

²State University in Novi Pazar, ³Faculty of Engineering, Kragujevac, Serbia

This paper presents the theoretical bases of propagation of elastic waves in a composite laminate. According to theoretical assumptions the paper analyses the stress and displacement fields of a laminate which is modeled by orthotropic multilayer shell elements by using PAK software package. Finally the paper shows the comparative result of different orientations monolayers of laminates and different types of dynamic loading by using the implicit Newmark method and the explicit central difference method for numerical integration of dynamic equations of motion.

E-mail of corresponding author: dragan.cukanovic@pr.ac.rs

Phone +381 64 3939 417

TM 25

PROPAGATION OF BULK WAVES IN A TRANSVERSAL ISOTROPIC MEDIUM

**Aleksandar Radaković¹, Dragan Čukanović², Dragan Milosavljević³
Gordana Bogdanović³, Ljiljana Veljović³**

¹State University in Novi Pazar, ²Faculty of Technical Sciences, Kosovska Mitrovica,
³Faculty of Engineering, Kragujevac, Serbia

The present paper treats the fiber reinforced medium as a homogeneous transversally isotropic medium. In order to follow influence of fiber direction on mechanical properties, constitutive relations are modeled as a function of fiber direction in each point of continuous media.

Testing the mechanical behavior of an anisotropic medium with propagation of bulk waves is done using Riemann-Christoffel's equation that defines three homogeneous linear equations which determine displacements amplitudes. Thus the propagation of bulk waves is analyzed as a system of homogeneous linear equations and represents a condition of propagation. Characteristics of these waves are determined by relations between waves' propagation and constitutive characteristics of the medium. The results are numerically presented using the concept of slowness curves which are inverse velocity curves. The results are processed in Matlab software.

E-mail of corresponding author: sale1311@gmail.com
Phone +381 60 6157 757

TM 26

ON THE TRANSITION FROM DELOCALIZED TO LOCALIZED VIBRON STATES IN MACROMOLECULAR CHAINS

M. Nešić, S. Galović, D. Čevizović

Vinča Institute of Nuclear Sciences, University of Belgrade, Serbia

In this paper, the nature of vibron excitation states in quasi-1D macromolecular chains is investigated. Our study makes use of the simple mean field approach, based on the usage of variational extension of the Lang-Firsov unitary transformation. We take into account vibron interaction with acoustical phonon modes. We find that quasi-free (Bloch-like) vibron states may occur at low temperature limit, but, increasing the system temperature, localized vibron states may appear.

E-mail of corresponding author: mioljub.nesic@vinca.rs

Phone +381 64 14 17 188

TM 27

SUBSTITUTION EFFECTS OF SUMANENE BENZYLIC POSITIONS WITH BORON AND NITROGEN ATOMS

**Stevan Armaković¹, Sanja J. Armaković², Igor J. Šetrajčić¹, Jovan P.
Šetrajčić^{1,3}**

¹University of Novi Sad, Faculty of Sciences, Department of Physics, Trg Dositeja
Obradovića 4, 21000, Novi Sad, Vojvodina, Serbia,

²University of Novi Sad, Faculty of Sciences, Department of Chemistry,
Biochemistry and Environmental Protection, Trg Dositeja Obradovića 3, 21000,
Novi Sad, Vojvodina, Serbia,

³Academy of Sciences and Arts of Republic of Srpska, Bana Lazarevića 1, 78000
Banja Luka, Republic of Srpska, B&H

Sumanene is typical representative of molecular bowls, also known as buckybowl or π -bowls, which has three sp^3 hybridized carbon atoms at the benzylic positions as characteristic structural feature. This specific structural property may allow functionalization of new bowl shaped structure.

In this work we investigated effects of substitution of carbon atoms of sumanene with boron and nitrogen atoms within DFT computations. We used quantum molecular descriptors in order to better understand sumanene molecule and the effects of substitution. Isomers obtained by substitution were further subjected to the study of NMR parameters, NICS, bowl-to-bowl inversion barrier, NBO/NPA analysis and DOS analysis.

E-mail of corresponding author: stevan.armakovic@df.uns.ac.rs

Phone +381 21 485 2816

TM 28

THERMAL PROPERTIES OF CARBON NANOTUBES

Z. P. Popović, M. Damnjanović, I. Milošević

NanoLab, Faculty of Physics, University of Belgrade, Serbia

We study thermal expansions and thermal conductance of single-wall CNTs (carbon nanotubes) and helically coiled CNTs by using atomistic simulations techniques based on the Brenner interatomic potential. Model of helically coiled CNTs consists of periodically incorporated heptagons and pentagons into the hexagonal network. Initial atomic coordinates, which are further subjected to the optimization procedure, are generated with a help of the graph theory and topological coordinate method. Temperature dependence of geometrical parameters of the both systems is determined by minimizing Helmholtz free energy at a finite temperature. Local harmonic approximation with implementation of line group symmetry is used for calculation of Helmholtz free energy. Molecular dynamic simulations are applied to calculate thermal transport of CNTs. Coefficients of thermal conductance is obtained by Fourier's law. Direct method for calculation of thermal conduction is applied to simulate thermal energy transport of CNTs, which mimics the natural thermal transport process.

E-mail of corresponding author: zokapop@yahoo.com

Phone +381 69 372 15 90

TM 29

RAMAN SCATTERING STUDY OF IRON-CHALCOGENIDE SUPERCONDUCTORS

N. Lazarević¹, S. Bogdanović¹, Hechang Lei², C. Petrović², Z. V. Popović¹

¹Center for Solid State Physics and New Materials, Institute of Physics Belgrade,
University of Belgrade, Pregrevica 118, 11080 Belgrade, Serbia

²Condensed Matter Physics and Materials Science Department, Brookhaven
National Laboratory, Upton, New York 11973-5000, USA

Since the first report on superconductivity in LaOFeAs—known as the “1111” structure from its formula unit—at the end of February 2008, the iron pnictide superconductor family has been quickly expanded to different types of structures. Two certainly most intriguing are 11 and 122* iron chalcogenides. Here we present detailed analysis of the temperature dependent polarized Raman scattering spectra of $\text{FeTe}_{1-x}\text{Se}_x$ and $\text{K}_x\text{Fe}_{2-y}\text{Se}_2$ based on the crystal structure peculiarities. In $\text{K}_x\text{Fe}_{2-y}\text{Se}_2$ we found 16 Raman-active modes corresponding to the vibrations of the atoms of the I4/m symmetry structure. Two additional modes, for which we believe that originates from the vibrations within the FeSe layer of the I4/mmm symmetry, have also been observed. This shows the presence of phase separation between superconducting and isolating regions that occurs at nano-scale as a consequence of Fe vacancy ordering.

E-mail of corresponding author: nenadl@ipb.ac.rs

TM 30

EFFECTS OF OPTICAL BEAM MULTIPLE REFLECTIONS TO PHOTOTHERMAL RESPONSE - CORRECTION FACTOR

**Slobodan Todosijević¹, Zlatan Šoškić¹, Nebojša Bogojević¹,
Dalibor Čevizović², Slobodanka Galović^{2,3}**

¹Faculty of Mech. Eng. Kraljevo, University of Kragujevac, Dositejeva 19, Kraljevo,
Serbia,

²Vinča Institute of Nuclear Sciences, University of Belgrade, p.o.Box 522, Belgrade,
Serbia

³Joint Institute for Nuclear Research, Bogoliubov Laboratory of Theoretical Physics
Dubna, Russia

The theoretical model of optically induced surface temperature changes is presented, which includes multiple reflections of incident beam at the boundaries of a sample. The prediction of the model is compared with previous models and correction factor is derived. It is shown that this correction can affect the values of the surface temperature if the sample is optically transparent. The consequences of this correction are discussed to photothermal evaluation of optical coefficient of absorption and other properties of transparent and semi-transparent samples.

E-mail of corresponding author: todosijevic.s@mfkv.rs

Phone +381 63 366682, +381 64 8938898

TM 31

SiC MOSFET – MODEL OF CURRENT-VOLTAGE CHARACTERISTICS AND POSSIBILITIES FOR IT'S IMPLEMENTATION IN BIOMEDICAL ELECTRO EQUIPMENT

**Zorana Z. Golubović¹, Petar M. Lukić¹,
Milan T. Milovanović², Vladan M. Lukić¹, Rajko M. Šašić³**

¹Faculty of Mechanical Engineering, University of Belgrade, Kraljice Marije 16,
11120 Belgrade, Serbia

²Military Medical Academy, Crnotravska 17, 11000 Belgrade, Serbia

³Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4,
11120 Belgrade

In this paper, the new analytical model of SiC MOSFET current-voltage characteristics is proposed. Possibilities for SiC based MOSFET implementation in biomedical electro equipment are considered. To make a high current, high voltage power MOSFET (Metal Oxide Semiconductor Field Effect Transistor), silicon carbide (SiC) can be used. Unique semiconductor SiC possesses superior electronic properties such as high electron saturation drift velocity, high breakdown electric field, high thermal conductivity, wide band gap energy; SiC can be thermally oxidized etc. Thus, SiC MOSFET is employed as power electronic device. One of the most important MOSFET characteristics is its current – voltage characteristics ($I - V$ characteristics). These characteristics depend on many parameters and parameters optimization requires great efforts. SiC MOSFETs can be used in biomedical electro devices which operate under extreme conditions.

E- mail of corresponding author: zgzolubovic@mas.bg.ac.rs

Phone +381 63 8326050

TM 32

STUDY OF THE SHUNGITE AND COANP STRUCTURES NONLINEAR PROPERTIES

Svetlana Likhomanova, Natalia V. Kamanina

Vavilov State Optical Institute, 12, Birzhevaya Line, St.-Petersburg, 199034, Russia

It is well known that carbon nanoobjects such as fullerenes, carbon nanotubes and graphene can activate the nonlinear properties of doped organic materials, which could be useful for power limiting application, for example, to protect the human eyes and photosensitive equipments. The aim of this study was an investigation the nonlinear properties of a natural material – shungite, which contains the fullerenes molecules, in comparison with 2-cyclooctylamino-5-nitropyridine (COANP) system. The suspensions of shungite in water and chloroform were used for measurements of output energy and calculations the attenuation of laser energy with wavelength 532 nm and pulse duration 10 ns. We have obtained the significant laser attenuation of shungite in water suspension. The same effect of decrease of laser power was observed in suspension with chloroform too. The results of our experiments can be explained by the changes of nonlinear parameters such as third-order nonlinear susceptibility and nonlinear refraction due to an existence of dipole moment in fullerene molecules and intrinsic the dipole moment of shungite. The explanation has been coincided with that made for COANP-fullerene system, which reveals the increased local volume polarizability testified via increased cubic nonlinearity.

E -mail of corresponding author: lsv-87@bk.ru

Phone +7 (921) 325 31 96

TM 33

STUDY OF THE CARBON NANOTUBES INFLUENCE ON THE REFLECTION SPECTRUM AND THE MICROHARDNESS OF KBr, BaF₂

Pavel V. Kuzhakov, Natalia V. Kamanina

Vavilov State Optical Institute, 12, Birzhevaya Line, St.-Petersburg, 199034, Russia

It is well known that the basic features of carbon nanotubes are regarded to their high conductivity, strong hardness of their C—C bonds as well as complicated and unique mechanisms of charge carrier moving. These peculiarities of carbon nanotubes and their possible optoelectronics application is placed under consideration of different scientific group are placed. In the current paper the effect of surface carbon nanotubes influence on the surface properties of the soft" materials of the UV and the IR spectral ranges are discussed. Among the materials the main aspect is given to the structures such as potassium bromide and barium fluoride. The spectral characteristics and mechanical features are considered. The correlation between spectral and mechanical parameters of the materials studied is shown. The possible areas of application are done.

E-mail of corresponding author: kpv_2002@mail.ru

TM 34

TG-DTG STUDY OF BULK As_2S_3 CHALCOGENIDE GLASSES DOPED WITH BISMUTH

M. R. Dimitrievska, M. Šeljegović, Lj. Dačanin, K. Čajko

University of Novi Sad, Department of Physics, Trg Dositeja Obradovića 4,
Novi Sad, Serbia

Thermal properties of the glasses from the system $\text{Bi}_x(\text{As}_2\text{S}_3)_{100-x}$ with $x = 0, 1.5, 3, 5$ and 7 at. % Bi were studied by thermo gravimetric analysis. TG-DTG curves were measured using the device TA Instruments SDT Q600. Analysis of TG-DTG recordings for the samples at the same heating rate allowed characterization of the decomposition process and weight loss as a function of the content of doping atoms. A more complex decomposition process in the reference to other samples is observed for the sample with 7 at. % Bi, which can be explained by melting of the already existing crystallization centers in amorphous matrix. The presence of the crystallization centers in the glass $\text{Bi}_7(\text{As}_2\text{S}_3)_{93}$ is confirmed by the appearance of peaks on X-ray pattern of this sample. Identification of existing crystallization centers was carried out by X-ray analysis. The diffractograms of the other investigated samples from the system confirmed their amorphous character.

E- mail of corresponding author: mirjana.dimitrievska@df.uns.ac.rs
Phone +381643218187

TM 35

SYNTHESIS AND CHARACTERIZATION OF $\text{NiFe}_2\text{O}_4/\text{BaTiO}_3$ THIN FILMS

**Branimir Bajac¹, Jelena Vukmirović¹, Ivan Tokić¹,
Stevan M. Ognjanović¹ Akos Kukovecz², Vladimir V. Srdić¹**

¹Department of Materials Engineering, Faculty of Technology, University of Novi
Sad, Novi Sad, Serbia

²Department of Applied and Environmental Chemistry, University of Szeged,
Szeged, Hungary

Recently there has been great research interest in composite multiferroic materials with the coexistence of both magnetic and polarization ordering. Among them, multiferroic films with alternating ferrite and piezoelectric oxide layers have special importance since they are less leaky and facilitate on-chip integration. In this paper different multilayered $\text{NiFe}_2\text{O}_4/\text{BaTiO}_3$ (NFO/BTO) films were obtained on glass substrates by using dip and spin coating deposition techniques of different precursor solutions at room temperature. Two NiFe_2O_4 sols obtained from nickel and iron nitrate salts were examined, one dissolved in water at pH=3, while the second one was dissolved in 2-methoxyethanol with pH=1.8. BaTiO_3 acetate sol was obtained at room temperature by mixing of Ba^{2+} ions (from BaCO_3) dissolved in concentrated acetic acid with tetrabutyl-orthotitanate. Optimal processing conditions for the preparation of multilayered NFO/BTO films were investigated. Viscosity and particle size of precursor sols were monitored with time and thin films were characterized by X-ray diffraction (XRD), scanning electron microscopy (SEM), transmission electron microscopy (TEM) and atomic force microscopy (AFM). It was shown that the prepared films were crack-free, with the thickness of 100-400 nm, uniform surface texture and rounded grains having grain size in the nanometer range.

E-mail of corresponding author: banebns@gmail.com
Phone +381 64 3149566

TM 36

PHASE DIAGRAM OF SYSTEM $\text{ZnGeAs}_2\text{-CdGeAs}_2$

**I.V. Fedorchenko¹, A.N. Aronov¹, A.H. Yusupov², L. Kilanski³,
V. Domuchowski³**

¹Kurnakov Institute of General and Inorganic Chemistry Russian Academy of Sciences, 119991, Moscow, Russia, Leninskiy pr.31

²National University of Science and Technology "MISIS", 119049, Moscow, Russia, Leninskiy pr.4

³Institute of Physics, Polish Academy of Sciences, 02-668, Warsaw, Poland, al. Lotnikow 32/46

The chalcopyrites ZnGeAs_2 and CdGeAs_2 are advanced materials for nonlinear optic applications. The interaction between ZnGeAs_2 and CdGeAs_2 was investigated with a view to create a new materials with widen range of energy gap and to choose the technological conditions of the $\text{Zn}_x\text{Cd}_{1-x}\text{GeAs}_2$ solid solution single crystal growth. Eleven equilibrium samples of $\text{Zn}_x\text{Cd}_{1-x}\text{GeAs}_2$ with interval 0.1 mol. were synthesised and studied by X-ray, DTA and microstructure analyses. Above 873 K solid solutions of $\text{Zn}_x\text{Cd}_{1-x}\text{GeAs}_2$ take place in all range of the concentration. Lower of this temperature the decay's region was observed. The samples ZnGeAs_2 and $\text{Zn}_{0.9}\text{Cd}_{0.1}\text{GeAs}_2$ crystallized as chalcopyrite and sphalerite structures. The temperature of the phase transitions were 1123 K. The reported study was supported by RFBR, research project No 12-03-31203.

E-mail of corresponding author: fedorkin-san@rambler.ru
Phone +7-903-973-59-13

TM 37

**THE INFLUENCE OF THE STRUCTURE OF THE POLYBUTADIENE BLOCK
TO THE MECHANICAL PROPERTIES OF THE POLYSTYRENE-b-
POLYBUTADIENE-b-POLY(METHYL METHACRYLATE) COPOLYMER**

**Dragutin Nedeljković, Aleksandar Stajčić, Aleksandar Grujić, Jasmina
Stevanović, Jasna Stajić-Trosić**

University of Belgrade, Institute of Chemistry, Technology and Metallurgy,
Njegoševa 12, 11000 Belgrade, Serbia

Triblock copolymers with the defined structure are the materials of the great importance in the class of elastic materials. Combining the hard, brittle block(s) with the block with the elastic properties, it is possible to get material with the wide range of properties and application. In this work, the influence of the structure of the elastic block to the macroscopic properties of the material was investigated. The mechanical properties were determined by the measurement of stress-strain diagram. In this series of experiments, it was found that higher amount of 1,4 butadiene causes moderate stress drops as well as the higher elongation.

E-mail of corresponding author: dragutin@tmf.bg.ac.rs

Phone +38 11-3370412

TM 38

INFLUENCE OF GOLD IONS IMPLANTATION AND OZONE TREATMENT ON SURFACE CHARACTERISTICS OF HIGH DENSITY POLYETHYLENE

Danilo Kisić¹, Miloš Nenadović¹, Borivoj Adnađević², Zlatko Rakočević²

¹Vinča Institute of Nuclear Sciences, Laboratory of Atomic Physics, University of
Belgrade, Mike Alasa 12–14, 11001 Belgrade, Serbia

²Faculty of Physical Chemistry, University of Belgrade, Studentski trg 12-16 11158
Belgrade, Serbia

The aim of this work was to investigate possible effects of gold implantation and ultraviolet (UV) ozone treatment on surface characteristics of High Density Polyethylene (HDPE). HDPE had been modified by implantation of gold ions with the energies of 50 keV, 100 keV, 150 keV and 200 keV, and dose of $5 \cdot 10^{15}$ ions/cm². After implantation, the samples have been treated with UV ozone. Atomic force microscopy (AFM) topography images have shown significant changes of structure after implantation, and even more considerable changes after UV ozone treatment. UV ozone treatment caused significant changes of roughness on HDPE implanted with gold ions of the 150 keV and 200 keV energies. AFM phase imaging has shown that the surface composition is homogenous after implantation and UV ozone treatment. The analysis of the phase image histograms has shown that after implantation of gold ions the surface of HDPE becomes harder, with dominant elastic character, while UV ozone treatment causes softening of HDPE surface, i.e. dominant viscous character.

E-mail of corresponding author: dankisic@vin.bg.ac.rs

Phone +381 11 340 88 90

TM 39

**SURFACE DIELECTRIC PROPERTIES AND FOURIER TRANSFORM
INFRARED (FTIR) SPECTROSCOPY SPECTRA OF LOW-DENSITY
POLYETHYLENE (LDPE) ETCHED BY TRICHLOROACETIC ACID AT
DIFFERENT TEMPERATURES**

K. Simonović¹, F. Marinković¹, J. Dojčilović¹, D. Dudić²

¹Faculty of Physics, University of Belgrade, Studentski trg 12-16, 11001 Belgrade,
Serbia

²Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001
Belgrade, Serbia

Samples of LDPE were treated with Trichloroacetic acid for a period of one hour, at temperatures ranging from 20 to 70 °C. After the treatment, samples were analyzed with FTIR spectroscopy. FTIR measurements revealed chemical and crystallinity changes on the surface, as a direct result of the treatment. Surface dielectric properties of acid etched LDPE were analyzed in frequency range from 20 Hz to 200 kHz. In comparison to the untreated LDPE, the etched samples showed significantly different values of the conductance (G) at low frequencies. It was also observed, at higher frequencies, that an increase of the acid treatment temperature can result with lower values of conductance as well as susceptance (B) of LDPE with respect to the acid treatment at room temperature.

E-mail of corresponding author: kosta@ff.bg.ac.rs
Phone +381691402333

TM 40

NOVEL ASYMMETRIC POLYETHERSULFONE MEMBRANES FOR ULTRAFILTRATION APPLICATION

**Aleksandra Nešić¹, Sava Veličković², Filip Radovanović³, Aleksandra
Nastasović³**

¹Vinča Institute of Nuclear Sciences, University of Belgrade, Mike Petrovića Alasa
12-14, Belgrade, Serbia

²Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4,
Belgrade, Serbia

³Institute for Chemistry, Technology and Metallurgy, University of Belgrade,
Studentski trg 16, Belgrade, Serbia

Ultrafiltration has recently become popular as a promising separation method in many industrial processes covering fractionation and concentration steps in the food, pharmaceutical and biotechnology industries as much as in water and wastewater treatments. This paper presents the synthesis of novel asymmetric polyethersulfone membranes containing an interpenetrating network of poly(glycidyl methacrylate) (PGMA). In order to improve the properties and application range of membranes, the epoxy groups from PGMA are converted to amine groups by ring opening under alkaline conditions. Membranes before and after functionalization are characterized by FTIR-ATR, elemental analysis and water permeability.

E-mail of corresponding author: anesic@vin.bg.ac.rs

Phone +381691402333

TM 41

SURFACE MODIFICATION OF PECTIN SPHERES CROSS-LINKED IN LEAD (IV) AND COPPER(II) SOLUTIONS

**Sanja Šešlija¹, Jasmina Stevanović², Tatjana Volkov-Husović³, Sava
Veličković³**

¹Innovation Centre of the Faculty of Technology and Metallurgy, University of
Belgrade, Belgrade, Serbia,

²Institute of Chemistry, Technology and Metallurgy, Department of
Electrochemistry, Belgrade, Serbia,

³University of Belgrade, Faculty of Technology and Metallurgy, Belgrade, Serbia

The purpose of this study was to investigate the surface modification of spherical pectin particles obtained during gelling pectin. Pectin was cross-linked in salt solutions of copper (II) and lead(IV). Depending on concentration (0.01, 0.05 and 0.1 mol dm⁻³) and type of salt dimensions and deviation from sphericity were measured. Sensitization and activation of the pectin spheres' surfaces was carried out in tin(II) chloride and palladium(II) chloride solutions (0.1 mol dm⁻³), respectively. In order to metallize them, the samples were treated in solution of silver(I) nitrate. For the characterization of the particles optical microscopy was used, and results were analyzed by image analysis.

E-mail of corresponding author: sseslija@tmf.bg.ac.rs

Phone +381 60 6262612

TM 42

FERROMAGNETIC POLYANILINE/TiO₂ NANOCOMPOSITES

**Marija Radoičić¹, Zoran Šaponjić¹, Gordana Ćirić–Marjanović²,
Zorica Konstantinović³, Jovan Nedeljković¹**

¹Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

²Faculty of Physical Chemistry, University of Belgrade, Belgrade 11158, Serbia

³Institut de Ciència de Materials de Barcelona, CSIC, Bellaterra 08193, Spain

Novel ferromagnetic polyaniline PANI/ TiO₂ nanocomposites were synthesized by the oxidative polymerization of aniline with ammonium peroxydisulfate in an aqueous medium, in the presence of colloidal TiO₂ nanoparticles ($d \sim 4.5$ nm), without added acid. The morphological, magnetic, structural, and optical properties of the PANI/TiO₂ nanocomposites prepared at initial aniline/TiO₂ mole ratios 80, 40, and 20 were studied by scanning electron microscopy, superconducting quantum interference device, FTIR, Raman, and UV-Vis spectroscopies. The electrical conductivity of synthesized composites was $\sim 10^{-3}$ Scm⁻¹. The room temperature ferromagnetic response was detected in all investigated PANI/TiO₂ nanocomposites.

E-mail of corresponding author: mradoicic@vinca.rs

Phone +381 11 3408 291

TM 43

COMPARISON OF DIELECTRIC PROPERTIES OF POLYVINYLIDENE FLUORIDE + Cu COMPOSITES MEASURED BY SINUSOIDAL AND TRIANGLE SIGNALS

I. Petronijević, K. Simonović, R. Dojčilović, V. Đoković, D. Dudić

¹Faculty of Physics, University of Belgrade, Studentski trg 12-16, 11001 Belgrade,
Serbia

²Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001
Belgrade, Serbia

Comparative analysis of dielectric properties of Polyvinylidene fluoride (PVDF) + Cu composites has been done by applying sinusoidal and triangle electrical signals. Composite samples were processed from the PVDF solution with different filler mass concentrations. Two basic dielectric quantities are measured: amplitude of electrical conductivity (Y) and delay angle (ϕ). These results were compared with the electrical response of parallel RC circuits. Dielectric responses of composites, measured by applying sine and triangle voltage waveforms, differ and that difference in some way is similar to a difference obtained by numerical simulation of parallel RC circuit. The results are showing that by comparing dielectric quantities measured by sinusoidal and triangle signals can get more information about physical parameters which define electrical behavior of polymers and polymer composites. This material and composites at all are interesting in aspect of wide field of application in electronics and open new field for research.

E-mail of corresponding author: ivanpetronijevic@ff.bg.ac.rs
Phone + 381 11 7158 176

TM 44

POLY(METHACRYLIC ACID)/ZEOLITE HYDROGEL COMPOSITES AS ADSORBENTS FOR CATIONIC DYE REMOVAL

Vesna V. Panić¹, Željka Madžarević², Tatjana Volkov-Husović², Sava J.Veličković²

¹Innovation Center of Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4, Belgrade, Serbia

²Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4, Belgrade, Serbia

Hydrogel composites based on poly(methacrylic acid) and two types of zeolite, zeolite A and zeolite ZSM-5, have been synthesized, characterized by FTIR and SEM analysis and their adsorption potentials toward model cationic dye, Basic Yellow 28, have been investigated under various experimental conditions. Adsorption kinetics and equilibrium studies were performed by batch technique. Adsorption capacities showed to be highly sensitive to change of the initial dye concentration, solution pH, sorbent mass and temperature, as well as zeolite concentration, much more than zeolite type. Image analysis proved to be useful method for analysis of uniformity of coloration along the hydrogel diameter.

E-mail of corresponding author: vpanic@tmf.bg.ac.rs
Phone + 381 63 83 27 583

TM 45

SYNTHESIS OF POLYSTYRENE-G-STARCH COPOLYMERS USING NEW INITIATOR/CO-INITIATOR SYSTEM

Vladimir Nikolić¹, Sava Veličković², Aleksandar Popović³

¹Innovation Center, Faculty of Chemistry, University of Belgrade, Studentski trg
12-16, 11000 Belgrade, Serbia

²Faculty of Technology and Metallurgy, University of Belgrade, Karnegijeva 4,
11000 Belgrade, Serbia

³Faculty of Chemistry, University of Belgrade, Studentski trg 12-16, 11000
Belgrade, Serbia

Potassium persulfate (PPS)/different aminewas the new initiator/co-initiator system for synthesis of PS-*g*-starch copolymers. Propyl-, iso-butyl-, pentyl-, hexyl-, triethylamine, *N,N*-dimethyl- and *N,N*-diethylethanolamine, 1-(2-hidroxyethyl) piperazine and 4-(2-hidroxyethyl) morpholine were used with aim to obtain graft copolymers with the highest percentage of grafting. All amines, except *N,N*-diethylethanolamineand piperzine derivate, shown activator influence on polymerization reaction. Maximal obtained percentage of grafting was doubled compared with system without amine (32.55 % with *N,N*-dimethylethanolaminecompared to 16.24 %). Copolymers were characterized with FTIR spectroscopy and SEM microscopy. PPS/amines were better systems for initiation than PPS, which was confirmed by higher percentage of grafting.

E-mail of corresponding author: nikolicv@chem.bg.ac.rs

Phone + 381 11 3303805

TM 46

A SURVEY OF GRP PIPES DEFECTS AND DAMAGE DUE TO FABRICATION PROCESSES

**Abdelrhman Aljadeed¹, Ahamed Younaes¹, Ramadan A. Al-Madani^{1*},
M. Jarnaz²**

¹Al-Jabel Algharbi University, Engineering Faculty, Gharian, Libya

²Academy of Graduate Studies, Tripoli - Libya

Glass Reinforced Plastics, commonly known by various standards, as Fiber Reinforced Plastics (FRP), GRP, Glass Fiber Reinforced Plastic (GFRP), Reinforced Plastic Mortar Pipe (RPMP) or Reinforced Thermo-set Resin Plastic (RTRP) is an amalgamation of resin, glass fiber, manufactured using appropriate additives and treatment methods. It is a composite engineering material uniquely capable of meeting a wide variety of specific processes and end product requirements of various applications of fluid transport requirements. GRP piping system is often utilized in almost all applications to withstand aggressive service, ambient and environmental conditions. It has been successfully used in various piping systems and applications over the entire world. GRP piping industry in Libya has just started (i.e. since 2009), so such manufacturing plants are facing some difficulties during commissioning and production trials stage. GRP piping industry has wide variations both in designs and manufacturing techniques used. This paper will focus and discuss the possible defects associated with the production of GRP pipes, from the point of view of human error, and or lower manufacturing skillness due to new technology introduction in the area. The survey will involve and demonstrate the usage of GRP for various aggressive fluids and its consequences. Finally, concluding remarks concerning defect types, causes, and the prevention of such defects will be presented.

E-mail of corresponding author: ramadanalmadani@ymail.com

Phone + 381 11 3303805

Y 1

INVESTIGATION OF THE STRUCTURE OF NANOSIZED Na_nCl_n ($n = 8, 16, 24, 32$) CLUSTERS USING GLOBAL OPTIMIZATION

J. Zagorac^{1,2}, J. C. Schön¹, M. Jansen¹

¹Max Planck Institute for Solid State Research, Stuttgart, Germany;

²Materials Science Laboratory, Vinča Institute of Nuclear Sciences, University of Belgrade, Serbia

The structure and stability of alkali halide clusters has long been a focus of research in an effort to understand crystal growth. In our study, several global optimization methods have been applied to locate both the global and metastable minima of Na_nCl_n ($n = 8, 16, 24, 32$) clusters. The global search methods using empirical potentials consisted of both simulated annealing (SA) and multiquench (MQ), for comparison, where both single and multi-atom moves were used to generate new atom configurations. For simulated annealing, different temperature schedules, initial temperatures and stopping criteria were applied, while for the multiquench algorithm the number of random walks, number of quenches and the initial temperatures were varied. Furthermore, we used four different types of moveclasses, varying the percentage of single and multi-atom moves. For the larger clusters ($n = 24, 32$), the multiquench method found an average lower energy minima than simulated annealing. The most relevant lowest-energy structures for the Na_nCl_n clusters are visualized and analysed with the structure drawing and analysis program KPLOT, where we analyze the degree to which the clusters found resemble cut-outs of bulk NaCl.

E-mail of corresponding author: d.zagorac@fkf.mpg.de

Phone +49 (0)711 689 1367

Y 2

ELECTRICAL PROPERTIES OF MULTIDOPED CERIA

M. Stojmenović¹, B. Matović¹, M. Žunić²

¹Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia,

²Institute for Multidisciplinary Research, University of Belgrade, Belgrade, Serbia

Ceria ceramics were obtained from multidoped nano sized ceria powders prepared by both modified glycine nitrate procedure (MGNP) and self - propagating reaction at room temperature (SPRT). Oxides of rare earth elements such as Nd, Sm, Gd, Dy, Y were used as dopants in constant range of 20 %. Samples were sintered at 1500 °C for 1 h in an air atmosphere. XRPD, SEM and electrochemical impedance spectroscopy (EIS) measurements were carried out on sintered samples. It was found that single - phase was formed for each of the sintered materials. The values of conductivity at 700 °C are higher for the SPRT sample than for MGNP, 2.19×10^{-2} and $1.40 \times 10^{-2} \text{ Scm}^{-1}$, respectively.

E-mail of corresponding author: mpusevac@vin.bg.ac.rs

Y 3

DOPED WITH Cu AND Mn ZINC OXIDE CATALYSTS FOR OXIDATION OF Co IN NOXIOUS EMISSIONS

Katya Milenova, Penko Nikolov

¹Institute of Catalysis, Bulgarian Academy of Sciences Akad. G. Bontchev str. bl 11,
1113, Sofia, Bulgaria

²University of Chemical Technology and Metallurgy, 8 Kliment Ohridski, 1756,
Sofia, Bulgaria

Quick pace of growth of energetics and chemical industry conduct to sharp contamination of the air, the water and the soil. The presence in the air of noxious emissions from CO, NO, H₂S and hydrocarbons from internal-combustion engines and from the industry is a serious ecological problem. Innovative technology for decrease of the air contamination is the catalytic disposal. Carbon monoxide can be removed using catalytic complete oxidation, which is achieved with catalysts possessing high effectiveness and stability on variable thermodynamic conditions. Supported catalysts, using Pt active component or catalysts with high content of oxides of 3d transition metals are widespread for decontamination of waste gases. There is a great interest in recent catalyst synthesis technologies related to low percentage catalysts with active phase Cu and Mn on supporting precursor. The choice of precursor for the active component and for the basic composite appears as important problem. In the recent work doped with Cu and Mn zinc oxide is investigated as catalyst for oxidation of CO. The pure ZnO and the doped with Cu and Mn zinc oxide samples are synthesized from carbonate, acetate and nitrate precursors. The content of Mn and Cu in the catalyst compositions is less than 1 wt%. For structural, textural and morphological characterization of the samples trivial methods were used: AAS analysis, X-ray Diffraction, BET, EPR, XPS, IR as well as measurements of catalytic activity. The results from recent investigation show high perspective for the catalysts, synthesized from carbonate precursors.

E-mail of corresponding author: kmilenova@mail.bg

Phone +359 897 401 552

Y 4

MINIMAL REQUIRED WALL THICKNESS COMPARATIVE ANALYSIS OF A STABLE TANK MADE OF STAINLESS STEEL WITH AND WITHOUT STRAIN HARDENING

Katarina Jovičić¹, Iva Najvirt², Ana Alil¹, Milica Pilipović³

¹Institut Goša d.o.o., Milana Rakića 35, 11000 Belgrade, Serbia

²Nordvik d.o.o., Pariske Komune 22, 11000 Belgrade, Serbia

³Grundfos Srbija d.o.o., Omladiskih Brigada 90b, 11000 Belgrade, Serbia

Minimal required wall thickness calculation of a stable horizontal tank, intended for storage of cryogenic gases, made of austenitic stainless steel X5CrNi 18-10 was performed. Tank wall thickness calculation is done in the case of both, strain hardened and non-treated materials. The strain hardened austenitic stainless steel X5CrNi 18-10 achieves significant quantity material savings. Also, the application of strain hardened steel for making tankers, besides savings in materials, increase energy efficiency.

E-mail of corresponding author: katarina.jovicic217@gmail.com

Phone +381 63 8332-887

Y 5

REDUCING THE PARTICLE SIZE OF Bi_2O_3 AND Fe_2O_3 FOR THE SYNTHESIS OF BiFeO_3

Maria Čebela¹, Marija Prekajski¹, Mia Omerašević¹, Branko Matović¹

¹Materials Science Laboratory, Vinča Institute of Nuclear Sciences, University of
Belgrade, Belgrade, Serbia

For the synthesis of BiFeO_3 from a Bi_2O_3 - Fe_2O_3 mixture it is necessary to reduce the particle size to promote the reaction between initial powders. The aim of this work was therefore to obtain a reduced particle size of the initial Bi_2O_3 and Fe_2O_3 powders. For this purpose, we used ball milling in a planetary mill. We analyzed the particles size using granulometric analysis and scanning electron microscopy (SEM). The phase composition of the sintered samples was determined by x-ray diffraction (XRD) analysis. The powders behavior was characterized by TG/DTA and Heating stage microscope.

E-mail of corresponding author: mcebela@vinca.rs

Phone +381 60 311-04-36

Y 6

COMPARATIVE ANALYSIS OF CESIUM SORPTION BEHAVIOR OF THERMALLY, MECHANO-CHEMICALLY MODIFIED AND RAW DIATOMITE

Mia O. Omerašević¹, Miljana M. Mirković¹, Uroš D. Jovanović², Anja Došen¹, Ljiljana M. Kljajević¹, Maria Ž. Čebela¹, Snežana S. Nenadović¹

¹Laboratory for Material Science, Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

²Laboratory of Chemical Dynamics and Permanent Education, Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

This paper presents the effect of cesium immobilization in raw and modified diatomaceous earth (DE). The modification of diatomite was carried out by mixing DE and Ti powder mechano-chemically and thermally. X-ray diffraction (XRD) of samples modified by different way was performed. Thermal processing leads to structural changes of diatomite. The specific surface of diatomite which is available for the immobilization of cesium is modified by mechano-chemical treatment for different time. The effect of cesium immobilization of diatomite increases after mechano-chemical treatment, especially for 22h.

E-mail of corresponding author: mia@vinca.rs

Phone +381 65 983-10-10

Y 7

ELECTRON SPIN RESONANCE STUDY OF GAMMA-IRRADIATED PLLA

D. Miličević, D. Milivojević, E. Suljovrujić

Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

In order to examine the influence of the initial preparation conditions on the evolution of free radicals in gamma irradiated PLLA, the samples with a large variation in crystallinity were prepared and exposed to gamma radiation, in air, to various absorbed doses (up to 100 kGy). The presence and evolution of free radicals were followed using ESR spectroscopy for three weeks. In addition, SEM, DSC and WAXD measurements were performed. Presented results show that depending on the initial preparation conditions, the radiation-induced changes in structure and properties of PLLA, as well as the evolution of free radicals, can differ significantly.

E-mail of corresponding author: dejanmilicevic@vinca.rs

Phone +381 11 3408 607

Y 8

ADSORPTION OF ARSENIC(III) FROM AQUEOUS SOLUTIONS ON CARBON CRYOGEL

**Tamara Minović, Biljana Babić, Milovan Stoiljković, Vesna Maksimović,
Jelena Gulicovski, Jelena Pantić, Branko Matović**

Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001
Belgrade, Serbia

Arsenic (III) removal from aqueous solutions using carbon cryogel (CC) as adsorbent is presented in this work. CC was obtained by pyrolyzing resorcinol-formaldehyde cryogels in an inert atmosphere. Characterization by nitrogen adsorption showed that the CC was micro and mesoporous materials with high specific surface areas. The adsorption of arsenic (III) on CC was studied as a function of arsenic concentrations, as well as a function of solution pH. Kinetic and equilibrium studies were performed.

E-mail of corresponding author: tamaraminovic@vinca.rs
Phone +381 63 739 6029

Y 9

THE EFFECT OF CHEMICAL COMPOSITION ON SWELLING AND MECHANICAL PROPERTIES OF P(HEMA/IA/OEGA) HYDROGELS

M. Mićić, E. Suljovrujić

Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

In order to investigate the influence of mole fractions of IA and OEGA on swelling properties of the P(HEMA/IA/OEGA) hydrogels, a series of hydrogels with different mole ratios of monomers was synthesised by gamma radiation. The hydrogels were characterised by swelling studies in wide pH and temperature range, confirming pH and thermo responsiveness and a large variation in the swelling capability. In addition, FTIR, SEM, and mechanical measurements were performed, confirming that the properties of hydrogels can be tuned by variation in their composition. Presented results show that the obtained hydrogels can be a beneficial synergetic combination for controlled drug delivery.

E-mail of corresponding author: majamicic@vinca.rs

Phone +381 11 3408607

Y 10

NUMERICAL SIMULATION OF ROPEWAY OF AIRLIFT VEHICLES: CALCULATION METHODS

Miranda Kullolli, Alfred Hasanaj

Polytechnic University of Tirana, Albania

In first phase of this paper we will briefly give some historical background on the ropeway usage and its importance. Also in this paper we will present the vortex-exited vibration on which we later will create bicable model ropeway via dynamic model. Continuing with these models we will discuss the technical and natural problems that airlift ropeway transportation face. Further we will present the mathematical model using numerical stimulation which satisfies the engineering criteria. Also we will discuss the simulation of load and stresses happening in different stages of air lift operations. The final point for discussion in this paper is the measuring method for chair lifts and bicable ropeways. Calculation for the operation and its validation and maintenance will be made using calculation models such are MBS and ADAMS programs.

E-mail of corresponding author: alfred.hasanaj@yahoo.com

Phone +355 (0) 692 109 838

Y 11

LOCAL AND ELECTRONIC STRUCTURE AROUND MANGANESE IN MULTI-COMPONENT SEMICONDUCTORS

**M. Medić, I. Radisavljević, N. Novaković, B. Kuzmanović, B. Paskaš
Mamula, N. Ivanović**

Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001
Belgrade, Serbia

Magnetic ions diluted in semiconductor host make it possible to manipulate spin of electron in addition to its charge, which has a great potential for application of these materials, known as diluted magnetic semiconductors (DMS), in future spintronic devices. This paper presents studies of manganese in multicomponent II-VI and IV-VI semiconductors by means of X-ray Absorption Fine Structure (XAFS). Complemented with the first-principles calculations, these studies provide detailed information on local and electronic structure around manganese ions which is essential for better understanding of magnetic interaction mechanisms in these technologically very important materials.

E-mail of corresponding author: mirjanamedic@vinca.rs
Phone +381 11 3408 601

Y 12

SURFACE DIELECTRIC RELAXATIONS OF ISOTACTIC POLYPROPYLENE (IPP) AFTER ILLUMINATION AT DIFFERENT TEMPERATURES

V. Čubrović¹, I. Petronijević¹, J. Dojčilović¹, D. Dudić²

¹Faculty of Physics, University of Belgrade, Studentski trg 12-16, 11001 Belgrade,
Serbia

²Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001
Belgrade, Serbia

Isotropic iPP sheets were obtained by compression molding at 470 K and pressure of 1.75 MPa, followed by quenching in water at room temperature. Square shaped samples (1x80x80 mm) were cut from the middle of the sheets. For electrodes we have used system of parallel metal wires, which has been hot pressed on the samples surface. Distance between two neighboring wires is 1.5 mm. Dielectrical measurements were performed at 1 kHz before, during and after illumination with visible light at different temperatures ranging from 20 to 50 °C. Prior to the measurements, samples were kept for seven days in the dark. Period of illumination is 10 s. After illumination dielectrical relaxation were recorded for four hours. Consequence of illumination is drop of AC conductivities and relaxation of this process is accelerated with increasing temperature.

E-mail of corresponding author: vcubrovic@ff.bg.ac.rs
Phone +381 65-28-44-923

Y 13

SURFACE DIELECTRIC PROPERTIES OF LOW-DENSITY POLYETHYLENE (LDPE) IN VARIABLE VISIBLE LIGHT CONDITIONS

F. Marinković¹, V. Čubrović¹, J. Dojčilović¹, D. Dudić²

¹Faculty of Physics, University of Belgrade, Studentski trg 12-16, 11001 Belgrade,
Serbia

²Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001
Belgrade, Serbia

Isotropic LDPE sheets were obtained by compression molding at 460 K and pressure of 1.75 MPa, followed by quenching in water at room temperature. Square shaped samples (1x80x80 mm) were cut from the middle of the sheets. For electrodes we have used system of parallel metal wires, which has been hot pressed on the samples surface. Distance between two neighboring wires is 1.5 mm. Dielectrical measurements were performed at 1 kHz before, during and after illumination with visible light. Prior the measurements, samples were kept for seven days in the dark. Period of illumination varied between 2 and 100 s. After illumination dielectrical relaxation were recorded for two hours. Since the samples were kept in the dark, illumination triggered the drop in electrical conductivities. This effect can be explained with formation of electron curtain around metal wires.

E-mail of corresponding author: filip@ff.bg.ac.rs

Phone +381605881360

Y 14

NEW FACILE SYNTHESIS ROUTE FOR OBTAINING PHASE PURE LiFePO₄/C COMPOSITE

M. Milović¹, D. Jugović¹, M. Mitrić², B. Jokić³, D. Uskoković¹

¹Institute of Technical Sciences of SASA, Belgrade, Serbia,

²Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia,

³Faculty of Technology and Metallurgy, University of Belgrade, Belgrade, Serbia
Carbon-coated, olivine-structured LiFePO₄ suitable for cathode material in rechargeable lithium batteries has been prepared via simple, both time and energy low-consuming synthesis route. Procedure involves soaking of cellulose-containing substrate with aqueous precursor solution followed by rapid (several minutes long) single heating step. Cellulose acts as (i) reducing agent and (ii) carbon source which suppresses growth and prevents agglomeration of LiFePO₄ particles and also creates a conductive network in LiFePO₄/C composite. The crystalline structure, morphology and charge/discharge performance of the as-prepared composite were investigated by means of XRD analysis, electron microscopy and galvanostatic charge-discharge tests. LiFePO₄/C powder appears to be nanocrystalline (with mean crystallite size of 37 nm) and electrochemically stable achieving 97% (165 mAhg⁻¹) of its theoretical capacity.

E-mail of corresponding author: milos.milovic@itn.sanu.ac.rs

Phone +381 11 2636 994, +381 11 2185 437

Y 15

BIOACTIVE COMPOSITE MATERIALS IN REGENERATION OF THE RESORBED BONE OF ALVEOLAR RIDGES

**Z. Ajduković¹, N. Ignjatović², J. Rajković³, S. Najman⁴, D. Mihailović⁵, N.
Petrović¹, D. Kenić Marinković¹, D. Uskoković²**

¹Department of Prosthodontics, Clinic of Stomatology, Faculty of Medicine University of
Niš, Serbia

²Institute of Technical Sciences of SASA, Belgrade, Serbia

³Department of Biology and Ecology, Faculty of Science and Mathematics,
University of Niš, Serbia

⁴Institute for Biomedical Research, Faculty of Medicine, University of Niš, Serbia

⁵Institute of Pathology, Faculty of Medicine, University of Niš, Serbia

Bone loss during the systemic osteoporosis has an important role in dentistry and medicine. The aim of the study was the application of bioactive micro-and nano-composite materials, alone, and in combination with autologous plasma in osteoporotic jaw bones of rats with artificially induced osteoporosis. The effect of these composites was measured by histomorphometric and atomic absorption spectrophotometric analysis. According to the best obtained results in regeneration and recovery of the resorbed alveolar bone, it can be concluded that nano-composite combined with autologous plasma may be the material of choice to replace the osteoporotic damaged jaw bone.

E-mail of corresponding author: jelena.rajkovic@gmail.com

Phone +381 64 2423287

Y 16

EFFECTS OF DIFFERENT CRYOPROTECTANTS ON MORPHOLOGY OF LYOPHILIZED POLY(ϵ -CAPROLACTONE) MICRO- AND NANOSPHERES

Petar Stupar¹, Magdalena Stevanović², Nenad Filipović², Vladimir Pavlović^{2,3}, Jana Nunić⁴, Sandra Cundrič⁵, Metka Filipič⁴, Dragan Uskoković²

¹Faculty of Physical Chemistry, University of Belgrade, Belgrade, Serbia

²Institute of Technical Sciences of SASA, Belgrade, Serbia

³Faculty of Agriculture, University of Belgrade, Belgrade, Serbia

⁴Department of Genetic Toxicology and Cancer Biology, National Institute of Biology, Ljubljana, Slovenia

⁵Biotechnical Faculty, University of Ljubljana, Ljubljana, Slovenia

A common limitation of using polymeric micro- and nanoparticles in long-term conservation is due to their poor physical and chemical stability. Freeze-drying is one of the most convenient methods that enables further reconstitution of micro- and nanoparticles for therapeutical use. Nevertheless, this process generates various stresses during freezing and desiccation steps. The aim of this study was to evaluate different cryoprotectants (protective excipients that are usually added to increase stability upon storage and protect the particles from freezing stress): sugars (glucose and sucrose) and polymers (PVA and PGA), on the outcome of freeze-dried poly(ϵ -caprolactone) micro- and nanospheres. The best freeze-drying results in terms of morphological characteristics, analyzed with SEM, were achieved with glucose at concentration of 1%. The FT-IR analysis confirmed that the molecular structure of PCL particles remained the same after the addition cryoprotectants.

E-mail of corresponding author: orepst@hotmail.com

Phone +381 63 75-153-70

Y 17

**SYNTHESIS AND CHARACTERIZATION OF SELENIUM NANOPARTICLES
IN THE PRESENCE OF BOVINE SERUM ALBUMIN OR POLY (L-
GLUTAMIC ACID) FOR BIOMEDICAL APPLICATION**

**Nenad Filipović¹, Magdalena Stevanović¹, Srečo D. Škapin², Ines
Bračko², Dragan P. Uskoković¹**

¹Institute of Technical Sciences of SASA, Knez Mihailova 35/IV, 11000 Belgrade,
Serbia

²Advanced Materials Department, Jožef Štefan Institute, Jamova 39, 1000
Ljubljana, Slovenia

It is well known that selenium is an essential trace element playing the important roles in mammalian biology. For years, its organic forms and some salts have been used in studying its biological effects, but recently, selenium nanoparticles have gained great attention as a possible material for cancer therapy and prevention. In this study we report the facile and reproducible method for synthesis the nano-selenium by reduction of sodium selenite with an ascorbic acid, in the presence of a two stabilizing agents, bovine serum albumin (BSA) or poly (L-glutamic acid) (PGA). The influence of the sodium selenite/ascorbic acid ratio as well as the types and concentration of two different stabilizers on the wide variety physicochemical characteristics of obtained particles were examined.

E-mail of corresponding author: nenad.filipovic@itn.sanu.ac.rs
Phone +381 11 2636 994

Y 18

THE EFFECT OF AGING OF MILLED CLAY ON THE REMOVAL EFFICIENCIES OF HEAVY METAL IONS

**Anđelka Đukić¹, Ksenija Kumrić², Tatjana Trtić-Petrović², Jasmina
Grbović Novaković¹, Sanja Milošević¹, Igor Milanović¹, Ljiljana Matović¹**

¹Laboratory of Materials Sciences, Vinča Institute of Nuclear Sciences, University
of Belgrade, P.O. Box 522, 11001 Belgrade, Serbia,

²Laboratory of Physics, Vinča Institute of Nuclear Sciences, University of Belgrade,
P.O. Box 522, 11001 Belgrade, Serbia

The influence of aging on sorption behavior of natural clay modified by mechanical milling for different milling periods has been investigated. The changes in microstructure of samples investigated by PSD and XRD analysis were correlated with adsorption properties. Aging affects the removal efficiencies of the heavy metal ions (Pb(II), Cd(II), Cu(II), Zn(II)) from aqueous solutions. The best removal efficiencies as function of time of aging show sample milled 19h in comparison with samples milled 1 h, 2 h and 10 h. This behavior is related to the re-crystallization of the amorphized 19h milled sample.

E-mail of corresponding author: andjelka.djukic@vinca.rs

Phone + 381 11 3408 507

Y 19

**DRUG DELIVERY PARADIGM OF SILVER IONS RELEASED FROM
AG/PVA HYDROGEL NANOCOMPOSITES OBTAINED BY RADIATION
CHEMISTRY NANOREACTOR METHOD**

**Jelena Krstić, Jelena Spasojević, Aleksandra Radosavljević,
Zorica Kačarević-Popović**

Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

The importance of silver ion in the biological response to nano-silver is widely recognized hence there is significant potential to improve nano-silver technologies through controlled release formulations. In this study, kinetic analysis of silver ion release of γ -radiolytically synthesized Ag/PVA hydrogel nanocomposites was performed. Different model fitting methods were applied to estimate kinetic parameters. The analysis showed that the curvilinear nature of the cumulative percent silver released versus time provides the closest fit between experimental observations and the nonlinear function. The investigation of antimicrobial properties indicate that the size of Ag nanoparticles beside their molar concentration, have meaningful importance for antibacterial activity.

E-mail of corresponding author: jelenak@vinca.rs

Phone +381 11 3408 291

Y 20

SWELLING STUDIES OF THERMO- AND pH-SENSITIVE Ag-POLY(NiPAAm/IA) HYDROGEL NANOCOMPOSITES SYNTHESIZED BY GAMMA IRRADIATION

**Jelena Spasojević¹, Jelena Krstić¹, Aleksandra Radosavljević¹,
Melina Kalagasidis-Krušić², Zorica Kačarević-Popović¹**

¹Vinča Institute of Nuclear Sciences, University of Belgrade, Belgrade, Serbia

²Faculty of Technology and Metallurgy, University of Belgrade, Belgrade, Serbia

This work describes radiation-chemical induced *in situ* incorporation of silver nanoparticles (Ag NPs) within poly(NiPAAm/IA) hydrogels, previously crosslinked also by gamma irradiation, in order to obtain hydrogel nanocomposites. The formation of Ag NPs was confirmed by appearance of characteristic surface plasmon absorption band at around 400 nm, while their average size was determined by Mie's theory. The physico-chemical properties of pure and nanocomposite polymer networks were investigated by swelling measurement under different conditions. Obtained results showed that amount of IA has strong influence on swelling and diffusion properties of hydrogels, whereas the influence of Ag NPs is almost negligible.

E-mail of corresponding author: jelenas@vinca.rs

Phone +381 11 3408 291

Y 21

KINETIC STUDY OF PALLADIUM SORPTION BY MACROPOROUS COPOLYMER

**Bojana Ekmešić¹, Danijela Maksin², Jelena Marković², Aleksandra
Nastasović¹ Antonije Onjia²**

¹University of Belgrade, Institute for Chemistry, Technology and Metallurgy,
Njegoševa 12, Belgrade, Serbia

²University of Belgrade, Vinča Institute of Nuclear Sciences, Mike Petrovića Alasa
12-14, Vinča, Serbia

Palladium is widely used in electronics, medicine and catalysis due its corrosion and oxidation resistance, electrical conductivity and catalytic activity. Because of high cost and low availability, the separation and recovery of palladium rises as important issue. In this study, the possibility of Pd ions recovery from aqueous solutions by macroporous poly(glycidyl methacrylate-co-ethylene glycol dimethacrylate) with attached diethylene triamine (PGME-deta) from acidic medium (pH 2) at 298 K was investigated. Four kinetic models were evaluated in order to understand the mechanism of palladium (II) sorption. The concentration of Pd(II) ions was measured by inductively coupled plasma optical emission spectroscopy (ICP-OES).

E-mail of corresponding author: bojanae@chem.bg.ac.rs
Phone +381 64 3657 333

Y 22

**EXAMINATION OF THE INFLUENCE OF NANOMATERIALS CALCIUM
PHOSPHATE/POLY-(DL-LACTIDE-CO-GLYCOLIDE) AND COBALT-
EXCHANGED HYDROXYAPATITE ON THE VIABILITY OF SAOS-2 CELLS**

**Stevo Najman¹, Sanja Stojanović¹, Žarko Mitić², Zorica Ajduković³,
Nenad Ignjatović⁴, Dragan Uskoković⁴**

¹University of Niš, Faculty of Medicine, Institute of Biology and Human Genetics,
Serbia

²University of Niš, Faculty of Medicine, Department of Pharmacy, Serbia

³University of Niš, Faculty of Medicine, Clinic of Stomatology, Department of
Prosthodontics, Niš, Serbia

⁴Institute of Technical Sciences of SASA, Belgrade, Serbia

We have examined the effect of extracts of the calcium phosphate/poly-(DL-lactide-co-glycolide) (CP/PLGA) and cobalt-exchanged hydroxyapatite (CoHAp), on the viability of Saos-2 osteoblast-like cells. Extracts were prepared by incubation for 3 days at 37⁰C in the cultivation medium. Conductivity and pH value of extracts were measured before viability assay. Cell viability was estimated by MTT test performed after 24 h incubation of cells with various concentrations of extracts. Extract of CP/PLGA acted more cytotoxic on osteoblasts than the extract of CoHAp. This difference in the effect of examined extracts can be explained by their different pH value and conductivity.

E-mail of corresponding author: s.sanja88@gmail.com

Phone +381 62 1348516

TH 1

POLYMORPHISM AND STABILITY OF MAGNESIUM BOROHYDRIDE

V. Ban¹, R. Černý², B. Richter³, T.R. Jensen³, C.J. Webb⁴
Y. Filinchuk¹

¹Institute of Condensed Matter and Nanosciences, Université Catholique de
Louvain, Place L. Pasteur 1, 1348 Louvain-la-Neuve, Belgium

²Laboratory of Crystallography, Université de Genève, 24 quai Ernest Ansermet,
1211 Genève, Switzerland

³Center for Materials Crystallography, Interdisciplinary Nanoscience Center &
Dept. of Chemistry, Aarhus University, Langelandsgade 140, 8000 Aarhus C,
Denmark

⁴Queensland Micro- and Nanotechnology Centre, Griffith University, Nathan 4111,
Brisbane, Australia

In order to understand stability and structural properties of magnesium borohydride, we studied, by synchrotron powder diffraction, temperature-induced transformation of its hexagonal α - and cubic γ -polymorphs, both under vacuum and hydrogen back-pressure. The decomposition pathways depend strongly on the experimental conditions. Under vacuum, γ -phase first turns into the β -phase, which further evolves into a related structure, hereafter called β' . α - $\text{Mg}(\text{BH}_4)_2$ under vacuum transforms into a new polymorph, called here ϵ . As a number of $\text{Mg}(\text{BH}_4)_2$ polymorphs resemble SiO_2 structures, the database of observed/predicted SiO_2 frameworks can help to identify new $\text{Mg}(\text{BH}_4)_2$ structures.

E-mail of the corresponding author: voraksm.ban@uclouvain.be

Phone: +32 10 479 005

TH 2

ACHIEVING HIGH CAPACITY REVERSIBLE HYDROGEN STORAGE: NaBH₄@Ni CORE-SHELL NANOPARTICLES

Meganne Christian, Kondo-François Aguey-Zinsou

Merlin Group, ARC Centre for Functional Nanomaterials, School of Chemical Engineering, The University of New South Wales, Sydney NSW 2052, Australia

Sodium borohydride (NaBH₄) is a promising on-board hydrogen storage material. This research aimed to pioneer high capacity reversible storage under reasonable conditions. Nanoprecipitation allowed synthesis of nano-NaBH₄, with some improvements but no reversibility. Nanoconfinement via a core-shell approach, NaBH₄@Ni, provided an effective strategy to contain the melted NaBH₄ core and its dehydrogenation products, preventing their dispersion. At 350 °C, a steady reversible capacity of 5 mass % was achieved with fast kinetics. Under a heating ramp hydrogen release began at only 60 °C. This strategy has great potential for major advancements in the design of effective hydrogen storage materials.

E-mail of the corresponding author: m.christian@student.unsw.edu.au
Phone +612 9385 4825

TH 3

HYDRAZINE BORANE AND DERIVATIVES AS CHEMICAL HYDROGEN STORAGE MATERIALS

Romain Moury, Umit B. Demirci, Philippe Miele

IEM (Institut Europeen des Membranes), UMR 5635 (CNRS-ENSCM-UM2),
Universite Montpellier 2, Place E. Bataillon, F- 34095, Montpellier, France

Hydrazine borane $\text{N}_2\text{H}_4\text{BH}_3$ has been recently presented as being a derivative of ammonia borane NH_3BH_3 with a potential in chemical hydrogen storage. Involved in this reserach field, we optimized the synthesis of this new material and established a complete data sheet (NMR, IR, XRD, TGA and DSC). However, under heating, it releases a significant amount of pure hydrazine and leads to the formation of a shock-sensitive solid residue, which make it unsuitable for application. Hence, we intensified our works on synthesizing derivatives, and we successfully prepared and characterized (NMR, IR, XRD, TGA and DSC) two different and novel (never-reported heretofore) materials. Both showed improved dehydrogenation properties while not displaying the drawbacks of their parent. Our last results will be presented during the *"the 1st European Early Stage Researcher's Conference on Hydrogen Storage"*.

E-mail of the corresponding author: romain.moury@iemm.univ-montp2.fr
Phone +33 (0)4 67 14 91 39

TH 4

PHYSICAL AND CHEMICAL PROPERTIES OF VOLATILE COMPLEX HYDRIDES

**E. Callini¹, A. Borgschulte¹, I. Lindemann², A.J. Ramirez-Cuesta³, M.
Chong⁴, C.M. Jensen⁴, O. Gutfleisch^{2,5}, A. Züttel¹**

¹Empa, Materials Sciences and Technology, CH-8600 Dübendorf, Switzerland

²IFW Dresden, Helmholtzstrasse 20, D-01069 Dresden, Germany

³ISIS Facility, Rutherford Appleton Laboratory, Chilton, Didcot, Oxon OX11 0QX,
United Kingdom

⁴Univ Hawaii Manoa, Dept Chem, Honolulu, HI 96822 USA

⁵Technische Universität Darmstadt, Petersenstrasse 23, 64287 Darmstadt,
Germany

The storage of hydrogen in liquid complex hydrides is an attractive option. However, difficulties in their characterization are a matter of fact, with particular emphasis on their gaseous products. The presented investigation method uses a thermogravimetric balance to monitor the sample weight loss, while the decomposition gas is detected by an infrared spectrometer. We investigate metathesis of $\text{Ti}(\text{BH}_4)_3$ and $\text{Al}(\text{BH}_4)_3$. It is shown that the analysis of the gas phase is fundamental in order to understand their decomposition reactions: these materials emit hydrogen containing volatile species, such as diborane, which is an issue for their reversibility.

E-mail of the corresponding author: elsa.callini@empa.ch

Phone +41 587654933

TH 5

AMMONIA BORANE NANOCONFINEMENT INTO CARBONACEOUS OR BORON-NITROGEN-BASED HOST MATERIALS

Georges Moussa, Umit B. Demirci, Samuel Bernard, Philippe Miele

IEM (Institut Europeen des Membranes), UMR 5635 (CNRS-ENSCM-UM2),
Universite Montpellier 2, Place E. Bataillon, F- 34095, Montpellier, France

Ammonia borane NH_3BH_3 is considered as the most promising chemical hydrogen storage material since the mid 2000s owing to its ability to liberate dihydrogen under heating. However, the 13 wt% H can be released only over the range 100-200 °C. To decrease these temperatures as well as to improve the purity of H_2 , one of the strategies we have adopted is to confine NH_3BH_3 (in tetrahydrofuran solution) into meso- and micro-porous hosts such as activated carbon or boron nitride. The as-formed composites were analyzed by ^{11}B MAS NMR, IR, XRD, TGA and DSC. One of the striking results is that NH_3BH_3 nanoconfined into activated carbon released dihydrogen in room conditions owing to acid-based reactions. Our last results will be presented during the *"the 1st European Early Stage Researcher's Conference on Hydrogen Storage"*

E-mail of the corresponding author: georges.moussa@iemm.univ-montp2.fr
Phone +33 (0)4 67 14 91 39

TH 6

THE EFFECTS OF Si AND EXPANDED PTFE SUBSTRATES ON FORMATION AND HYDROGENATION OF Mg AND Mg-Ti FILMS

M. Lelis^{1,2}, D. Milcius¹, R. Zostautiene³

¹Lithuanian Energy Institute, Breslaujos st.3, LT-44403 Kaunas, Lithuania

²Vytautas Magnus University, Vileikos st. 8, LT-44404 Kaunas, Lithuania

³Kaunas University of Technology, K.Donelaičio st.73, LT-44029 Kaunas, Lithuania

Usually metallic films for metal hydrides research are deposited on hard and flat substrates such as silicon, magnesium oxide etc. With the film thickness increases all films during hydrogenation deals with typical problems such as film brittleness and cracking. In this paper we demonstrate that metal Mg and Mg-Ti films for hydride formation can be successfully deposited on the flexible Expanded PTFE substrates. The received results for ePTFE substrates are compared to the films being deposited on Si with different plasma pretreatment. It is discussed that different interface properties between substrate and film can have effect on film morphology, crystallinity and its reaction with hydrogen.

E-mail of corresponding author martynas@hydrogen.lt

Phone: +370 612 52924

TH 7

METAL HYDRIDE FILMS FOR OPTO-ELECTRONIC APPLICATIONS

Trygve Mongstad

Institute for Energy Technology, Norway

Metal hydrides cannot only reversibly store hydrogen: they can also be applied as active materials in thin-film electrical and optical devices. A number of metal-hydride materials have been investigated for such purposes. This talk will give an overview of advances with respect to opto-electronic applications of thin-film metal hydrides. Such applications comprise technologies as smart windows, hydrogen sensors and potentially photovoltaics. I will go through relevant recent research results by our group: a study of the semiconducting properties of Mg-Ni hydride films and the recent discovery of a strong photochromic effect in films of yttrium hydride.

E-mail of corresponding author: trygve.mongstad@ife.no

Phone +47 99228200

TH 8

LASER SURFACE MODIFICATION OF MATERIALS

Sumsun Naher

Faculty of Engineering and Computing, Dublin City University, Dublin 9, Ireland

A series of alloys (H13 tool steel, 316 stainless steel, Ti-6Al-4V) and WC-CoCr coating surface were modified by a 1.5 kW CO₂ laser in both pulsed and continuous mode in a high speed sample processing condition. Several design of experiments (DoE) were carried out to optimize the laser processing parameters which include, peak power, spot diameter, pulse repetition frequency, irradiance etc. The effect of laser processing parameters on the microstructure, surface roughness, melt pool depth, phase transformation, residual strain, microhardness, and chemical composition were investigated. In most cases, laser surface modification has resulted in better surface properties of materials.

E-mail of corresponding author: sumsun.naher@dcu.ie

Phone: +353 1 851484946

TH 9

MODIFICATIONS IN MICROSTRUCTURE OF MgH_2 -BASED COMPOSITE UPON HYDROGEN CYCLING

A. Montone, A. Aurora, D. Mirabile Gattia, M. Vittori Antisari

ENEA – R.C. Casaccia – Technical Unit of Materials - Via Anguillarese, 301 00123
Rome, Italy

Kinetics modifications upon hydrogen cycling of composite materials obtained by ball milling MgH_2 with different additives were studied by the mean of a Sievert volumetric Apparatus. The microstructural evolution of the material during a sequence of H_2 sorption cycles was explored by the Scanning Electron Microscopy. Independently of the nature of the additive, the original microstructure of MgH_2 is strongly affected by the cycling procedure. Besides particle coarsening, the presence of material protruding from the particle surface in a worm-like structure is observed. Hollow particles constituted by empty MgO shell box are also noticed.

E-mail of corresponding author: daniele.mirabile@enea.it
Phone +39 0630483484

TH 10

TUNING THE HYDROGEN STORAGE OF MOFS BY INTRODUCING METALS ON THE LINKER SIDE

**Petra Ágota Szilágyi, Jana-Juan Alcañiz, Jorge Gascon, Hans Geerlings,
Bernard Dam**

Delft University of Technology, Netherlands

Hydrogen storage is arguably the biggest challenge in developing a hydrogen economy. While different types of materials are being explored as hydrogen carriers, porous materials have the advantage of offering fast kinetics for hydrogen sorption as well as reversibility over multiple cycles. Among them, metal-organic frameworks (MOFs) are being increasingly considered as promising materials for non-dissociative hydrogen adsorption. They display high crystallinity as well as high and regular porosity. In addition, their syntheses can be carried out under mild conditions, allowing for their rational design and facile pre- or post-synthetic modification. Hydrogen-storage capacity of MOFs can either be enhanced by increasing their surface area and pore volume or by increasing the hydrogen's isosteric heat of adsorption. While there are several publications showing that the introduction of metal ions in the pores by cation exchange or doping, it has only been recently suggested that the introduction of cations on the the linker site could also result in an increased hydrogen uptake. However, to date no convincing evidence has been provided for this theory, This work supplies the first data on how metal coordination on the linker site influences the hydrogen-storage properties of metal-organic frameworks. Pd²⁺ was infused in the pores of NH₂-MIL-101 as it specifically bind to the NH₂- functional groups. This enables the selective introduction of the Pd²⁺ on the linker site (as long as the loading remains sufficiently low). Subsequently hydrogen-adsorption data was collected at 77 K and ambient temperatures and the results were compared to those obtained on the pristine NH₂-MIL-101 under the same conditions. In the present work, the findings of this comparison, i.e. the effect of the metal infusion on the MOF's linker site will be discussed.

E-mail of corresponding author: P.A.Szilagyi@tudelft.nl
Phone+31 (0) 152782637

TH 11

SYNTHESIS AND HYDROGENATION OF $\text{Mg}_6\text{Pd}_{1-x}\text{TM}_x$ (TM = Ag, Cu) NOVEL SYSTEMS

M. Ponthieu^{1,2}, J. F. Fernandez¹, F. Cuevas², J.R. Ares¹, P. Adeva³, C. Sanchez¹

¹MIRE, Dpto. Física de Materiales, Facultad de Ciencias, Universidad Autónoma de Madrid, 28049, Madrid, Spain

²ICMPE/CNRS-UPEC UMR 7182, 2-8 rue Henri Dunant, 94320 Thiais Cedex, France

³Department of Physical Metallurgy, CENIM-CSIC, Av. De Gregorio del Amo 8, 28040 Madrid, Spain

Mg_6Pd intermetallic compound has been actively investigated as a hydrogen storage material. Substituting Pd by other transition metals can destabilize the hydride and reduce the cost of the alloy. $\text{Mg}_6\text{Pd}_{0.5}\text{TM}_{0.5}$ (TM = Ag, Cu) samples have been synthesized by induction melting. Scanning Electron Microscopy analysis, X-ray powder diffraction and Electron Probe Micro-Analysis were used to determine phase composition and abundance. Cu solubility in Mg_6Pd is limited to $x = 0.3$. For the Ag containing sample the ternary phase $\text{Mg}_{5.8}\text{Pd}_{0.7}\text{Ag}_{0.5}$ is formed. This one reversibly absorbs up to 3.1 wt.% H, decomposing into MgH_2 , Mg_5Pd_2 and MgPd phases, with Ag partially substituting Pd atoms. Both systems have been investigated by High Pressure Differential Scanning Calorimetry. Several (endo-) exothermic peaks are observed, indicating that (de-)hydrogenation occurs in several steps. The mechanisms involved during hydrogenation process will be discussed.

E-mail of corresponding author: ponthieu@icmpe.cnrs.fr

Phone +34 91 497 56 27

TH 12

STRUCTURE AND HYDROGEN STORAGE PROPERTIES OF THE HEXAGONAL LAVES PHASE $\text{Sc}(\text{Al}_{1-x}\text{Ni}_x)_2$

Martin Sahlberg

Department of Chemistry – Ångström Laboratory, Uppsala University, Box 538, SE-751 21 Uppsala, Sweden

The crystal structures of hydrogenated and unhydrogenated $\text{Sc}(\text{Al}_{1-x}\text{Ni}_x)_2$ Laves phases have been determined by combining several diffraction techniques and it is shown that hydrogen is situated interstitially in the A_2B_2 -site, which has the maximum number of scandium neighbours. The hydrogen absorption/desorption behaviour has also been investigated. It is shown that hydrogen absorption forms a solid solution of hydrogen in the mother compound. The hydrogen storage capacity exceeds 1.7 H/f.u. at 374 K. The activation energy of hydrogen desorption was determined to 4.6 kJ/mol H_2 .

E-mail of corresponding author: Martin.Sahlberg@kemi.uu.se

TH 13

MICROSTRUCTURE EVOLUTION IN THE CHIPS OF MG-10 AT.% NI ALLOY DURING HYDROGENATION

Larisa Popilevsky¹, Vladimir Skripnyuk¹, Yuri Estrin², A.K. Dahle³, Eugen Rabkin¹

¹Department of Materials Science and Engineering, Technion-Israel Institute of Technology, Haifa, Israel

²Centre for Advanced Hybrid Materials, Monash University, Clayton, Victoria 3800, Australia

³Materials Engineering, The University of Queensland, Frak White Bldg. #43, Brisbane, QLD 4072, Australia

The aim of the study was to establish the correlation between hydrogenation properties and microstructure of hypoeutectic Mg-10 at.% Ni alloy in the as-cast state and after processing by equal channel angular pressing (ECAP). The initial microstructures of as-cast and ECAP-processed alloys are quite similar, but intermittent hydrogenation behavior is very different. It was found that hydrogenation promotes growth of large faceted Mg single crystals in the chips of both alloys. Although a significant morphological difference between the chips of as-cast and ECAP-processed alloys was observed during initial hydrogenation cycles, after nine cycles the morphology and hydrogenation kinetics of both samples become nearly identical.

E-mail of corresponding author: laura.mat.eng@gmail.com

TH 14

**COMBINATORIAL SYNTHESIS AND HIGH THROUGHPUT
EXAMINATION OF HYDROGEN STORAGE ALLOY LIBRARIES**

M. Polanski, I. Kunce, J. Bystrzycki

Department of Advanced Materials and Technology, Military University of
Technology, 2 Kaliskiego Str., 00-908 Warsaw, Poland

Binary and ternary alloy libraries (from Fe-Ti and Fe-Ti-Ni systems) were synthesized by Laser Engineered Net Shaping technique. Discrete and continuous micro-libraries as well as large in-situ alloyed samples were prepared in a course of one experiment. Chemical composition examination, X-Ray phase analysis have shown that synthesis was performed with success. Fe-Ti and Fe-Ti-Ni samples absorbed significant amounts of hydrogen close to theoretical without any complication activation. New method for high throughput examination of hydrogen storage properties based on custom made multi-Temperature Programmed Desorption device is presented.

E-mail of corresponding author: mpolanski@wat.edu.pl

Phone +48 22 683 76 28, +48 502 768 678

TH 15

**STRUCTURAL CHARACTERIZATION AND ELECTROCHEMICAL
HYDROGEN STORAGE PROPERTIES OF Ti₂Ni AND (Ti_{1-x}Zr_x)₂Ni (x=0,
0.05, 0.1) ALLOYS PREPARED BY MECHANICAL ALLOYING**

Xianda Li¹, Omar Elkedim¹, M. Nowak², M. Jurczyk²

¹FEMTO-ST, Fc-Lab, Université de Technologie de Belfort-Montbéliard, Site de
Sévenans, 90010 Belfort cedex, France

²Institute of Materials Science and Engineering, Poznan University of Technology,
Sklodowska- Curie 5 Sq., 60-965 Poznan, Poland

Nominal Ti₂Ni were synthesized by ball milling. The effect of milling time on the characteristic and hydrogen performance was investigated. Amorphous phase was obtained after 60 hours of milling, which showed the best hydrogen performance. The sample containing mainly amorphous phase had stable discharge capacity at about 100mAh/g under room temperature without the need of activation. Zr substitution for Ti was also introduced to the 60h milling time Ti₂Ni powder with a formula (Ti_{1-x}Zr_x)₂Ni (x=0, 0.05, 0.1). It is found that Zr hindered amorphization process and increased the MT temperature of Ti-Ni system, thus an austenitic crystal structure was formed, which had steady but poor hydrogen discharge capacity under room temperature.

E-mail of corresponding author: xianda.li @utbm.fr
Phone + 33 3 84583545

TH 16

COMPUTATIONAL SCREENING OF MIXED METAL HALIDE AMMINES

Peter Bjerre Jensen¹, Steen Lysgaard¹, Ulrich Quaade², Tejs Vegge¹

¹DTU Energy Conversion, Technical University of Denmark, Fysikvej 307, DK-2800
Kgs. Lyngby ²AmmineX A/S, Gladsaxevej 363, DK-2860 Søborg

Metal halide ammines, e.g. $\text{Mg}(\text{NH}_3)_6\text{Cl}_2$ and $\text{Sr}(\text{NH}_3)_8\text{Cl}_2$, can reversibly store ammonia, with high volumetric hydrogen storage capacities. In this project we are searching for improved mixed materials with optimal desorption temperature and kinetics. We apply DFT calculations on mixed compounds selected by a Genetic Algorithm (GA), relying on biological principles of natural selection. The GA is evolving from an initial (random) population and selecting those with highest fitness, e.g. stability, release temperature and storage capacity. The search space includes all alkaline, alkaline earth, 3d and 4d metals and the four lightest halides, giving in total almost two million combinations.

E-mail of corresponding author: pbjen@dtu.dk
Phone +45 45253204

TH 17

**INVESTIGATION OF SURFACE AND NEAR-SURFACE EFFECTS ON
HYDROGEN DESORPTION PROPERTIES OF MgH_2**

**Sandra Kurko, Sanja Milošević, Igor Milanović, Radojka Vujasin,
Ljiljana Matović, Jasmina Grbović Novaković, Nikola Novaković**

Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11000
Belgrade, Serbia

In attempt to comprehend and improve desorption behaviour of MgH_2 we have investigated the influence of structural changes introduced by ion irradiation into MgH_2 matrix. The changes were followed by both, experiment and theory using XRD, particle size analysis, TPD and Ab initio calculations. Results suggest that there are several mechanisms involved in desorption process, which depend on defect concentration and their interaction and ordering. It has been demonstrated that the changes in near-surface area play the crucial role in desorption kinetics. It is also confirmed that there is possibility to control the thermodynamic parameters by controlling vacancies concentration.

E-mail of corresponding author: skumric@vinca.rs
Phone +381 113408507

TH 18

**COMPUTATONAL STUDY OF H STORAGE IN Pd AND Pt
NANOPARTICLES SUPPORTED ON MgO**

**Sergey M. Kozlov¹, Hristiyan A. Aleksandrov^{1,2}, Konstantin M.
Neyman^{1,3}**

¹Departament de Química Física and Institut de Química Teòrica i Computacional (IQTCUB), Universitat de Barcelona, C/Marti Franques 1, 08028 Barcelona, Spain

²Faculty of Chemistry, University of Sofia, 1126 Sofia, Bulgaria

³Institució Catalana de Recerca i Estudis Avançats (ICREA), 08010 Barcelona, Spain

The hydrogen storage properties of Pd are well known and could be improved by its nanostructuring. There is experimental evidence that Pt nanoparticles are also able to absorb hydrogen, in contrast with bulky Pt samples. In this work we investigate H adsorption and absorption at low and high coverage on scalable to bulk Pd and Pt nanoparticles supported on MgO(100) using density functional methods. It is found that high H surface coverage promotes H absorption into the bulk sites and makes this process feasible on Pd nanoparticles, but in the case of calculated Pt nanoparticles H absorption remains highly endothermic.

E-mail of corresponding author: sergey.m.kozlov@ub.edu

Phone +34 934034836

TH 19

HYDROGEN STORAGE PROPERTIES OF SUMANENE

Stevan Armaković¹, Sanja J. Armaković², Jovan P. Šetrajčić^{1,3}

¹University of Novi Sad, Faculty of Sciences, Department of Physics, Trg Dositeja Obradovića 4, 21000, Novi Sad, Vojvodina, Serbia,

²University of Novi Sad, Faculty of Sciences, Department of Chemistry, Biochemistry and Environmental Protection, Trg Dositeja Obradovića 3, 21000, Novi Sad, Vojvodina, Serbia,

³Academy of Sciences and Arts of Republic of Srpska, Bana Lazarevića 1, 78000 Banja Luka, Republic of Srpska, B&H

Hydrogen is concerned as an ideal energy carrier and ideal energy source for replacement of limited and harmful fossil fuels. Adsorbents of hydrogen have to have large surface areas and suitable binding energies. It is important to obtain intermediate binding energies between H₂ and adsorbing structure since these energies allow both adsorption and release. These conditions are reached by nanotubes and fullerenes, and, according to our work, by sumanene as well.

We investigated adsorption properties of sumanene towards H₂ using density functional theory computations. Investigation encompassed study of MEP surfaces, quantum molecular descriptors, NMR parameters, NICS, NBO/NPA analysis and DOS.

E-mail of corresponding author: stevan.armakovic@df.uns.ac.rs

Phone +381 21 485 2816

TH 20

**STRUCTURAL AND HYDROGEN SORPTION PROPERTIES OF $\text{SmNi}_{5-x}\text{Ga}_x$
SYSTEM – AN EXPERIMENTAL AND THEORETICAL STUDY**

Nikola Biliškov, Goran I. Miletić, Ivka Štefanić, Antun Drašner

Laboratory of Solid State and Complex Compounds Chemistry, Division of
Materials Chemistry, Ruđer Bošković Institute, Bijenička c. 54, HR-10000 Zagreb,
Croatia

A series of ternary alloys of general formula $\text{SmNi}_{5-x}\text{Ga}_x$ was prepared by melting stoichiometric mixtures of Sm, Ni and Ga in arc furnace under argon. From X-ray powder diffraction lattice parameters are determined and it is found that hexagonal structure of SmNi_5 (CaCu_5 type) is retained for all considered values of x . Hydrogen absorption ability and thermodynamic quantities of the systems are determined by pressure-composition desorption isotherms. The obtained properties are compared with those of previously reported SmNi_5 and SmNi_4Ga . DFT calculations were performed for selected members of the system and the results are discussed with regard to experiment.

E-mail of corresponding author: nbilis@irb.hr
Phone + 385 1 4571269

TH 21

INDUCTION PLASMA SYNTHESIS OF Mg_2Ni NANOPARTICLES FOR HYDROGEN STORAGE

Burak Aktekin¹, Gülhan Çakmak², Tayfur Öztürk¹

¹Middle East Technical University, Metallurgical and Materials Eng. Dept., Ankara, Turkey

²Bülent Ecevit University, Metallurgical Eng. Dept., Zonguldak, Turkey

Mg_2Ni is a well-known alloy that could store 3.6 wt.% hydrogen. Efforts to reduce the stability of this alloy mainly involved compositional alterations as well as reduction in the structural scale with such methods as mechanical milling or thin film deposition. In the present study, Mg_2Ni nanoparticles were synthesized in RF inductively coupled plasma with a nanopowder reactor. Conventional approach of feeding the prealloyed powder into the torch has resulted in disintegration of the intermetallic alloy, producing Mg and Ni nanopowders with a small fraction of targeted Mg_2Ni nanoparticles. As an alternative approach, the prealloyed powders were fed into the plasma from the bottom injector where the Mg_2Ni yield was slightly higher. Feeding the starting material in the form of elemental powders, Ni from the top and Mg from the bottom injector has yielded nanopowders <100 nm in size, where Mg_2Ni had a significant proportion. PCT isotherms of the synthesized nanopowders are being studied currently by a Sievert's type apparatus.

E-mail of corresponding author: baktekin@metu.edu.tr
Phone +90 544 7387499

TH 22

**POLYANIONIC GALLIUM-HYDROGEN CLUSTER IONS AS POSSIBLE
INTERMEDIATES IN DEHYDROGENATION REACTIONS**

Henrik Fahlquist

Stockholm University, 106 91 Stockholm, Sweden

LiBH_4 and NaAlH_4 have been regarded as promising hydrogen storage materials since both contain a high gravimetric content of hydrogen. The major challenge is the poor kinetics of hydrogen release, hence much research effort has been put in to enhance the dehydrogenation reactions. The cluster ions $[\text{Ga}(\text{GaH}_3)_4]^{5-}$, $[\text{Ga}_n\text{H}_{2n}]^{n-}$ and $[\text{Ga}_3\text{H}_8]^{3-}$ are formed in the air sensitive compounds $\text{Rb}_8\text{Ga}_5\text{H}_{15}$, AGaH_2 ($\text{A}=\text{K}, \text{Rb}$) and $\text{Cs}_{10}\text{Ga}_9\text{H}_{25}$ respectively by exposing a melt of alkali metals ($\text{K}, \text{Rb}, \text{Cs}$) and gallium to hydrogen gas at elevated temperatures. These compounds represent a new group of compounds featuring polyanionic gallium hydrogen clusters that may serve as intermediates in dehydrogenation reactions for the compounds mentioned above.

E-mail of corresponding author: henrik.fahlquist@mmk.su.se
Phone +46 737698301

TH 23

EFFECT OF A $\text{CaF}_{2-x}\text{H}_x$ SOLID SOLUTION FORMATION IN THE MIXED SYSTEM $\text{CaH}_2 + \text{CaF}_2 + \text{MgB}_2$

C. Pistidda¹, F. Karimi¹, S. Garroni², C. Milanese³, L. Rude⁴, J. Skibsted⁴, T. R. Jensen⁴, P. Nolis⁵, C. Horstmann¹, C. Gundlach⁶, M. D. Baró⁷, T. Klassen¹, M. Dornheim¹

¹Institute of Materials Research, Materials Technology, Helmholtz-Zentrum Geesthacht, Max-Planck-Straße 1, D-21502 Geesthacht, Germany;

²Dipartimento di Chimica, Università di Sassari and INSTM, Via Vienna 2, I-07100 Sassari, Italy;

³Pavia H2 Lab, C.S.G.I. & Dipartimento di Chimica, Sezione di Chimica Fisica, Università di Pavia, Viale Taramelli 16, I-27100 Pavia, Italy;

⁴Center for Materials Crystallography, iNANO, and Department of Chemistry, Aarhus University, Langelandsgade 140, DK-8000. Denmark;

⁵Servei de Resonància Magnètica Nuclear (SeRMN), Universitat Autònoma de Barcelona, 08193 Bellaterra, Spain;

⁶MAX-lab, Lund University, S-22100 Lund, Sweden;

⁷Departament de Física, Universitat Autònoma de Barcelona, 08193 Bellaterra, Spain

In this work the effect of a partial replacement of CaH_2 by CaF_2 on the sorption properties of the system $\text{CaH}_2 + \text{MgB}_2$ was studied. The first five hydrogen absorption and four desorption reactions of the $\text{CaH}_2 + \text{MgB}_2$ and the $3\text{CaH}_2 + \text{CaF}_2 + 4\text{MgB}_2$ systems were investigated by means of volumetric measurements, High Pressure Differential Scanning Calorimetric technique, MAS NMR analysis and in situ Synchrotron Radiation Powder X-ray diffraction. It was observed that already during the mixing of the reactants, the formation of a non-stoichiometric $\text{CaF}_{2-x}\text{H}_x$ solid solution takes place. The formation of the $\text{CaF}_{2-x}\text{H}_x$ solid solution sensibly affects the overall hydrogen sorption reactions of the system $\text{CaH}_2 + \text{MgB}_2$.

E-mail of the corresponding author: claudio.pistidda@hzg.de

TH 24

DESIGN, SIMULATION AND OPTIMISATION OF AN ANODE SUPPORTED SOFC

Hirad Azmin, Zahir Dehouche

School of Engineering and Design, Brunel University, Middlesex, UB8 3PH, United
Kingdom

Solid oxide fuel cell is one the technologies that has the potential to be leading the market in the race of clean energies. The first aspect of this research is the production of a sample fuel cell using the most common methods of powder compression and screen printing. The test cell is made using powders of the materials and then sintered in a furnace before other layers are added to complete the cell. The second aspect is the validation of a simulation model using the experimental data available from a commercialised fuel cell. The results of the simulation were validated by comparison between the experimental data. These results suggest a maximum theoretical power density of 0.41 W/cm^2 for the designed cell.

E-mail of corresponding author: hirad.azmin@brunel.ac.uk

Phone +44 (0)1895 267559

TH 25

HYDROGEN STORAGE IN POROUS MEDIA AT ROOM TEMPERATURE

Liga Grinberga, Janis Kleperis

Institute of Solid State Physics, University of Latvia, 8 Kengaraga Street, Riga, LV-1063,
Latvia

The main aim of this work is to find a porous material that is able to store hydrogen at room temperature and at pressure range from 1 to 10 bars. In our research activated natural zeolite samples are used. Activation of the samples is carried out in ion-exchange process and in pyrolysis process. As the result of the activation the samples were coated by palladium nanoparticles. Amount of stored hydrogen in the samples is determined by thermogravimetric and volumetric methods. The results show that the hydrogen sorption capacity of the coated material is greater than for zeolite or Pd separately.

E-mail of corresponding author: ligagr@cfi.lu.lv

Phone +37127038016

TH 26

ENHANCING THE HYDROGEN CARBON INTERACTION

Andreas Bliersbach

Empa, Swiss Federal Laboratories for Materials Science and Technology, Hydrogen
and Energy, Überlandstrasse 129, CH-8600 Dübendorf, Switzerland

Carbon is an easily available, light weight and abundant element, able to store large quantities of hydrogen. Covalently bound hydrogen, as in conventionally used hydrocarbons, is inconvenient for hydrogen access due to either high activation or binding energies. For physisorbed molecular hydrogen on high specific surface area carbonaceous materials the binding energy is low and cryogenic temperatures are required for storage. An increased, intermediate, binding energy of 20 – 40 kJ/mol H₂ is desirable and can be achieved by additional attracting forces between the hydrogen molecule and the host material either in the form of electrostatic (dipole or quadrupole) or orbital interaction (Kubas interaction).

E -mail of corresponding author: andreas.bliersbach@empa.ch

Phone: +41 58 765 48 62

TH 27

**HYDROGEN STORAGE PROPERTIES OF DIFFERENT METAL-DOPED
CARBON-BASED COMPOSITES**

**A. Ampoumogli¹, D. Giasafaki¹, A. Bourlinos², G. Charalambopoulou¹,
A. Stubos¹, D. Gournis³, Th. Steriotis¹**

¹National Center for Scientific Research "Demokritos", Athens/ Greece

²Department of Physics, University of Ioannina/ Greece

³Department of Material Science Engineering, University of Ioannina/Greece

An interesting approach for the efficient hydrogen solid storage is the use of metal doped carbon materials, where metal nanoparticles (e.g. Pd, Pt) seem to catalytically enhance hydrogen sorption at room temperature on the basis of the so-called spillover mechanism. In the present work we compare the room temperature hydrogen sorption behavior of different Pd-doped carbon-based composites resulting from the combination of various nanostructured carbons (foams, single- and multi- wall carbon nanotubes, fullerenes and graphite) in an attempt to elucidate possible binding sites that may assist the weak chemisorption hydrogen spillover process. Our low pressure (up to 1 bar) hydrogen uptake measurements point to possible intercalation of atomic hydrogen between the graphitic or multi-wall nanotube layers.

E -mail of corresponding author: asemam@chem.demokritos.gr

Phone: +30 210 6503636

TH 28

**PRODUCTION OF SINGLE – WALL CARBON NANOTUBES BY KrF
EXCIMER LASER ABLATION**

**Petru-Marian Bota, Iulian Boerasu, Dorel Dorobantu, Dionezie Bojin,
Marius Enachescu**

Centre of Surface Science and Nanotechnology – University Politehnica of
Bucharest, Splaiul Independentei 313, Romania

We report about catalyst influence on carbon single-wall nanotubes (SWNT) synthesis by means of KrF excimer laser ablation of catalyst doped targets prepared by a new technique. The novel target preparation eliminates the need for pressing or hot pressing on the target. The products have been investigated by high resolution transmission electron microscopy (HRTEM) and Raman spectroscopy. The laser synthesized SWNTs, at a furnace temperature of 1100 °C, have a narrow diameter distribution. Bundles of SWNTs intertwined with some amorphous carbon species and catalyst particles can be observed in the as-produced SWNTs deposits. The ratio of mixed metallic and semiconducting electronic type of as-grown SWNTs was investigated by means of micro-Raman spectroscopy. The products obtained will be tested in complex systems for their hydrogen storage capacity and other possible applications.

E-mail of corresponding author: marian.bota@gmail.com

Phone +40740699544

TH 29

**CONFINEMENT OF LiBH₄ IN NANOPOROUS CARBON MATERIALS:
SIZE AND INTERFACE EFFECTS ON HYDROGEN MOBILITY AND
RELEASE**

A. Nale, P.E. de Jongh

Inorganic Chemistry and Catalysis, Debye Institute for Nanomaterials Science,
Utrecht University, P.O. Box 80083, 3508 TB Utrecht, The Netherlands

Lithiumborohydride attracts much attention as a promising material for hydrogen storage (total hydrogen content: 18.5 wt%) but the elevated temperatures required to release the gas are an important obstacle for practical application. In this work we investigate nanosizing effects on decomposition kinetics and we evaluate the limiting factors for hydrogen release. LiBH₄ has been confined in different carbon hosts by melt-infiltration; decomposition profiles have been investigated by means of calorimetry, showing a lower dehydrogenation temperature and suggesting a different decomposition pathway than for bulk LiBH₄. Surprisingly, nanosizing affects even the LiBH₄ solid-solid phase transition temperature and Neutron Scattering and NMR highlighted changes in ion mobility close to interface between LiBH₄ and scaffold.

E-mail of corresponding author: ac.nale@uu.nl
Phone +31 6 22736392

TH 30

SCANDIUM FUNCTIONALIZED CARBON AEROGELS FOR HYDROGEN STORAGE

Payam Javadian¹, Marek Polanski², Thomas Plocinski², Dariusz Zasada², Jerzy Bystrzycki², Flemming Besenbacher³, Torben R. Jensen¹

¹Center for Energy Materials, Interdisciplinary Nanoscience Center (iNANO), and Department of Chemistry, Aarhus University, DK-8000 Aarhus, Denmark

²Faculty of Advanced Technology and Chemistry, Military University of Technology, 2 Kaliskiego Str., 00-908 Warsaw, Poland

³Interdisciplinary Nanoscience Center (iNANO) and Department of Physics and Astronomy, Aarhus University, DK-8000 Aarhus C, Denmark

A new nanocomposite material is synthesized for hydrogen storage based on nanoparticles of NaAlH_4 melt infiltrated into sc-functionalized carbon aerogel in order to explore possible synergetic effects between nanoconfinement and scaffold functionalization. The scaffold functionalization is utilized by a novel infiltration method, using aqueous ScCl_3 as a precursor. Upon drying of the functionalized scaffold, the new compound ScOCl is obtained. Nanoparticles of (5 wt%) ScOCl is residing inside the mesopores of the aerogel, confirmed by HR-STEM and EDS. The kinetic effect of hydrogen release is improved by 3 °C, for the sc-doped scaffold, measured by TPD-MS. During hydrogen release and uptake measured by Sieverts', the effect induced by melt infiltration into sc-functionalized aerogel leads to a significantly improved reversibility, remaining 62 % of its hydrogen storage capacity, compared to 54 % using pristine aerogel.

E-mail of corresponding author: payam_javadian@chem.au
Phone +45 22529132

TH 31

**DEVELOPMENT OF NOVEL 3D TEM SAMPLE STRUCTURING
METHODOLOGIES AND ADVANCED FIB AND TEM TOMOGRAPHY
APPLICATIONS FOR NANOPOROUS MATERIALS**

Meltem Sezen

Sabanci University Nanotechnology Research and Application Center, Orhanli,
Tuzla 34956 Istanbul, Turkey

Electron Tomography applications, in particular, are needed for imaging and analysis of ceramic and composite based 3D material systems, such as, hydrogen storage materials, solid oxide fuel cells. This study concerns the development of novel TEM/FIB tomography methodologies to reveal the geometric and chemical distribution of such material systems. In this scope, while the samples to be investigated in the TEM provide 3D information at the nanometer scale and below; via FIB tomography sectioning, the information in the scale ranging from micrometers to tens of nanometers is collected from the identical sample and the data are comparatively and complementarily evaluated.

E-mail of corresponding author: meltemsezen@sabanciuniv.edu
Phone +90 216 4839886

E 1

HYDROGEN SORPTION PROPERTIES OF $\text{MgH}_2/\text{NaBH}_4$ COMPOSITES

Sandra Kurko¹, Annalisa Aurora², Daniele Mirabile Gattia², Vittoria Contini², Amelia Montone², Željka Rašković-Lovre¹, Jasmina Grbović Novaković¹

¹Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11000 Belgrade, Serbia

²ENEA, UTTMAT Research Centre of Casaccia, Via Anguillarese 301, 00123 Rome, Italy

The hydrogen sorption properties of $\text{MgH}_2/\text{NaBH}_4$ composites prepared by high energy ball milling under Ar atmosphere were investigated. Samples with different content of NaBH_4 and different milling time were prepared. Microstructural and morphological analyses were done by X-Ray Diffraction and Scanning Electron Microscopy. Kinetic analyses and cycling under hydrogen pressure were performed in a Sievert's volumetric apparatus. Results suggest that both low milling time and lower content of NaBH_4 have beneficial influence on reactions kinetics and on the amount of sorbed hydrogen.

E-mail of corresponding author: skumric@vinca.rs
Phone +381 11 3408 507

E 2

COMBINED SOLID STATE AND HIGH PRESSURE HYDROGEN STORAGE

Elisabeth Grube, Torben R. Jensen

Center for Energy Materials, Interdisciplinary Nanoscience Center, and
Department of Chemistry, Aarhus University, Langelandsgade 140, DK-8000
Aarhus, Denmark

The aim is to enable future use of hydrogen as a carrier of renewable energy for transportation that may eliminate particle emissions and fossil fuels dependency. One approach is to combine solid state and high pressure onboard hydrogen storage as this has the potential to increase the storage capacity considerably. The idea is to design chemical reactions that are solely controlled by the pressure and the temperature in the storage tank and are able to absorb both hydrogen and heat. This concept has a potential for fast refueling, high capacity and long driving range.

E-mail of corresponding author: egrube@inano.au.dk
Phone +45 2467 8655

E 3

STRUCTURE AND HYDROGEN STORAGE PROPERTIES OF HIGH ENTROPY TiVCrNiFe AND TiVCrNiFeZr ALLOYS SYNTHESIZED BY USING LASER ENGINEERED NET SHAPING (LENS)

I. Kunce, M. Polanski, J. Bystrzycki

Department of Advanced Materials and Technology, Military University of Technology, 2 Kaliskiego Str., 00-908 Warsaw, Poland

TiVCrNiFe and TiVCrNiFeZr multi-principal-element alloys were fabricated by using the Laser Engineered Net Shaping technique. The influence of the LENS process parameters on the microstructure and hydrogen storage properties of the alloys was studied. The phase analysis was performed with X-ray diffraction. Thereafter, SEM/BSE with EDS mode was used to determine the chemical composition and homogeneity of the alloys. The P-C-T isotherms were measured at 50°C after activation for 2h at 500°C under vacuum. The obtained results showed that alloy with C14 Laves phase structure (FeNiTiCrVZr) are capable of storing relatively large amount of the hydrogen (~1,8% wt.) with rapid absorption/desorption kinetics at room temperature.

E-mail of corresponding author: ikunce@wat.edu.pl

Phone +48 22 683 76 93, +48 666 064 799

E 4

**CHARACTERIZATION OF MULTILAYERED HYDROGEN STORAGE
MATERIALS BY TRANSMISSION ELECTRON MICROSCOPY**

Tyché Perkisas¹, Sara Bals¹, Lennard Mooij², Bernard Dam²

¹Electron Microscopy for Materials Research (EMAT), University of Antwerp,
Groenenborgerlaan 171, 2020 Antwerp, Belgium, ²Department of Chemical
Engineering, Delft University of Technology, Julianalaan 136, 2628 BL, Delft, The
Netherlands

Transmission electron microscopy was used to characterize the structure and composition of Si-Fe-Mg-Fe-Pd and Si-Ti-Mg-Ti-Pd multilayered samples that have great interest for hydrogen storage. Both native and hydrogenated states were investigated. High-resolution electron microscopy and electron diffraction were used to derive a structure model for the layers. These results are complimented by chemical characterization done by EDX (Energy Dispersive X-ray Spectroscopy) and EELS (Electron Energy Loss Spectroscopy), which provides additional information regarding layer thickness and intermixing. It was found that the magnesium layer thickness is more consistent in the Si-Ti-Mg-Ti-Pd samples. Also the Mg to MgH₂ transition will be discussed.

E-mail of corresponding author: Tyche.Perkisas@ua.ac.be
Phone: +32 498 18 44 45

E 5

ELECTRONIC STRUCTURE AND FORMATION ENERGIES OF HfV_2H_x LAVES PHASE

Jana Radaković, Jelena Belošević-Čavor, Vasil Koteski, Katarina Ćirić

Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11000
Belgrade

We present calculations of electronic structure, formation enthalpies, and electric field gradients in cubic $\text{HfV}_2\text{H}_{x=1,2,3,4}$ Laves phase, where hydrogen occupies three interstitial positions, 96g, 32e, and 8b. To determine which one of these interstitials is most favorable for storing hydrogen, formation enthalpies were calculated for every site, while changing x . To investigate electronic structure modifications before and after hydrogenation, we calculated electric field gradients, and compared them with the nuclear magnetic resonance measurements, which enabled us to study the distribution of hydrogen in the crystal lattice, as well as the occupation of interstitials.

E-mail of corresponding author: janar@vinca.rs

Phone: 381 11 3408 601

E 6

STRUCTURAL AND ELECTRONIC FACTORS GOVERNING HYDROGEN STORAGE PROPERTIES OF TMNi COMPOUNDS (TM=Ti, Zr, Hf)

Katarina Ćirić, Jana Radaković, Vasil Koteski, Dragica Stojić

Laboratory for Nuclear and Plasma Physics, Vinča Institute of Nuclear Sciences,
University of Belgrade, P.O. Box 522, 11000 Belgrade

Hydrides of binary and ternary compounds formed between IVb group transition metals and nickel are studied using the full potential (L)APW+lo method within DFT. Differences in hydride formation behavior between ZrNi, HfNi and TiNi are addressed from structural and electronic point of view. Also, electronic structure, bonding and thermodynamical parameters of ternary compound $\text{Ti}_{0.64}\text{Zr}_{0.36}\text{Ni}$ and appropriate orthorhombic hydride are investigated. Contrary to TiNiH , $\text{Ti}_{0.64}\text{Zr}_{0.36}\text{NiH}$ is found to be thermodynamically favorable for hydrogen storage applications, with properties similar to ZrNiH and HfNiH . This implies that structure mainly determines properties of all investigated hydrides in this class of intermetallics.

E-mail of corresponding author: kciric@vinca.rs
Phone +381643937369

E 7

**HYDROGEN STORAGE, MICROSTRUCTURE AND MECHANICAL
PROPERTIES OF $\text{Mg}_{65}\text{Ni}_{20}\text{Cu}_5\text{Y}_{10}$ METALLIC GLASS DEFORMED BY
HIGH-PRESSURE TORSION**

Á. Révész¹, Á. Kis-Tóth¹, L.K. Varga², E. Schafler³, T. Spassov⁴

¹Department of Materials Physics, Eötvös University, Budapest, H-1518, P.O.B. 32,
Budapest, Hungary

²Research Institute for Solid state Physics and Optics, Hungarian Academy of
Sciences,
H-1525 Budapest, P.O.B. 49, Hungary

³Physics of Nanostructured Materials, Faculty of Physics, University of Vienna, A-
1090 Vienna, Austria

⁴Department of Chemistry, University of Sofia "St. Kl. Ohridski", 1 J. Bourchier str.,
1164 Sofia, Bulgaria

Rapidly quenched amorphous $\text{Mg}_{65}\text{Ni}_{20}\text{Cu}_5\text{Y}_{10}$ metallic glass compacts were subjected to heavy shear deformation by high pressure torsion until different amount of ultimate strain. High resolution X-ray diffraction analysis and scanning electron microscopy revealed that high-pressure torsion resulted in a deformation dependent microstructure. Nanoindentation measurements indicated that the heavy shear deformation yields an increase in hardness. High-pressure calorimetry measurements revealed that hydrogen uptake in the fully amorphous alloy occurs at a significantly lower temperature compared to the fully crystallized state, while the amount of absorbed hydrogen increased considerably after heavy shear deformation due to the formation of Mg_2Ni crystals.

E-mail of corresponding author: skiszkao@gmail.com

Phone +36-1-372-2823

E 8

EFFECTS OF PARTICLE SIZE AND TYPE OF CONDUCTIVE ADDITIVE ON THE ELECTRODE PERFORMANCES OF GAS ATOMIZED AB₅-TYPE HYDROGEN STORAGE ALLOY

L. Scherbakova, M. Spodaryk, Yu Solonin, A. Samelyuk

Institute for Problems of Materials Science (NASU), Krzhizhanovsky str., 3, Kyiv,
Ukraine

The structure, phase composition, surface condition of gas atomized powders of LaNi_{4.5}Al_{0.5} depending on the particle size (fractions: $\leq 50 \mu\text{m}$; 160-316 μm ; 630-1000 μm) have been studied. The electrochemical and storage properties of electrodes made from these powders with the addition of conductive additives as electrolytic copper powder or composite carbon additive (1 wt.% nanotubes + 7 wt.% nanosized carbon black) have been examined. In this study were used such methods as XRD, SEM, RPS, EIS and several electrochemical methods. The results showed that gas atomized powders with larger particle size in the initial state show better kinetics of hydrogen exchange reactions and high discharge capacity (up to 300 mA*h/g). The electrodes from powders of all fractions have good ability to high-rate discharge. The discharge process of electrodes, regardless of particle size and type of conductive additive is controlled by diffusion of hydrogen in the electrode. With increasing particle size of the alloy in the electrode diffusion coefficients of hydrogen increase from $6\text{-}8 \times 10^{-12} \text{ cm}^2/\text{s}$ for small and medium fractions to $3.5 \times 10^{-10} \text{ cm}^2/\text{s}$ for large fraction respectively. For electrode from LaNi_{4.5}Al_{0.5} with composite carbon additive diffusion coefficients of hydrogen during discharge, calculated from the data obtained by EIS are in good agreement with those obtained from cyclic current-voltage curves ($2\text{-}4 \times 10^{-9} \text{ cm}^2/\text{s}$).

E-mail of corresponding author: poshtamary@ukr.net

Phone +380685417935

E 9

HYDROGEN DESORPTION PROPERTIES OF $\text{MgH}_2/\text{LiAlH}_4$ COMPOSITES

**Igor Milanović¹, Sanja Milošević¹, Ljiljana Matović¹, Radojka Vujasin¹,
Nikola Novaković¹, Riccardo Checchetto², Jasmina Grbović Novaković¹**

¹Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11000
Belgrade, Serbia

²Faculty of Physics, University of Trento, 38123 Povo, Italy

Hydrogen storage properties of $\text{MgH}_2\text{-LiAlH}_4$ composite obtained by mechanical milling for different milling times and different weight percentage of LiAlH_4 have been investigated by temperature programmed desorption (TPD). The morphology of samples has been investigated by XRD and SEM analysis and correlated with desorption properties. $\text{MgH}_2\text{-LiAlH}_4$ composites show improved hydrogen desorption characteristics in comparison with the pure milled MgH_2 . It has been shown that quick mechanical milling/mixing of MgH_2 with small amount of LiAlH_4 shifts two characteristic hydrogen desorption maxima to the lower temperatures. The kinetics of hydrogen desorption in composites is also improved.

E-mail of corresponding author: igorm@vinca.rs
Phone +381 11 3408 507

E 10

EVALUATION OF THE HYDROGEN EMBRITTLEMENT OF METALLIC MEMBRANES BY THE DISK TEST METHOD

A. Metallari, I. Markja

Polytechnic University of Tirana, Albania

The safety of metallic structures or membranes used for hydrogen storage is strongly conditioned by the sensitivity of their materials to hydrogen. This sensitivity is studied by the disk pressure testing and the external hydrogen embrittlement index is defined as the ratio of rupture pressures under helium and hydrogen, p_{He}/p_{H_2} . The technique used simulates well the working conditions of pressure vessels and allows the identification of different effects related to the parameters of the environment, loading and material. The obtained data contribute to a better understanding of the embrittlement phenomenon, leading to a safer selection of appropriate materials for applications under gaseous hydrogen.

E-mail of corresponding author: metallarialban@hotmail.com

Phone +355 66 60 90 001

E 11

ELECTRONIC STRUCTURE AND CHARGE TOPOLOGY STUDY OF ALKALI HYDRIDES

**Bojana Paskaš Mamula, Mirjana Medić, Bojana Kuzmanović,
Ivana Radisavljević, Nenad Ivanović, Nikola Novaković**

Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001
Belgrade, Serbia

Electronic structure and Bader charge analysis calculation of alkali hydride series has been performed. Classification of charge topology structures of alkali hydrides within the broader frame of NaCl structures (along with alkali halides) has been established. Additionally, connection between charge topology and experimentally observed trends in macroscopic properties, such as B1-B2 phase transformation pressures, melting points, elastic properties etc., was considered.

E-mail of corresponding author: bpmamula@vinca.rs

Phone +381 11 3408 610

E 12

**REACTIVELY DEPOSITED THIN FILMS OF MAGNESIUM NICKEL
HYDRIDE**

**Željka Rašković-Lovre¹, Trygve Mongstad², Simon Lindberg³, Stefano
Deleda², Smagul Karazhanov²**

¹Vinča Institute of Nuclear Sciences, University of Belgrade, P. O. Box 552,
Belgrade, Serbia,

²Institute for Energy Technology, Box 40, NO-2027 Kjeller, Norway

³Chalmers University of Technology, SE-412 96 Gothenburg, Sweden

Uniform magnesium nickel hydride thin film was synthesized on large area glass substrate by reactive sputtering. Electrical and optical properties have shown the material of semiconducting nature with band gap of 1.6eV for the amorphous as-deposited sample and 2.1eV for the annealed sample with crystal structure. XRD results followed these transformations. Performed Hall measurements have shown wide range of resistivity. TPD measurements confirm release of hydrogen and also the release of impurity gases.

E-mail of corresponding author: zeljka.raskovic@vinca.rs

Phone +381 11 3408 507

E 13

INVESTIGATION OF NUCLEATION PROCESS IN MgH_2 THIN FILMS

Željka Rašković – Lovre¹, Sandra Kurko¹, Nenad Ivanović¹, Jasmina
Grbović Novaković¹, Jose Francisco Fernández², Jose Ramon Ares
Fernández², Carlos Sánchez², Nikola Novaković¹

¹Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11000
Belgrade, Serbia

²Departamento de Física de Materiales C-IV, Facultad de Ciencias, Universidad
Autónoma de Madrid, Cantoblanco, 28049 Madrid, Spain

In attempt to understand and improve the desorption process, the nucleation of MgH_2 thin films during desorption were followed by *in situ* optical microscopy coupled with MS. We have traced the changes during desorption process in both irradiated and non irradiated thin films. It has been noticed that the nuclei change shape depends on quantity of defects. In fact, in non irradiated samples we have observed spherical nuclei while in irradiated sample the shape is irregular. Both films exhibit modification in optical properties during desorption. To explain this feature an ab initio numerical calculation were done.

E-mail of corresponding author: zeljka.raskovic@vinca.rs
Phone +381 11 3408 507

E 14

PREPARATION AND CHARACTERIZATION OF MgH_2 -BASED COMPOSITE PELLETS

D. Mirabile Gattia¹, A. Montone¹, L. Pasquini²

¹ENEA – R.C. Casaccia –Technical Unit of Materials - Via Anguillarese, 301 00123
Rome, Italy

²Department of Physics and CNISM, University of Bologna, V. Berti-Pichat 6/2,
40127 Bologna, Italy

Focusing the attention to the application of MgH_2 -base composites inside tanks for hydrogen storage, studies were carried out with MgH_2 -based pellets prepared with the addition of Nb_2O_5 or Fe and carbon based materials. The sorption kinetic of the pellets was studied revealing the material to have a similar kinetic with respect to the powder. X-Ray diffraction analysis (XRD) and Scanning Electron Microscopy (SEM) observations of the inner core of the pellets (cross-sections), before and after the cycling process, were performed. The effects of air exposure were also investigated.

E-mail of corresponding author: daniele.mirabile@enea.it
Phone Tel.: +39 0630483484

E 15

HYDROGEN-INDUCED DEFECTS IN THIN FILMS CHARACTERIZED BY VARIABLE ENERGY SLOW POSITRON ANNIHILATION SPECTROSCOPY

O. Melikhova¹, J. Cizek¹, W. Anwand², G. Brauer²

¹Charles University in Prague, Faculty of Mathematics and Physics, V
Holesovickach 2, CZ-180 00 Praha 8, Czech Republic

²Institut für Strahlenphysik, Helmholtz-Zentrum Dresden-Rossendorf, PO Box 510
119, D-01314 Dresden, Germany

Positron annihilation spectroscopy (PAS) is a non-destructive technique which enables to characterize open volume defects and open space in lattice which is available for hydrogen. Positron is a unique probe of inter-atomic regions, grain boundaries and pores. Positron in solid matter is confined at open volume defects and annihilation radiation carries information about local electronic structure of these defects. Variable energy slow positron annihilation spectroscopy (VEPAS) uses slow positron beam with tuneable energy. Since positron penetration depth increases with increasing positron kinetic energy VEPAS enables to perform depth resolved defect studies of solid materials. In this work VEPAS was employed for characterization of hydrogen-induced defects in thin films. Hydrogen loading causes a significant volume expansion, which is isotropic in free-standing bulk materials. Contrary to bulk samples, thin films are clamped to an elastically hard substrate which prevents in-plane expansion. As a consequence, hydrogen loaded thin film may expand only in the out-of-plane direction. Hence, hydrogen-induced expansion of thin films is strongly non-uniform and hydrogen loading of thin films introduces very high internal stresses (in the GPa range). Due to these reasons, behaviour of hydrogen loaded thin films differs substantially from that of corresponding bulk samples. Using VEPAS it was found that stresses introduced by hydrogen cause plastic deformation of thin films and introduce a high concentration of open volume defects.

E-mail of corresponding author: Oksana.Melikhova@mff.cuni.cz
Phone + 420221912788

E 16

MAGNESIUM NANOPARTICLES AND NANODOTS: SYNTHESIS, STRUCTURE AND HYDROGEN SORPTION

**Mattia Sacchi¹, Alan Molinari^{1,2}, Horst Hahn², Christiaan Boelsma³,
Bernard Dam³, Luca Pasquini¹**

¹Department of Physics and Astronomy, University of Bologna, v. Berti-Pichat 6/2,
40127 Bologna, Italy

²Institute of Nanotechnology, Karlsruhe Institute of Technology, 76344
Eggenstein-Leopoldshafen, Germany

³Department of Chemical Engineering, Faculty of Applied Sciences, Delft
University of Technology, Julianalaan 136, 2628BL Delft, The Netherlands

We report on the synthesis of magnesium (Mg) nanostructures by means of two different techniques: Mg nanoparticles were prepared by inert gas condensation in a He flow, while Mg nanodots were grown by molecular beam epitaxy on a MgO substrate coated by ultra-thin alumina masks. The structure and morphology were characterized by electron microscopy, atomic force microscopy, and X-ray diffraction. First experiments aimed at the determination of the thermodynamics of hydrogen sorption by means of optical methods showed a dramatic increase of the plateau pressure for hydrogen absorption in Mg nanoparticles.

E-mail of corresponding author: luca.pasquini@unibo.it
Phone +390512095113

E 17

VANADIUM OXIDE AS HYDROGEN TECHNOLOGY MATERIAL

Dragica M. Minić¹, Dejan G. Minić², Vladimir A. Blagojević^{1,3}

¹University of Belgrade, Faculty of Physical Chemistry, Belgrade, Serbia

²Kontrola LLC, Austin, TX, USA

³Vida Holdings Corp. Ltd, Toronto, Ontario, Canada

Vanadium oxide has been well known for its catalytic activity, while its various nanostructures have been synthesized for years now. In spite of this, there have been relatively few studies of use of vanadium oxide as a catalytic material for hydrogen technologies. We present an investigation of use of very thin films of vanadium oxide, using different supports, for hydrogen technology, with a view of anticipating the behavior of synthesized vanadium oxide nanostructures and determining the factors essential to performance of these materials and ways to incorporate these into physical systems.

E-mail of corresponding author: vladab64@gmail.com

E 18

POSSIBLE PATHS OF HYDROGEN DIFFUSION IN TiO_2 – ROLE OF THE SURFACE

**Radojka Vujasin, Igor Milanović, Sandra Kurko, Bojana Paskaš
Mamula, Nikola Novaković**

Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11001
Belgrade, Serbia

We have investigated the hydrogen interaction with surface of rutile TiO_2 using pseudopotential method implemented in *Abinit* code. The hydrogen diffusion behavior and thermodynamic properties were calculated by means of full relaxation of structure in every step of bulk diffusion. The results show the existence of potential barriers close to every atomic layer and the trend of lowering of barriers and overall system energy away from surface. This goes in favor of previously experimental findings of TiO_2 low surface H coverage and easy diffusion of hydrogen into TiO_2 bulk.

E-mail of corresponding author: radojka.vujasin@vinca.rs
Phone +381 11 3408 552

E 19

DESORPTION PROPERTIES OF MgH_2 DESTABILIZED WITH NaNH_2 CATALYST

Sanja Milošević¹, Igor Milanović¹, Bojana Paskaš Mamula¹, Anđelka Đukić¹, Ljiljana Matović¹, Luca Pasquini², Jasmina Grbović Novaković¹

¹Vinča Institute of Nuclear Sciences, University of Belgrade, P.O. Box 522, 11000 Belgrade, Serbia

²Dipartimento di Fisica, Università di Bologna and CNISM, v.le Berti-Pichat 6/2, I-40127 Bologna, Italy

To destabilize MgH_2 lattice structure and improve hydrogen desorption properties mechanical milling of MgH_2 with low amount of NaNH_2 has been performed. Therefore, the amide is used as a catalyst. Pre-milling of MgH_2 has been performed followed with composite formation during 15, 30 and 60 minutes. Microstructure of samples has been characterized by XRD and SEM analysis and correlated to desorption properties examined using Hydrogen Sorption Analyser and DSC analysis. DSC analysis show that desorption temperature is shifted towards lower values and also pointed out the significance of milling duration. It has been demonstrated that desorption kinetics is significantly improved.

E-mail of corresponding author: sanjam@vinca.rs

Phone +381 11 3408 507

E 20

MICROSTRUCTURE OF ULTRAFINE-GRAINED WE43+12wt% Zn MAGNESIUM ALLOY AFTER EQUAL CHANNEL ANGULAR PRESSING

Jan Duchoň, Aleš Jäger, Viera Gärtnerová

Institute of Physics ASCR, v. v. i., Na Slovance 1999/2, 182 21 Praha 8

Mg-based alloys are considered to be one of the most promising candidates as hydrogen storage media. In this work, we investigated microstructure of complex WE43 magnesium alloy modified by the addition of 12wt% Zn. In order to refine microstructure, the alloy was subjected to equal channel angular pressing with back-pressure (ECAP-BP) at temperature of 473 K. Samples after 4 passes by Bc route were analysed using light microscopy, scanning electron microscopy (SEM) and transmission electron microscopy (TEM). TEM microstructural examination revealed MgZnY icosahedral quasicrystalline phase, Zn rich phase and various particles in Mg matrix. An average grain size of magnesium matrix was about 400 nm.

E-mail of corresponding author: duchon@fzu.cz

E 21

SEVERE PLASTIC DEFORMATION OF Mg-Mn ALLOY AT ROOM TEMPERATURE

Erik Švec, Jan Duchoň, Viera Gärtnerová, Aleš Jäger

Institute of Physics of the AS CR, v.v.i., Na Slovance 2, 182 21 Prague 8, Czech
Republic

Aim of this work is to obtain extremely refined Mg-Mn alloy with an excess of defects which can serve as traps for hydrogen. Mg-Mn binary alloy was processed by Equal Channel Angular Pressing (ECAP) with backpressure at room temperature (RT). The microstructure observed by electron back-scatter diffraction (EBSD) and transmission electron microscopy (TEM) revealed significant grain refinement with an average grain size $\sim 0.5\mu\text{m}$. Thermal stability was examined by dilatometry up to 480°C . The results show first structural instability at $\sim 80^\circ\text{C}$ during heating. For thorough characterization tensile tests at RT were also conducted after 3 and 4 passes by ECAP.

E-mail of corresponding author: svece@fzu.cz
Phone +420266052458

E 22

ENHANCING THE HYDROGEN GRAPHENE INTERACTION BY METAL DECORATION

Andreas Bliersbach

Empa, Swiss Federal Laboratories for Materials Science and Technology, Hydrogen
and Energy, Überlandstrasse 129, CH-8600 Dübendorf, Switzerland.

Theoretical calculations of metal decorated graphene predict an attractive orbital interaction, the Kubas interaction, of molecular hydrogen to the metal centers. In this case a noticeably higher heat of adsorption for the bound hydrogen and hence a higher desorption temperature as compared to physisorbed hydrogen is expected. Molecular hydrogen bound to metal decorated graphene is an ideal model system to study the Kubas interaction in detail and promote the understanding of the underlying fundamental principles. Furthermore it is a promising candidate for a cheap, lightweight, intermediate temperature hydrogen storage material.

E-mail of corresponding author: andreas.bliersbach@empa.ch
Phone +41 58 765 48 62

E 23

THE POTENTIAL APPLICATION OF HYDROGEN IN DEVELOPING COUNTRIES

Rabah Arkam

Euro T-Systems, 05, rue Chafai Ahmed, Tizi Ouzou Algeria;

Given the promising hydrogen internal combustion engines, an energy system can then be offered as an alternative or complementary energy system. This potential is driven by three drivers: 1. CO₂: The problem is to make all efforts to reduce emissions of greenhouse gases, particularly CO₂. 2. Energy independence: Using hydrogen and improve their geopolitical situation. 3. The inevitable exhaustion of fossil resources, we want to keep as a raw material (plastic ...) rather than burn them. Chemical, hydrogen energy, It is produced by reforming natural gas by electrolysis or thermo chemical relying on nuclear energy or renewable.

E-mail of corresponding author: r.arkam@yahoo.fr

Phone +213 554 007 992

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